SEARCH REQUEST FORM

Scientific and Technical Information Center

Requester's Full Name: Additional Phone Mail Box and Bldg/Room Location	Kawa TATYA Number 308 - 128 n: <u>CP3 2E16</u> R	A//Examiner # : 76 19 Serial Numbersults Format Preferre	608 Date: 4/10/03 er: 09/4/7, 674 d (circle): PAPER DISK E-M.	AIĹ
If mor than one search is subr	nitted, please priori		The same and	
****************************** Please provide a detailed statement of the Include the elected species or structures, utility of the invention. Define any terms known. Please attach a copy of the cover	*************** be search topic, and descrift keywords, synonyms, act s that may have a special	********* be as specifically as possible on specifically	ole the subject matter to be searched	****
Title of Invention:	ee astoche	d	•	
Inventors (please provide full names):	<i>II</i>	<i>y</i>		
Earliest Priority Filing Date:	28/01/200	0		
*For Sequence Searches Only * Please inclu appropriate serial number.	de all pertinent informatio	n (parent, child, divisional, c		;
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STAFF USE ONLY	Type of Search	**************************************	t cost where applicable	
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rate Searcher Picked Up:	Bibliographic	Dr.Link_		
rate Completed:	Litigation	Lexis/Nexis_		
earcher Prep & Review Time: 20	Fulltext	Sequence Systems		
lerical Prep Time:	Patent Family	www/Internet	·	
nline Time:	Other	Other (specify)		
10-1590 (1-2000) N Rules				

=> FILE REG

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STRUCTURE FILE UPDATES: 14 APR 2003 HIGHEST RN 502958-40-9 DICTIONARY FILE UPDATES: 14 APR 2003 HIGHEST RN 502958-40-9

TSCA INFORMATION NOW CURRENT THROUGH MAY 20, 2002

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Crossover limits have been increased. See HELP CROSSOVER for details.

Experimental and calculated property data are now available. See HELP PROPERTIES for more information. See STNote 27, Searching Properties in the CAS Registry File, for complete details: http://www.cas.org/ONLINE/STN/STNOTES/stnotes27.pdf

=> FILE HCAPLUS

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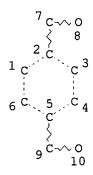
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FILE COVERS 1907 - 15 Apr 2003 VOL 138 ISS 16 FILE LAST UPDATED: 14 Apr 2003 (20030414/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> D QUE L48

L3 STR



Structure 1

NODE ATTRIBUTES:
CONNECT IS E1 RC AT 8
CONNECT IS E1 RC AT 10
DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 10

STEREO ATTRIBUTES: NONE
L4 STR

CH2: $C \sim C = 0$ 1 2 3 4

NODE ATTRIBUTES: DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:
RING(S) ARE ISOLATED OR EMBEDDED
NUMBER OF NODES IS 4

STEREO ATTRIBUTES: NONE

L9 4197 SEA FILE=REGISTRY SSS FUL L3 AND L4 L10 STR

 Subset search with this structure
of 4,197
912 compounds

4, 197 Compounds from 1 and 2

NODE ATTRIBUTES:
DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:
RING(S) ARE ISOLATED OR EMBEDDED
NUMBER OF NODES IS 7

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STEREO ATTRIBUTES: NONE
           912 SEA FILE=REGISTRY SUB=L9 SSS FUL L10
L14
            416 SEA FILE=HCAPLUS ABB=ON L14
250 SEA FILE=HCAPLUS ABB=ON L15(L)(PREP OR IMF OR SPN)/RL
12 SEA FILE=HCAPLUS ABB=ON L16(L)BINDER?
L15
L16
L17
                          Subset search with this structure
704 compounds
L20
0-Ak-0-Ak-0
NODE ATTRIBUTES:
DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED
GRAPH ATTRIBUTES:
RING(S) ARE ISOLATED OR EMBEDDED
NUMBER OF NODES IS
STEREO ATTRIBUTES: NONE
            704 SEA FILE=REGISTRY SUB=L9 SSS FUL L20
L22
         167072 SEA FILE=REGISTRY ABB=ON POLYESTER/PCT
L23
L24
            498 SEA FILE=REGISTRY ABB=ON L22 AND L23
L25
         275052 SEA FILE=REGISTRY ABB=ON POLYACRYLIC/PCT
            477 SEA FILE=REGISTRY ABB=ON L24 AND L25
L26
L27
         227111 SEA FILE=REGISTRY ABB=ON POLYETHER/PCT
L28
            369 SEA FILE=REGISTRY ABB=ON L26 AND L27
L29
            230 SEA FILE=HCAPLUS ABB=ON L28
            129 SEA FILE=HCAPLUS ABB=ON L29(L)(PREP OR IMF OR SPN OR POF)/RL
L31
L32
             12 SEA FILE=HCAPLUS ABB=ON L31(L)BINDER?
               3 SEA FILE=REGISTRY ABB=ON PET/CN
L33
L34
          60365 SEA FILE=HCAPLUS ABB=ON L33
          11340 SEA FILE=HCAPLUS ABB=ON L34(L) (PREP OR IMF OR SPN OR POF)/RL
L35
            156 SEA FILE=HCAPLUS ABB=ON L35(L)?ACRYL?
L36
              2 SEA FILE=HCAPLUS ABB=ON L36(L)BINDER?
L37
             23 SEA FILE=HCAPLUS ABB=ON L17 OR L32 OR L37
L38
              2 SEA FILE=HCAPLUS ABB=ON L36(L)OLIGOM?
L39
                                          (L16 OR L31)(L)OLIG?
L40
              3 SEA FILE=HCAPLUS ABB=ON
              4 SEA FILE=HCAPLUS ABB=ON L36(L)OLIG?
L41
             28 SEA FILE=HCAPLUS ABB=ON
                                          (L38 OR L39 OR L40 OR L41)
L42
             5 SEA FILE=HCAPLUS ABB=ON L16(L) POF/RL(L) (BINDER? OR OLIG?)
28 SEA FILE=HCAPLUS ABB=ON L42 OR L43
L43
L44
                 ANALYZE L*** 1- RN LNK$:
L***
                                                   38 TERMS
             39 SEA FILE=REGISTRY ABB=ON L*** OR L33
L45
          11415 SEA FILE=HCAPLUS ABB=ON L45(L) (PREP OR IMF OR SPN OR POF)/RL
L46
              4 SEA FILE=HCAPLUS ABB=ON L46(L)?ACRYL?(L) (BINDER? OR OLIG?)
L47
             28 SEA FILE=HCAPLUS ABB=ON L44 OR L47
L48
=> D L48 ALL 1-28 HITSTR
   ANSWER 1 OF 28 HCAPLUS COPYRIGHT 2003 ACS
ΑN
     2003:216775 HCAPLUS
DN
     138:230844
     Coated polyester film with high blocking resistance for
     high-recording-density magnetic recording medium base film
IN
     Tojo, Mitsumine; Murooka, Hirofumi
PA
     Teijin-Du Pont Film Inc., Japan
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Jpn. Kokai Tokkyo Koho, 11 pp.

SO

CODEN: JKXXAF DΤ Patent LΑ Japanese IC ICM B32B027-36 ICS G11B005-73 77-8 (Magnetic Phenomena) CC Section cross-reference(s): 38, 46 FAN.CNT 1 APPLICATION NO. DATE PATENT NO. KIND DATE ______ _____ JP 2003080652 A2 20030319 JP 2001-276319 20010912 PΤ PRAI JP 2001-276319 20010912 The polyester [e.g., poly(ethylene terephthalate), polyethylene-2,6naphthalene dicarboxylate] film for a digital recording-type magnetic tape, etc., has a coating layer contg. binder resins, inert particles, and surfactants comprising (A) 0.5-15% (based on coating layer wt.) primary surfactants having HLB value 10-14 and (B) 1-10% secondary surfactants having HLB value 16-20. The other side of the film may be coated with a layer contg. cellulose derivs. and surfactants. A magentic recording medium consisting of the coated film, a magnetic layer (e.g., ferromagnetic metal film) on the former coating layer, and a back coating layer on the other side of the film, is also claimed. magnetic tape polyester film coating surfactant ST TΤ Alcohols, uses RL: TEM (Technical or engineered material use); USES (Uses) (C11-15-secondary, ethoxylated, surfactant, Nonipol Soft SS 70; blocking-resistant polyester film coated with surfactant-contg. layer for high-recording-d. magnetic recording medium) IT Polyesters, uses RL: IMF (Industrial manufacture); TEM (Technical or engineered material . use); PREP (Preparation); USES (Uses) (acrylic, binder, coating layer contg.; blocking-resistant polyester film coated with surfactant-contg. layer for high-recording-d. magnetic recording medium) Magnetic tapes IT Plastic films Surfactants (blocking-resistant polyester film coated with surfactant-contg. layer for high-recording-d. magnetic recording medium) IT Polyesters, uses RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (blocking-resistant polyester film coated with surfactant-contg. layer for high-recording-d. magnetic recording medium) IT Acrylic polymers, uses RL: TEM (Technical or engineered material use); USES (Uses) (particles, coating layer contq.; blocking-resistant polyester film coated with surfactant-contg. layer for high-recording-d. magnetic recording medium) 345300-53-0P, Butyl acrylate-diethylene glycol-ethylene IT glycol-glycidyl methacrylate-isophthalic acid-methyl methacrylate-5sodiosulfoisophthalic acid-terephthalic acid copolymer 501037-16-7P, Diethylene glycol-ethylene glycol-glycidyl methacrylate-isophthalic acid-methyl methacrylate-5-sodiosulfoisophthalic acid copolymer RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
 (binder, coating layer contg.; blocking-resistant polyester

film coated with surfactant-contq. layer for high-recording-d. magnetic

recording medium) 26915-97-9, Ethyl acrylate-2-hydroxyethyl methacrylate-methacrylic IT acid-methyl methacrylate copolymer 87139-72-8, Diethylene glycol-ethylene glycol-isophthalic acid-5-sodiosulfoisophthalic acid-terephthalic acid copolymer RL: TEM (Technical or engineered material use); USES (Uses) (binder, coating layer contg.; blocking-resistant polyester film coated with surfactant-contg. layer for high-recording-d. magnetic recording medium) 25038-59-9P, Dimethyl terephthalate-ethylene glycol IT 24968-11-4P copolymer, uses 25853-85-4P, Dimethyl 2,6-naphthalenedicarboxylateethylene glycol copolymer RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (blocking-resistant polyester film coated with surfactant-contg. layer for high-recording-d. magnetic recording medium) IT 9004-67-5, Methylcellulose RL: TEM (Technical or engineered material use); USES (Uses) (coating layer contg.; blocking-resistant polyester film coated with surfactant-contq. layer for high-recording-d. magnetic recording medium) IT 11104-61-3, Cobalt oxide RL: TEM (Technical or engineered material use); USES (Uses) (magnetic layer; blocking-resistant polyester film coated with surfactant-contq. layer for high-recording-d. magnetic recording medium) IT 9016-45-9, Nonipol 100 9036-19-5, Octapol 80 9063-89-2, Octapol 400 220204-87-5, Nonipol 700 RL: TEM (Technical or engineered material use); USES (Uses) (surfactant; blocking-resistant polyester film coated with surfactant-contg. layer for high-recording-d. magnetic recording medium) 345300-53-0P, Butyl acrylate-diethylene glycol-ethylene TΤ glycol-glycidyl methacrylate-isophthalic acid-methyl methacrylate-5sodiosulfoisophthalic acid-terephthalic acid copolymer RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (binder, coating layer contg.; blocking-resistant polyester film coated with surfactant-contg. layer for high-recording-d. magnetic recording medium) 345300-53-0 HCAPLUS RN 1,3-Benzenedicarboxylic acid, 5-sulfo-, monosodium salt, polymer with CN 1,3-benzenedicarboxylic acid, 1,4-benzenedicarboxylic acid, butyl 2-propenoate, 1,2-ethanediol, methyl 2-methyl-2-propenoate, oxiranylmethyl 2-methyl-2-propenoate and 2,2'-oxybis[ethanol] (9CI) (CA INDEX NAME) CM

CRN 6362-79-4 CMF C8 H6 O7 S . Na

Na

CM 2

CRN 141-32-2 CMF C7 H12 O2

CM 3

CRN 121-91-5 CMF C8 H6 O4

CM 4

CRN 111-46-6 CMF C4 H10 O3

$${\tt HO-CH_2-CH_2-O-CH_2-CH_2-OH}$$

CM 5

CRN 107-21-1 CMF C2 H6 O2

$$HO-CH_2-CH_2-OH$$

CRN 106-91-2 CMF C7 H10 O3

CM 7

CRN 100-21-0 CMF C8 H6 O4

CM 8

CRN 80-62-6 CMF C5 H8 O2

L48 ANSWER 2 OF 28 HCAPLUS COPYRIGHT 2003 ACS

AN 2002:489019 HCAPLUS

DN 137:202212

On the UV curability and mechanical properties of novel binder systems derived from poly(ethylene terephthalate) (PET) waste for solventless magnetic tape manufacturing, 2 methacrylated oligoesters

AU Farahat, Medhat S.; Nikles, David E.

CS Center for Materials for Information Technology, University of Alabama, Tuscaloosa, AL, 35487-0209, USA

SO Macromolecular Materials and Engineering (2002), 287(5), 353-362 CODEN: MMENFA; ISSN: 1438-7492

PB Wiley-VCH Verlag GmbH

DT Journal

LA English

CC 38-3 (Plastics Fabrication and Uses)
Section cross-reference(s): 35

AB Recycling of poly(ethylene terephthalate) PET waste by chem. methods is a well-known process that generates value-added products. Depolymn. products of PET recycling were commonly applied as starting materials for the synthesis of polyurethanes, satd. and unsatd. polyester resins. In

this current work we are reporting on a novel application of the depolymd. products obtained by glycolysis of PET by converting the hydroxyl functional groups to methacrylate groups. The obtained methacrylated oligoesters were tested for UV curability by UV irradn., in the presence of 2-benzyl-2-dimethylamino-1-(4-morpholinophenyl)-1-butanone (BDMB) as a photo initiator. This gave cured films of high mech. properties when these methacrylated oligoesters were either cured alone or as mixts. With other com. available diacrylate/dimethacrylate monomers. The measured tensile properties were in the range of 7.21-43 MPa for max. tensile strength and 0.90-3.0 GPa for Young's modulus.

ST PET glycolysis diol methacrylate binder magnetic tape; UV cured PET methacrylate diacrylate compn

IT Tensile strength

(at break; property of UV curable binder compn. based on methacrylate monomer of glycolyzed poly(ethylene terephthalate) (PET) waste for magnetic tape manufg.)

IT Transesterification

(in prepn. of UV curable binder compn. based on methacrylate monomer of glycolyzed poly(ethylene terephthalate) (PET) waste for magnetic tape manufg.)

IT Polyesters, properties

RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(in prepn. of UV curable binder compn. based on methacrylate monomer of glycolyzed poly(ethylene terephthalate) (PET) waste for magnetic tape manufg.)

IT Polymerization

(photopolymn.; in prepn. of UV curable binder compn. based on methacrylate monomer of glycolyzed poly(ethylene terephthalate) (PET) waste for magnetic tape manufg.)

IT Binders

Glycolysis

Magnetic tapes

Recycling of plastics and rubbers

(prepn. of UV curable binder compn. based on methacrylate monomer of glycolyzed poly(ethylene terephthalate) (PET) waste for magnetic tape manufg.)

IT Elongation, mechanical

Young's modulus

(property of UV curable binder compn. based on methacrylate monomer of glycolyzed poly(ethylene terephthalate) (PET) waste for magnetic tape manufg.)

IT Polyesters, preparation

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(unsatd.; in prepn. of UV curable binder compn. based on methacrylate monomer of glycolyzed poly(ethylene terephthalate) (PET) waste for magnetic tape manufg.)

IT 920-46-7, Methacryloyl chloride

RL: RCT (Reactant); RACT (Reactant or reagent)
(in prepn. of UV curable binder compn. based on methacrylate monomer of
glycolyzed poly(ethylene terephthalate) (PET) waste for magnetic tape

nanufg.)

IT 25052-77-1P, Diethylene glycol-ethylene glycol-terephthalic acid copolymer RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(polyester polyol; in prepn. of UV curable binder compn. based on methacrylate monomer of glycolyzed poly(ethylene terephthalate) (PET)

```
waste for magnetic tape manufg.)
     452963-93-8P, Diethylene glycol-ethylene glycol-terephthalic acid
IT
     copolymer methacrylate homopolymer 452963-94-9P, Diethylene
     glycol-ethylene glycol-terephthalic acid copolymer methacrylate-diethylene
     glycol diacrylate copolymer 452963-95-0P, Diethylene
     glycol-ethylene glycol-terephthalic acid copolymer methacrylate-diethylene
     glycol dimethacrylate copolymer 452963-96-1P, Diethylene
     glycol-ethylene glycol-terephthalic acid copolymer methacrylate-
     triethylene glycol dimethacrylate copolymer 452963-97-2P,
     Diethylene glycol-ethylene glycol-terephthalic acid copolymer
     methacrylate-hexanediol diacrylate copolymer 452963-98-3P,
     Diethylene glycol-ethylene glycol-terephthalic acid copolymer
     methacrylate-hexanediol dimethacrylate copolymer
     RL: PRP (Properties); SPN (Synthetic preparation); PREP
     (Preparation)
        (prepn. of UV curable binder compn. based on methacrylate
        monomer of glycolyzed poly(ethylene terephthalate) (PET) waste for
        magnetic tape manufg.)
     452963-92-7P, Diethylene glycol-ethylene glycol-terephthalic acid
IT
     copolymer methacrylate
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (unsatd. polyester; in prepn. of UV curable binder compn. based on
        methacrylate monomer of glycolyzed poly(ethylene terephthalate) (PET)
        waste for magnetic tape manufg.)
              THERE ARE 50 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE.CNT
RE
(1) 'A quote from Plasticycle Corp; 257 Maxmaroneck Ave
(2) A quote from Waste Recyclers of Tuscaloosa Inc; 907 31st Avenue
(3) Abdel-Azim, A; Polym Adv Technol 1995, V6, P688 HCAPLUS
(4) Ahn, K; J Appl Polym Sci 1999, V71, P2033 HCAPLUS
(5) Anon; ASTM Designation D 882-91
(6) Anon; 502 Thomas Jones Way
(7) Anon; Chem Market Rep 2001, P257
(8) Anon; Chem Week 1989, P65
(9) Chen, J; J Appl Polym Sci 1999, V73, P35 HCAPLUS
(10) Cheng, S; CHEMTECH 1995, V25, P35 HCAPLUS
(11) Cheng, S; Waste Manage 1995, V15, P257 HCAPLUS
(12) Davies, C; Polym Int 1996, V41, P215 HCAPLUS
(13) Farahat, M; Macromol Mater Eng 2000, V283, P1 HCAPLUS (14) Farahat, M; Macromol Mater Eng 2001, V286, P695 HCAPLUS
(15) Farahat, M; Polym Int 2002, V51, P183 HCAPLUS
(16) Fritz, D; Anal Chem 1979, V51, P7 HCAPLUS
(17) Fritz, J; Anal Chem 1959, V31, P1808 HCAPLUS
(18) Gao, F; Polym Prepr (Am Chem Soc, Div Polym Chem) 2000, V4, P125
(19) Gueclue, G; J Appl Polym Sci 1998, V69, P2311
(20) Halacheva, N; Polymer 1995, V36, P867 HCAPLUS
(21) Harper, C; Handbook of Plastics, Elastomers and Composites 3rd ed 1996
(22) Hoyle, C; Am Chem Soc 1990, P92
(23) Koster, E; Technology 1987, V1, P98
(24) Krzan, A; J Appl Polym Sci 1998, V69, P1115 HCAPLUS
(25) Krzan, A; Polym Adv Technol 1999, V10, P603 HCAPLUS
(26) Lee, S; J Appl Polym Sci 1994, V52, P869 HCAPLUS
(27) Lusinchi, J; J Appl Polym Sci 1998, V69, P657
(28) Malucelli, G; J Appl Polym Sci 1996, V65, P491
(29) Malucelli, G; Polymer 1996, V37, P2565 HCAPLUS
(30) Mancini, S; J Appl Polym Sci 2000, V76, P266 HCAPLUS
(31) Nikles, D; ACS Symp Ser 2000, V766, P18
```

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(32) Odian, G; Principles of Polymerization 2nd ed 1981, P40
(33) Oku, A; J Appl Polym Sci 1997, V63, P595 HCAPLUS
(34) Oku, V; J Appl Polym Sci 1997, V63, P595
(35) Priola, A; Polymer 1993, V34, P3653 HCAPLUS
(36) Rebeiz, K; ACI Mater J 1994, V91, P313 HCAPLUS
(37) Rebeiz, K; Cem Concr Compos 1995, V17, P119 HCAPLUS
(38) Rebeiz, K; J Appl Polym Sci 1992, V44, P1649 HCAPLUS
(39) Rebeiz, K; J Mater Civil Eng 1994, V6, P150 HCAPLUS
(40) Rebeiz, K; J Mater Civil Eng 1995, V7, P129 HCAPLUS
(41) Rebeiz, K; J Structural Eng 1995, V121, P1370
(42) Seewald, N; Chem Week 2000, P43
(43) Siggia, S; Quantitative organic analysis via functional group analysis 3rd
    ed 1963, P8
(44) Sorenson, W; Preparative Methods of Polymer Chemistry 3rd ed 2001, P358
(45) Stetzler, R; Anal Chem 1962, V34, P194 HCAPLUS
(46) Suh, D; Polymer 2000, V41, P461 HCAPLUS
(47) Tong, S; Polymer 1983, V24, P469 HCAPLUS
(48) Vaidya, U; Ind Eng Chem Res 1987, V26, P194 HCAPLUS (49) Vaidya, U; J Appl Polym Sci 1987, V34, P235 HCAPLUS
(50) Watkins, K; Chem Eng News 2000, V78, P11
     452963-93-8P, Diethylene glycol-ethylene glycol-terephthalic acid
IT
     copolymer methacrylate homopolymer 452963-94-9P, Diethylene
     glycol-ethylene glycol-terephthalic acid copolymer methacrylate-diethylene
     glycol diacrylate copolymer 452963-95-0P, Diethylene
     glycol-ethylene glycol-terephthalic acid copolymer methacrylate-diethylene
     glycol dimethacrylate copolymer 452963-96-1P, Diethylene
     glycol-ethylene glycol-terephthalic acid copolymer methacrylate-
     triethylene glycol dimethacrylate copolymer 452963-97-2P,
     Diethylene glycol-ethylene glycol-terephthalic acid copolymer
     methacrylate-hexanediol diacrylate copolymer 452963-98-3P,
     Diethylene glycol-ethylene glycol-terephthalic acid copolymer
     methacrylate-hexanediol dimethacrylate copolymer
     RL: PRP (Properties); SPN (Synthetic preparation); PREP
     (Preparation)
         (prepn. of UV curable binder compn. based on methacrylate
        monomer of glycolyzed poly(ethylene terephthalate) (PET) waste for
        magnetic tape manufg.)
RN
     452963-93-8 HCAPLUS
CN
     1,4-Benzenedicarboxylic acid, polymer with 1,2-ethanediol and
     2,2'-oxybis[ethanol], 2-methyl-2-propenoate, homopolymer (9CI) (CA INDEX
     NAME)
     CM
          1
          452963-92-7
     CRN
          (C8 H6 O4 . C4 H1O O3 . C2 H6 O2)x . x C4 H6 O2
     CMF
          CM
                2
          CRN
               79-41-4
          CMF
                C4 H6 O2
    CH<sub>2</sub>
```

Me-C-CO2H

CRN 25052-77-1

CMF (C8 H6 O4 . C4 H10 O3 . C2 H6 O2)x

CCI PMS

CM 4

CRN 111-46-6 CMF C4 H10 O3

 ${\tt HO-CH_2-CH_2-O-CH_2-CH_2-OH}$

CM 5

CRN 107-21-1 CMF C2 H6 O2

 ${\tt HO-CH_2-CH_2-OH}$

CM 6

CRN 100-21-0 CMF C8 H6 O4

RN 452963-94-9 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, polymer with 1,2-ethanediol and 2,2'-oxybis[ethanol], 2-methyl-2-propenoate, polymer with oxydi-2,1-ethanediyl di-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 4074-88-8 CMF C10 H14 O5

CM 2

CRN 452963-92-7

CMF (C8 H6 O4 . C4 H10 O3 . C2 H6 O2)x . x C4 H6 O2

CM 3

CRN 79-41-4 CMF C4 H6 O2

 $\begin{array}{c} \text{CH}_2 \\ || \\ \text{Me-C-CO}_2\text{H} \end{array}$

CM 4

CRN 25052-77-1

CMF (C8 H6 O4 . C4 H10 O3 . C2 H6 O2)x

CCI PMS

CM 5

CRN 111-46-6 CMF C4 H10 O3

но- сн2- сн2- о- сн2- сн2- он

CM 6

CRN 107-21-1 CMF C2 H6 O2

 $HO-CH_2-CH_2-OH$

CM 7

CRN 100-21-0 CMF C8 H6 O4

HO2C CO2H

RN 452963-95-0 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, polymer with 1,2-ethanediol and 2,2'-oxybis[ethanol], 2-methyl-2-propenoate, polymer with oxydi-2,1-ethanediyl bis(2-methyl-2-propenoate) (9CI) (CA INDEX NAME)

CM 1

CRN 2358-84-1 CMF C12 H18 O5

CM 2

CRN 452963-92-7

CMF (C8 H6 O4 . C4 H1O O3 . C2 H6 O2)x . x C4 H6 O2

CM 3

CRN 79-41-4 CMF C4 H6 O2

$$\begin{array}{c} \text{CH}_2 \\ || \\ \text{Me-C-CO}_2 \text{H} \end{array}$$

CM 4

CRN 25052-77-1

CMF (C8 H6 O4 . C4 H10 O3 . C2 H6 O2) \times

CCI PMS

CM 5

CRN 111-46-6 CMF C4 H10 O3

 ${\tt HO-CH_2-CH_2-O-CH_2-CH_2-OH}$

CM 6

CRN 107-21-1 CMF C2 H6 O2

 $HO-CH_2-CH_2-OH$

CM 7

CRN 100-21-0

CMF C8 H6 O4

RN 452963-96-1 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, polymer with 1,2-ethanediol and 2,2'-oxybis[ethanol], 2-methyl-2-propenoate, polymer with 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate) (9CI) (CA INDEX NAME)

CM 1

CRN 109-16-0 CMF C14 H22 O6

CM 2

CRN 452963-92-7 CMF (C8 H6 O4 . C4 H1O O3 . C2 H6 O2)x . x C4 H6 O2

CM 3

CRN 79-41-4 CMF C4 H6 O2

CM 4

CRN 25052-77-1

CMF (C8 H6 O4 . C4 H10 O3 . C2 H6 O2)x

CCI PMS

CM 5

CRN 111-46-6 CMF C4 H10 O3

HO-CH2-CH2-O-CH2-CH2-OH

CRN 107-21-1 CMF C2 H6 O2

 $HO-CH_2-CH_2-OH$

CM 7

CRN 100-21-0 CMF C8 H6 O4

RN 452963-97-2 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, polymer with 1,2-ethanediol and 2,2'-oxybis[ethanol], 2-methyl-2-propenoate, polymer with 1,6-hexanediyl di-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 13048-33-4 CMF C12 H18 O4

CM 2

CRN 452963-92-7

CMF (C8 H6 O4 . C4 H10 O3 . C2 H6 O2)x . x C4 H6 O2

CM 3

CRN 79-41-4 CMF C4 H6 O2

$$\begin{array}{c} \text{CH}_2 \\ || \\ \text{Me-C-CO}_2 \text{H} \end{array}$$

CM 4

CRN 25052-77-1

CMF (C8 H6 O4 . C4 H10 O3 . C2 H6 O2)x

CCI PMS

CM 5

CRN 111-46-6 CMF C4 H10 O3

 ${\tt HO-CH_2-CH_2-O-CH_2-CH_2-OH}$

CM 6

CRN 107-21-1 CMF C2 H6 O2

 $HO-CH_2-CH_2-OH$

CM 7

CRN 100-21-0 CMF C8 H6 O4

RN 452963-98-3 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, polymer with 1,2-ethanediol and 2,2'-oxybis[ethanol], 2-methyl-2-propenoate, polymer with 1,6-hexanediyl bis(2-methyl-2-propenoate) (9CI) (CA INDEX NAME)

CM 1

CRN 6606-59-3 CMF C14 H22 O4

CM 2

CRN 452963-92-7

CMF (C8 H6 O4 . C4 H10 O3 . C2 H6 O2)x . x C4 H6 O2

CM 3

CRN 79-41-4 CMF C4 H6 O2

CH₂ || Me-C-CO₂H

CM 4

CRN 25052-77-1

CMF (C8 H6 O4 . C4 H10 O3 . C2 H6 O2) \times

CCI PMS

CM 5

CRN 111-46-6 CMF C4 H10 O3

но- сн2- сн2- о- сн2- сн2- он

CM 6

CRN 107-21-1 CMF C2 H6 O2

 ${\rm HO-CH_2-CH_2-OH}$

CM 7

CRN 100-21-0 CMF C8 H6 O4

L48 ANSWER 3 OF 28 HCAPLUS COPYRIGHT 2003 ACS

AN 2002:349314 HCAPLUS

DN 136:378886

TI Heat-resistant polyester films for thin magnetic recording media

IN Tojo, Mitsumine

- PA Teijin Ltd., Japan
- SO Jpn. Kokai Tokkyo Koho, 10 pp. CODEN: JKXXAF
- DT Patent
- LA Japanese
- IC ICM G11B005-73 ICS C08J007-04; C08L067-02
- CC 77-8 (Magnetic Phenomena)
- Section cross-reference(s): 38

FAN.CNT 1

PΙ

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2002133643	A2	20020510	JP 2000-332179	20001031

PRAI JP 2000-332179 20001031

- AB The polyester film, useful for digital video tapes, has at least on one side a layer with a surface microstructure comprising concave and convex lines with width 0.1-3 .mu.m, wherein the layer with surface roughness Ra (measured by AFM) 3-20 nm contains inert microparticles with av. diam. 10-200 nm. The layer may be deposited on a back side of a magnetic tape. Thermal distortion in vapor deposition of magnetic layers is prevented with the films.
- ST polyester film magnetic recording uneven backside; heat resistance polyester digital video tape
- IT Polyesters, uses
 - RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)
 - (acrylic, binder; heat-resistant polyester films for thin digital magnetic tapes having back layers with uneven network structures)
- IT Polyoxyalkylenes, uses
 - RL: MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses)
 - (alkyl group-terminated, surfactant; heat-resistant polyester films for thin digital magnetic tapes having back layers with uneven network structures)
- IT Polyesters, uses
 - RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
 - (base film; heat-resistant polyester films for thin digital magnetic tapes having back layers with uneven network structures)
- IT Magnetic tapes
 - Plastic films
 - (heat-resistant polyester films for thin digital magnetic tapes having back layers with uneven network structures)
- IT Acrylic polymers, uses
 - RL: MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses)
 - (microparticles, surface layer contg.; heat-resistant polyester films for thin digital magnetic tapes having back layers with uneven network structures)
- IT 24968-11-4P, Dimethyl 2,6-naphthalate-ethylene glycol copolymer, SRU 25038-59-9P, Dimethyl terephthalate-ethylene glycol copolymer, uses 25853-85-4P, Dimethyl 2,6-naphthalenedicarboxylate-ethylene glycol copolymer
 - RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
 - (base film; heat-resistant polyester films for thin digital magnetic tapes having back layers with uneven network structures)
- IT 9002-89-5, Polyvinyl alcohol 9004-65-3, Hydroxypropylmethyl cellulose

87139-72-8, Diethylene glycol-ethylene glycol-isophthalic acid-5-sodiosulfoisophthalic acid-terephthalic acid copolymer 422324-41-2, Diethylene glycol-ethylene glycol-glycidyl methacrylate-isophthalic acid-methyl methacrylate-5-sodiosulfoisophthalic acid-terephthalic acid copolymer RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses) (binder; heat-resistant polyester films for thin digital magnetic tapes having back layers with uneven network structures) IT 7631-86-9, Silica, uses RL: MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses) (surface layer contg.; heat-resistant polyester films for thin digital magnetic tapes having back layers with uneven network structures) 9016-45-9, Polyethylene glycol nonylphenyl ether IT RL: MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses) (surfactant; heat-resistant polyester films for thin digital magnetic tapes having back layers with uneven network structures) 422324-41-2, Diethylene glycol-ethylene glycol-glycidyl IT methacrylate-isophthalic acid-methyl methacrylate-5-sodiosulfoisophthalic acid-terephthalic acid copolymer RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses) (binder; heat-resistant polyester films for thin digital magnetic tapes having back layers with uneven network structures) RN 422324-41-2 HCAPLUS 1,3-Benzenedicarboxylic acid, 5-sulfo-, monosodium salt, polymer with CN 1,3-benzenedicarboxylic acid, 1,4-benzenedicarboxylic acid, 1,2-ethanediol, methyl 2-methyl-2-propenoate, oxiranylmethyl 2-methyl-2-propenoate and 2,2'-oxybis[ethanol] (9CI) (CA INDEX NAME) CM 1 CRN 6362-79-4 CMF C8 H6 O7 S . Na

Na

CM 2

CRN 121-91-5 CMF C8 H6 O4

CRN 111-46-6 CMF C4 H10 O3

 ${\tt HO-CH_2-CH_2-O-CH_2-CH_2-OH}$

CM 4

CRN 107-21-1 CMF C2 H6 O2

 $HO-CH_2-CH_2-OH$

CM 5

CRN 106-91-2 CMF C7 H10 O3

$$\begin{tabular}{c|c} O & O & CH_2 \\ & \parallel & \parallel \\ CH_2-O-C-C-Me \end{tabular}$$

CM 6

CRN 100-21-0 CMF C8 H6 O4

CM '

CRN 80-62-6 CMF C5 H8 O2

```
H<sub>2</sub>C O
    Me-C-C-OMe
```

L48 ANSWER 4 OF 28 HCAPLUS COPYRIGHT 2003 ACS

2002:349313 HCAPLUS AN

136:378885 DN

Blocking- and scratch-resistant flat polyester films for magnetic ΤI recording media

IN Tojo, Mitsumine

PA Teijin Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 9 pp. CODEN: JKXXAF

DT Patent

Japanese LA

ICM G11B005-73 IC ICS B32B027-36

77-8 (Magnetic Phenomena) CC Section cross-reference(s): 38

FAN.CNT 1

KIND DATE PATENT NO. APPLICATION NO. DATE _____ ____ ----------_____ A2 JP 2002133642 20020510 JP 2000-330285 20001030 PΙ

PRAI JP 2000-330285 20001030

The polyester film with surface roughness (measured by AFM) Ra 0.1-3 nm (<3) and Rz 5-40 nm has at least on one side a layer comprising (A) binder resins, (B) inert microparticles, (C) surfactants, and (D) 1-40% styrene polymers contg. sulfonate groups. The films are particularly useful for digital video tapes.

polyester film magnetic recording medium antiblocking; scratch resistance STpolyester digital video tape; styrene polymer flat polyester magnetic tape

ΙT Polyesters, uses

> RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)

(acrylic, binder; blocking- and scratch-resistant flat polyester films for digital magnetic tapes)

IT Polyoxyalkylenes, uses

> RL: MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses)

(alkyl group-terminated, surfactant; blocking- and scratch-resistant flat polyester films for digital magnetic tapes)

ITPolyesters, uses

> RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(base film; blocking- and scratch-resistant flat polyester films for digital magnetic tapes)

IT Magnetic tapes

Plastic films

(blocking- and scratch-resistant flat polyester films for digital magnetic tapes)

IT Acrylic polymers, uses

> RL: MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses)

(microparticles, surface layer contg.; blocking- and scratch-resistant flat polyester films for digital magnetic tapes)

```
24968-11-4P 25038-59-9P, Dimethyl terephthalate-ethylene glycol
                       25853-85-4P, Dimethyl 2,6-naphthalenedicarboxylate-
     copolymer, uses
     ethylene glycol copolymer
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (base film; blocking- and scratch-resistant flat polyester films for
        digital magnetic tapes)
     87139-72-8, Diethylene glycol-ethylene glycol-isophthalic
IT
     acid-5-sodiosulfoisophthalic acid-terephthalic acid copolymer
     422324-41-2, Diethylene glycol-ethylene glycol-glycidyl
     methacrylate-isophthalic acid-methyl methacrylate-5-sodiosulfoisophthalic
     acid-terephthalic acid copolymer
     RL: POF (Polymer in formulation); TEM (Technical or engineered
     material use); USES (Uses)
        (binder; blocking- and scratch-resistant flat polyester films
        for digital magnetic tapes)
ΙT
     7631-86-9, Silica, uses
     RL: MOA (Modifier or additive use); TEM (Technical or engineered material
     use); USES (Uses)
        (surface layer contq.; blocking- and scratch-resistant flat polyester
        films for digital magnetic tapes)
IT
     63767-37-3D, Ammonium styrenesulfonate, polymers
     RL: POF (Polymer in formulation); TEM (Technical or engineered material
     use); USES (Uses)
        (surface layer contg.; blocking- and scratch-resistant flat polyester
        films for digital magnetic tapes)
ΙT
     9016-45-9, Polyethylene glycol nonylphenyl ether
     RL: MOA (Modifier or additive use); TEM (Technical or engineered material
     use); USES (Uses)
        (surfactant; blocking- and scratch-resistant flat polyester films for
        digital magnetic tapes)
     422324-41-2, Diethylene glycol-ethylene glycol-glycidyl
     methacrylate-isophthalic acid-methyl methacrylate-5-sodiosulfoisophthalic
     acid-terephthalic acid copolymer
     RL: POF (Polymer in formulation); TEM (Technical or engineered
     material use); USES (Uses)
        (binder; blocking- and scratch-resistant flat polyester films
        for digital magnetic tapes)
     422324-41-2 HCAPLUS
RN
     1,3-Benzenedicarboxylic acid, 5-sulfo-, monosodium salt, polymer with
CN
     1,3-benzenedicarboxylic acid, 1,4-benzenedicarboxylic acid,
     1,2-ethanediol, methyl 2-methyl-2-propenoate, oxiranylmethyl
     2-methyl-2-propenoate and 2,2'-oxybis[ethanol] (9CI) (CA INDEX NAME)
     CM
     CRN 6362-79-4
```

CMF C8 H6 O7 S . Na

Na

CM 2

CRN 121-91-5 CMF C8 H6 O4

CM 3

CRN 111-46-6 CMF C4 H10 O3

 ${\tt HO-CH_2-CH_2-O-CH_2-CH_2-OH}$

CM 4

CRN 107-21-1 CMF C2 H6 O2

 ${\tt HO-CH_2-CH_2-OH}$

CM 5

CRN 106-91-2 CMF C7 H10 O3

$$\stackrel{\text{O}}{\longleftarrow} \begin{array}{c} \text{O} & \text{CH}_2 \\ \parallel & \parallel \\ \text{CH}_2\text{--O-C-C-Me} \end{array}$$

CRN 100-21-0 CMF C8 H6 O4

CM

CRN 80-62-6 CMF C5 H8 O2

$$\begin{array}{ccc} ^{\text{H}_2\text{C}} & \text{O} \\ & \parallel & \parallel \\ \text{Me-C-C-OMe} \end{array}$$

L48 ANSWER 5 OF 28 HCAPLUS COPYRIGHT 2003 ACS

AN 2002:104690 HCAPLUS

136:135581 DN

ΤI Low-odor (meth)acrylic ester binders with low viscosity and good storage stability prepared starting from aromatic polyesters applicants

Dally, Moya; De Cooman, Ria; Meyer, Werner IN

Sika A.-G., Vorm. Kaspar Winkler & Co., Switz. PA

Eur. Pat. Appl., 13 pp. SO CODEN: EPXXDW

DT Patent

LΑ English

IC ICM C09D167-07

ICS C09J167-07; C08L067-07; C08G063-21; C08J011-22

37-3 (Plastics Manufacture and Processing) Section cross-reference(s): 38, 42, 58, 60

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE 20000801 A1 20020206 EP 2000-116565 PΙ EP 1178092 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO US 2001-917674 20010731 US 2002042486 A1 20020411 20020514 JP 2001-234072 20010801 JP 2002138124 A2 PRAI EP 2000-116565 Α 20000801

Title (meth)acrylate binders, particularly useful in adhesives, coatings and floorings, comprises an oligomer A(R1)nXR2O[COR3COR4]mA or AOR2X(R1)n[COR3COR4]mA (A = CH2:C(R5)CO-; R1 = arom. polyester; R2 = substituted linear and branched C3-20 alkylene, cycloalkylene and aralkylene, substituted dioxyalkylene, trioxyalkylene or tetraoxyalkylene and substituted heterocyclic radicals; R3 = hydroxy- or carboxylic-substituted linear and branched aliph. and arom. and araliph. divalent C3-14 radical; R4 = -OR2X(R1)n- or -(R1)nXR20- or -XR20-; R5 = H, Me; X = 0, NH; n = 1-4; and m = 0-3). Thus, 475 parts PET flakes (from recycled beverage bottles), was mixed with ethoxylated neopentyl glycol 520, and tert-Bu titanate 5 parts for 2 h at 240.degree., then reacted with 460 parts methacrylic acid contg. 0.2 parts phenothiazine and 15 parts p-toluenesulfonic acid in 365 parts toluene at .apprx. 130.degree. for .apprx. 4 h and cured at room temp. with 2% benzoyl peroxide and 0.5% dimethylaniline, showing tensile strength >35 MPa, elongation at break 9%, good scratch resistance and compressive strength, and weather and chem. resistance.

ST acrylic ester binder low odor coating; adhesive arom polyester polyacrylate binder; PET recycled acrylic polyoxyalkylene binder flooring; arom polyester methacrylic acid reaction polyacrylate prepn

IT Polyoxyalkylenes, preparation

RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (acrylic-polyester-; low-odor (meth)acrylic ester binders with low viscosity and good storage stability prepd. starting from recycled polyethylene terephthalate)

IT Polyesters, preparation

RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (acrylic-polyoxyalkylene-; low-odor (meth)acrylic ester binders with low viscosity and good storage stability prepd. starting from recycled polyethylene terephthalate)

IT Polyesters, preparation

RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (acrylic; low-odor (meth)acrylic ester binders with low viscosity and good storage stability prepd. starting from recycled polyethylene terephthalate)

IT Binders

Coating materials

Recycling of plastics and rubbers

(low-odor (meth)acrylic ester binders with low viscosity and good storage stability prepd. starting from recycled polyethylene terephthalate)

IT Adhesives

Mortar

(low-odor (meth)acrylic ester binders with low viscosity and good storage stability prepd. starting from recycled polyethylene terephthalate for)

393582-81-5P, Ethylene glycol-ethoxylated neopentyl glycol-terephthlic acid copolymer dimethacrylate homopolymer 393582-83-7P, Ethylene glycol-ethoxylated neopentyl glycol-terephthlic acid copolymer methacrylate n-nonyl ester, homopolymer 393582-85-9P, Diethylene glycol-ethylene glycol-phthalic anhydride-terephthlic acid copolymer diacrylate homopolymer 393582-87-1P, Diethylene glycol-ethylene glycol-phthalic anhydride-terephthlic acid copolymer dimethacrylate homopolymer 393582-89-3P, Diethylene glycol-ethylene glycol-terephthlic acid-trimellitic anhydride copolymer methacrylate homopolymer RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(low-odor (meth)acrylic ester **binders** with low viscosity and good storage stability prepd. starting from recycled polyethylene terephthalate)

RE.CNT 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD

```
(1) Ciba Geigy Ag; EP 0514147 A 1992 HCAPLUS
(2) Ostermann & Scheiwe Gmbh & Co; EP 1002842 A 2000 HCAPLUS
(3) Sherwin Williams Co; EP 0558905 A 1993 HCAPLUS
     393582-81-5P, Ethylene glycol-ethoxylated neopentyl
     glycol-terephthlic acid copolymer dimethacrylate homopolymer
     393582-83-7P, Ethylene glycol-ethoxylated neopentyl
     glycol-terephthlic acid copolymer methacrylate n-nonyl ester, homopolymer
     393582-85-9P, Diethylene glycol-ethylene glycol-phthalic
     anhydride-terephthlic acid copolymer diacrylate homopolymer
     393582-87-1P, Diethylene glycol-ethylene glycol-phthalic
     anhydride-terephthlic acid copolymer dimethacrylate homopolymer
     393582-89-3P, Diethylene glycol-ethylene glycol-terephthlic
     acid-trimellitic anhydride copolymer methacrylate homopolymer
     RL: IMF (Industrial manufacture); POF (Polymer in
     formulation); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (low-odor (meth)acrylic ester binders with low viscosity and
        good storage stability prepd. starting from recycled polyethylene
        terephthalate)
RN
     393582-81-5 HCAPLUS
     1,4-Benzenedicarboxylic acid, polymer with .alpha.,.alpha.'-(2,2-dimethyl-
CN
     1,3-propanediyl)bis[.omega.-hydroxypoly(oxy-1,2-ethanediyl)] and
     1,2-ethanediol, bis(2-methyl-2-propenoate), homopolymer (9CI) (CA INDEX
     NAME)
     CM
          1
     CRN
          393582-80-4
          (C8 H6 O4 . C2 H6 O2 . (C2 H4 O)n (C2 H4 O)n C5 H12 O2)x . 2 C4 H6 O2
          CM
               2
          CRN
               79-41-4
               C4 H6 O2
          CMF
   CH<sub>2</sub>
Me-C-CO2H
               3
          CM
               393582-79-1
               (C8 H6 O4 . C2 H6 O2 . (C2 H4 O)n (C2 H4 O)n C5 H12 O2)x
          CMF
          CCI
               PMS
               CM
                    4
               CRN
                    82973-76-0
                    (C2 H4 O)n (C2 H4 O)n C5 H12 O2
               CMF
               CCI
                    PMS
```

$$\begin{array}{c|c} \text{HO} & \begin{array}{c|c} \text{CH}_2 - \text{CH}_2 - \text{O} \end{array} \end{array} \begin{array}{c} \text{Me} \\ \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 \end{array} \begin{array}{c} \text{OH} \\ \text{Me} \end{array}$$

CRN 107-21-1 CMF C2 H6 O2

 ${\rm HO}-{\rm CH_2}-{\rm CH_2}-{\rm OH}$

CM 6

CRN 100-21-0 CMF C8 H6 O4

RN 393582-83-7 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, polymer with .alpha.,.alpha.'-(2,2-dimethyl-1,3-propanediyl)bis[.omega.-hydroxypoly(oxy-1,2-ethanediyl)] and 1,2-ethanediol, mono(2-methyl-2-propenoate), nonyl ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 393582-82-6 CMF C9 H20 O . (C8 H6 O4 . C2 H6 O2 . (C2 H4 O)n (C2 H4 O)n C5 H12 O2)x . C4 H6 O2

CM 2

CRN 143-08-8 CMF C9 H20 O

 $Me^-(CH_2)_8-OH$

CM 3

CRN 79-41-4 CMF C4 H6 O2

$$\begin{array}{c} \text{CH}_2 \\ || \\ \text{Me-} \text{C-} \text{CO}_2 \text{H} \end{array}$$

CRN 393582-79-1

(C8 H6 O4 . C2 H6 O2 . (C2 H4 O)n (C2 H4 O)n C5 H12 O2)x CMF

CCI

CM 5

CRN 82973-76-0

 ${\tt CMF}$ (C2 H4 O)n (C2 H4 O)n C5 H12 O2

CCI PMS

CM 6

CRN 107-21-1

CMF C2 H6 O2

 $HO-CH_2-CH_2-OH$

CM7

CRN 100-21-0 C8 H6 O4 CMF

393582-85-9 HCAPLUS RN

1,4-Benzenedicarboxylic acid, polymer with 1,2-ethanediol, CN 1,3-isobenzofurandione and 2,2'-oxybis[ethanol], di-2-propenoate, homopolymer (9CI) (CA INDEX NAME)

CM

CRN 393582-84-8

CMF (C8 H6 O4 . C8 H4 O3 . C4 H10 O3 . C2 H6 O2)x . 2 C3 H4 O2

CM 2

CRN 79-10-7 CMF C3 H4 O2

CM 3

CRN 96664-01-6

CMF (C8 H6 O4 . C8 H4 O3 . C4 H10 O3 . C2 H6 O2) x

CCI PMS

CM 4

CRN 111-46-6

CMF C4 H10 O3

$${\tt HO-CH_2-CH_2-O-CH_2-CH_2-OH}$$

CM 5

CRN 107-21-1

CMF C2 H6 O2

CM 6

CRN 100-21-0

CMF C8 H6 O4

CM 7

CRN 85-44-9

CMF C8 H4 O3

RN 393582-87-1 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, polymer with 1,2-ethanediol, 1,3-isobenzofurandione and 2,2'-oxybis[ethanol], bis(2-methyl-2-propenoate), homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN · 393582-86-0

CMF (C8 H6 O4 . C8 H4 O3 . C4 H10 O3 . C2 H6 O2)x . 2 C4 H6 O2

CM 2

CRN 79-41-4 CMF C4 H6 O2

$$\begin{array}{c} \text{CH}_2 \\ || \\ \text{Me-} \text{C-} \text{CO}_2 \text{H} \end{array}$$

CM 3

CRN 96664-01-6

CMF (C8 H6 O4 . C8 H4 O3 . C4 H10 O3 . C2 H6 O2)x

CCI PMS

CM 4

CRN 111-46-6 CMF C4 H10 O3

 ${\tt HO-CH_2-CH_2-O-CH_2-CH_2-OH}$

CM 5

CRN 107-21-1 CMF C2 H6 O2

 $HO-CH_2-CH_2-OH$

CM 6

CRN 100-21-0 CMF C8 H6 O4

CM 7

CRN 85-44-9 CMF C8 H4 O3

RN 393582-89-3 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, polymer with 1,3-dihydro-1,3-dioxo-5-isobenzofurancarboxylic acid, 1,2-ethanediol and 1,3-isobenzofurandione, 2-methyl-2-propenoate, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 393582-88-2

CMF (C9 H4 O5 . C8 H6 O4 . C4 H1O O3 . C2 H6 O2) \times . \times C4 H6 O2

CM 2

CRN 79-41-4 CMF C4 H6 O2

$$\begin{array}{c} \text{CH}_2 \\ || \\ \text{Me-C-CO}_2 \text{H} \end{array}$$

CM 3

CRN 135187-98-3

CMF (C9 H4 O5 . C8 H6 O4 . C4 H10 O3 . C2 H6 O2)x

CCI PMS

CM 4

CRN 552-30-7 CMF C9 H4 O5

CRN 111-46-6 CMF C4 H10 O3

 ${\tt HO-CH_2-CH_2-O-CH_2-CH_2-OH}$

CM 6

CRN 107-21-1 CMF C2 H6 O2

 $HO-CH_2-CH_2-OH$

CM 7

CRN 100-21-0 CMF C8 H6 O4

L48 ANSWER 6 OF 28 HCAPLUS COPYRIGHT 2003 ACS

AN 2001:916762 HCAPLUS

DN 136:217421

TI On the UV curability and mechanical properties of novel binder systems derived from poly(ethylene terephthalate) (PET) waste for solventless magnetic tape manufacturing, 1 Acrylated oligoesters

AU Farahat, Medhat S.; Nikles, David E.

CS Center for Materials for Information Technology, University of Alabama, Tuscaloosa, AL, 35487-0209, USA

SO Macromolecular Materials and Engineering (2001), 286(11), 695-704 CODEN: MMENFA; ISSN: 1438-7492

PB Wiley-VCH Verlag GmbH

DT Journal

LA English

- CC 37-3 (Plastics Manufacture and Processing)
 Section cross-reference(s): 77
- PET waste obtained from beverage bottles was depolymd. by a glycolysis AΒ reaction, using diethylene glycol (DEG) as the glycolyzing system and manganese acetate as a transesterification catalyst. The glycolysis reaction was conducted at two different molar ratios of PET:DEG, namely 1:2.15 and 1:1.03, for the sake of obtaining oligoester polyols of varying mol. wts. The hydroxyl values of the obtained oligoesters were 361 and 330 mg KOH/g. Modification of these oligoester polyols was carried out by acrylation reactions of the available hydroxyl groups by acryloyl chloride. This gave acrylated oligoesters curable under UV or electron beam irradn. The curability of these newly synthesized acrylated oligoesters was tested by UV irradn., in the presence of 2-benzyl-2-dimethylamino-1-(4-morpholinophenyl)-1-butanone (BDMB) as a photo initiator. This gave cured films of high mech. properties when the acrylated oligoesters were either cured alone or as mixts. with other com. available diacrylate/dimethacrylate monomers. The measured tensile properties were in the range of 4.62-45 MPa for max. tensile strength and 0.074-2.0 GPa for Young's modulus.
- ST polyethylene terephthalate waste polyol acrylate binder; diethylene glycol reaction product PET acrylate binder; magnetic tape binder PET polyol acrylate
- IT Crosslinking catalysts
 Glycolysis
 Magnetic tapes
 Polymerization catalysts
 Tensile strength
 Waste plastics and rubbers
 Young's modulus

(UV curability and mech. properties of oligoester acrylates from poly(ethylene terephthalate) wastes for binders for solventless

magnetic tape manufg.)

IT Bottles

(polyester; UV curability and mech. properties of oligoester acrylates from poly(ethylene terephthalate) wastes for binders for solventless magnetic tape manufg.)

- IT Polyesters, preparation
 - RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (reaction products with diethylene glycol, acrylates, homopolymers and copolymers; UV curability and mech. properties of oligoester acrylates from poly(ethylene terephthalate) wastes for binders for solventless magnetic tape manufg.)
- IT 119313-12-1, 2-Benzyl-2-dimethylamino-1-(4-morpholinophenyl).-1-butanone RL: CAT (Catalyst use); USES (Uses)
 (UV curability and mech. properties of oligoester acrylates from

poly(ethylene terephthalate) wastes for binders for solventless magnetic tape manufg.)

IT 109-16-0DP, Triethylene glycol dimethacrylate, polymers with poly(ethylene terephthalate) -diethylene glycol reaction product acrylates Diethylene glycol, reaction products with poly(ethylene terephthalate), 814-68-6DP, Acryloyl chloride, acrylates, homopolymers and copolymers esters with poly(ethylene terephthalate)-diethylene glycol reaction 2358-84-1DP, Diethylene glycol products, homopolymers and copolymers dimethacrylate, polymers with poly(ethylene terephthalate)-diethylene 4074-88-8DP, Diethylene glycol glycol reaction product acrylates diacrylate, polymers with poly(ethylene terephthalate)-diethylene glycol reaction product acrylates 13048-33-4DP, polymers with poly(ethylene terephthalate) - diethylene glycol reaction product acrylates

25038-59-9DP, Poly(ethylene terephthalate), reaction products with diethylene glycol, acrylates, homopolymers and copolymers 58264-26-9DP, Hexanediol dimethacrylate, polymers with poly(ethylene terephthalate) - diethylene glycol reaction product acrylates RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (UV curability and mech. properties of oligoester acrylates from poly(ethylene terephthalate) wastes for binders for solventless magnetic tape manufg.) THERE ARE 42 CITED REFERENCES AVAILABLE FOR THIS RECORD RE.CNT (1) ASTM; Standard Test Methods for Tensile Properties of Thin Plastic Sheeting D 882-91 (2) Abdel-Azim, A; Polym Adv Technol 1995, V6, P688 HCAPLUS (3) Ahn, K; J Appl Polym Sci 1999, V71, P2033 HCAPLUS (4) Beck, R; National Post-Consumer Plastics Recycling Rate Study 1993 (5) Chen, J; J Appl Polym Sci 1999, V73, P35 HCAPLUS (6) Cheng, S; Chemtech 1995, V25, P35 HCAPLUS (7) Cheng, S; Waste Manage 1995, V15, P257 HCAPLUS (8) Davies, C; Polym Int 1996, V41, P215 HCAPLUS (9) Farahat, M; Macromol Mater Eng 2000, V283, P1 HCAPLUS (10) Fritz, D; Anal Chem 1979, V51, P7 HCAPLUS (11) Fritz, J; Anal Chem 1959, V31, P1808 HCAPLUS (12) Gao, F; Polym Prepr (Am Chem Soc, Div Polym Chem) 2000, V41, P125 HCAPLUS (13) Gueclue, G; J Appl Polym Sci 1998, V69, P2311 (14) Halacheva, N; Polymer 1995, V36, P867 HCAPLUS (15) Harper, C; Handbook of Plastics, Elastomers and Composites 1996 (16) Hoyle, C; ACS Symp Ser 1990, V417, P92 (17) Koster, E; Magnetic Recording - Technology 1987, V1, P98 (18) Krzan, A; J Appl Polym Sci 1998, V69, P1115 HCAPLUS (19) Krzan, A; Polym Adv Technol 1999, V10, P603 HCAPLUS (20) Lee, S; J Appl Polym Sci 1994, V52, P869 HCAPLUS (21) Lusinchi, J; J Appl Polym Sci 1998, V69, P657 (22) Malucelli, G; J Appl Polym Sci 1996, V65, P491 (23) Malucelli, G; Polymer 1996, V37, P2565 HCAPLUS (24) Mancini, S; J Appl Polym Sci 2000, V76, P266 HCAPLUS (25) Nikles, D; ACS Symp Ser 2000, V766, P18 (26) Odian, G; Principles of Polymerization 1981, P40 (27) Oku, A; J Appl Polym Sci 1997, V63, P595 HCAPLUS (28) Priola, A; Polymer 1993, V34, P3653 HCAPLUS (29) Rader, C; ACS Symp Ser 1995, V609, P152(30) Rebeiz, K; ACI Mater J 1994, V91, P313 HCAPLUS (31) Rebeiz, K; Cem Concr Compos 1995, V17, P119 HCAPLUS (32) Rebeiz, K; J Appl Polym Sci 1992, V44, P1649 HCAPLUS (33) Rebeiz, K; J Mater Civil Eng 1994, V6, P150 HCAPLUS (34) Rebeiz, K; J Mater Civil Eng 1995, V7, P129 HCAPLUS (35) Rebeiz, K; J Struct Eng 1995, V121, P1370 (36) Siggia, S; Quantitative organic analysis via functional group analysis 1963, P8 (37) Sorenson, W; Preparative Methods of Polymer Chemistry 2001, P358 (38) Stetzler, R; Anal Chem 1962, V34, P194 HCAPLUS (39) Suh, D; Polymer 2000, V41, P461 HCAPLUS (40) Tong, S; Polymer 1983, V24, P469 HCAPLUS

(41) Vaidya, U; Ind Eng Chem Res 1987, V26, P194 HCAPLUS (42) Vaidya, U; J Appl Polym Sci 1987, V34, P235 HCAPLUS

25038-59-9DP, Poly(ethylene terephthalate), reaction products with

diethylene glycol, acrylates, homopolymers and copolymers RL: PRP (Properties); SPN (Synthetic preparation); PREP

(Preparation)

(UV curability and mech. properties of **oligoester** acrylates from poly(ethylene terephthalate) wastes for binders for solventless magnetic tape manufg.)

RN 25038-59-9 HCAPLUS

CN Poly(oxy-1,2-ethanediyloxycarbonyl-1,4-phenylenecarbonyl) (9CI) (CA INDEX NAME)

L48 ANSWER 7 OF 28 HCAPLUS COPYRIGHT 2003 ACS

AN 2001:662307 HCAPLUS

DN 135:345389

TI Novel binder systems derived from poly(ethylene terephthalate) PET waste for solventless magnetic tape manufacturing. II. Investigation on the mechanical properties of the methacrylated oligoesters

AU Farahat, Medhat S.; Nikles, David E.

CS Center for Materials for Information Technology, University of Alabama, Tuscaloosa, AL, 35487-0209, USA

SO Polymer Preprints (American Chemical Society, Division of Polymer Chemistry) (2001), 42(2), 558-559
CODEN: ACPPAY; ISSN: 0032-3934

PB American Chemical Society, Division of Polymer Chemistry

DT Journal; (computer optical disk)

LA English

CC 38-3 (Plastics Fabrication and Uses) Section cross-reference(s): 77

AB The investigation on synthesizing new UV curable acrylated oligoesters, initially derived from PET waste, for the solventless magnetic tape manufg., was successfully established. Methacrylated oligoesters obtained by glycolyzing PET with diethylene glycol (DEG) at the molar ratio of PET:DEG (1:1.03) showed better mech. properties than those results obtained by glycolyzing PET at the molar ratio of PET:DEG (1:2.15). The idea of the current investigation is a novel one and it opens the door for many new applications for the oligoesters derived from PET waste.

ST methacrylated polyethylene terephthalate oligomer waste mech property; solventless magnetic tape methacrylated PET waste

IT Strain

(at break; mech. properties of methacrylated oligoesters from PET wastes as binder systems for solventless magnetic tape manufg.)

IT Polyesters, uses

RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(glycolyzed, reaction products with methacryloyl chloride and other (meth)acrylates; mech. properties of methacrylated oligoesters from PET wastes as binder systems for solventless magnetic tape manufg.)

IT Binders

```
Magnetic tapes
     Solid wastes
     Tensile strength
     Young's modulus
        (mech. properties of methacrylated oligoesters from PET wastes as
       binder systems for solventless magnetic tape manufg.)
ΙT
     Crosslinking
        (photochem.; mech. properties of methacrylated oligoesters from PET
        wastes as binder systems for solventless magnetic tape manufg.)
     109-16-0DP, Triethylene glycol dimethacrylate, reaction products with
IT
                                                        111-46-6DP, Diethylene
     glycolyzed PET chloride and other (meth)acrylates
     glycol, reaction products with poly(ethylene terephthalate) and
     methacryloyl chloride and other (metha) acrylates
                                                        920-46-7DP,
     Methacryloyl chloride, reaction products with glycolyzed poly(ethylene
                                                2358-84-1DP, Diethylene glycol
     terephthalate) and other (meth)acrylates
     dimethacrylate, reaction products with glycolyzed PET chloride and other
     (meth)acrylates 13048-33-4DP, reaction products with glycolyzed PET
     chloride and other (meth)acrylates 25038-59-9DP, Poly(ethylene
     terephthalate), glycolyzed, reaction products with methacryloyl
     chloride and other (meth) acrylates 58264-26-9DP, Hexanediol
     dimethacrylate, reaction products with glycolyzed PET chloride and other
     (meth)acrylates
     RL: IMF (Industrial manufacture); PRP (Properties); TEM
     (Technical or engineered material use); PREP (Preparation); USES
     (Uses)
        (mech. properties of methacrylated oligoesters from
        PET wastes as binder systems for solventless magnetic tape
        manufq.)
              THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE.CNT
(1) Anon; ASTM Designation: D V882-91
(2) Cheng, S; Waste Management 1995, V15(4), P257 HCAPLUS
(3) Halacheva, N; Polymer 1995, V36(4), P867 HCAPLUS
(4) Harper, C; Handbook of Plastics, Elastomers and Composites, 3rd ed 1996
(5) Malucelli, G; Polymer 1996, V37(12), P2565 HCAPLUS
(6) Rader, C; ACS symposium series 1995, V609, P152
(7) Sorenson, W; Preparative Method of Polymer Chemistry 1968, P155
     25038-59-9DP, Poly(ethylene terephthalate), glycolyzed, reaction
TT
     products with methacryloyl chloride and other (meth)
     acrylates
     RL: IMF (Industrial manufacture); PRP (Properties); TEM
     (Technical or engineered material use); PREP (Preparation); USES
        (mech. properties of methacrylated oligoesters from
        PET wastes as binder systems for solventless magnetic tape
        manufg.)
     25038-59-9 HCAPLUS
RN
     Poly(oxy-1,2-ethanediyloxycarbonyl-1,4-phenylenecarbonyl) (9CI) (CA INDEX
CN
     NAME)
```

L48 ANSWER 8 OF 28 HCAPLUS COPYRIGHT 2003 ACS

ÁN 2001:509864 HCAPLUS

DN 136:184443

TI Novel binder systems derived from poly(ethylene terephthalate) PET waste for solventless magnetic tape manufacturing. I. Investigation of the mechanical properties of the acrylated oligomers

AU Farahat, Medhat S.; Nikles, David E.

CS Center for Materials for Information Technology, University of Alabama, Tuscaloosa, AL, 35487-0209, USA

SO International SAMPE Symposium and Exhibition (2001), 46(2001: A Materials and Processes Odyssey, Book 1), 172-178

CODEN: ISSEEG; ISSN: 0891-0138

PB Society for the Advancement of Material and Process Engineering

DT Journal

LA English

CC 37-3 (Plastics Manufacture and Processing)
Section cross-reference(s): 77

AB PET waste obtained from beverage bottles was first depolymd. by glycolysis reaction using diethylene glycol (DEG) as glycolyzing system and manganese acetate as a transesterification catalyst. The glycolysis reaction was conducted at two different molar ratios of PET:DEG (1:2.15 and 1:1.03) to obtain oligoester polyols of varying mol. wts. The obtained oligoester polyols were purified and characterized for hydroxyl values (mg KOH/g). Modification of the obtained oligoester polyols was carried out by acrylation or methacrylation of the available hydroxyl groups by acryloyl chloride or methacryloyl chloride. The curability of these newly synthesized acrylated/methacrylated oligoesters was tested by UV irradn. in the presence of 2-benzyl-2-dimethylamino-1-(4-morphlinophenyl)-1-butanone (BDMB) as a photoinitiator. The films of these cured polymers showed tensile strength in the range of 4.62 - 45 MPa and Young's modulus in the range of 0.074 - 2.0 GPa.

ST polyethylene terephthalate bottle waste glycolysis recycling; acrylate methacrylate oligoester polyol UV radical polymn; recycled polyester mech property

IT Polyesters, properties

RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PRP (Properties); RCT (Reactant); PROC (Process); RACT (Reactant or reagent)

(PET bottle waste recycling by glycolysis and prepn. of oligoesters from)

IT Recycling of plastics and rubbers

(PET bottle waste recycling for prepn. of acrylic and methacrylic PET polymers)

IT Glycolysis

(acrylic and methacrylic PET polymers prepd. from PET bottle waste

```
recycling by)
IT
     Polyesters, preparation
    RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (acrylic; PET bottle waste recycling by glycolysis and prepn. of
        oligoesters for synthesis of acrylic and methacrylic PET polymers)
IT
     Bottles
        (beverage; PET bottle waste recycling for prepn. of acrylic and
        methacrylic PET polymers)
ΙT
    Elongation, mechanical
    Tensile strength
    Young's modulus
        (of acrylic and methacrylic PET polymers prepd. from PET bottle waste
        recycling by glycolysis)
ΙT
     Polymerization
        (photochem., radical; PET bottle waste recycling for prepn. of acrylic
        and methacrylic polymers by)
     111-46-6, Diethylene glycol, reactions
ΙT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (PET bottle waste recycling by glycolysis and prepn. of oligoesters by
        transesterification in the presence of)
     400615-97-6P 400615-98-7P 400615-99-8P
TΤ
     400616-00-4P 400616-01-5P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP
     (Preparation)
        (PET bottle waste recycling by glycolysis and prepn. of
        oligoesters for synthesis of acrylic and methacrylic PET
        polymers)
ΙT
     400615-96-5P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP
     (Preparation)
        (PET bottle waste recycling by glycolysis and prepn. of
        oligoesters for synthesis of acrylic and methacrylic PET-based
        polymers)
IT
     25038-59-9, Poly(ethylene terephthalate), properties
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
    process); PRP (Properties); RCT (Reactant); PROC (Process); RACT (Reactant
     or reagent)
        (PET bottle waste recycling by glycolysis and prepn. of oligoesters
        from)
RE.CNT
       30
              THERE ARE 30 CITED REFERENCES AVAILABLE FOR THIS RECORD
RF.
(1) ASTM; ASTM Designation: D 1708-96
(2) Abdel-Azim, A; Polym Adv Technol 1995, V6, P688 HCAPLUS
(3) Ahn, K; J Appl Polym Sci 1999, V71, P2033 HCAPLUS
(4) Beck, R; National Post-Consumer Plasties Recycling Rate Study 1993
(5) Chen, J; J Appl Polym Sci 1999, V73, P35 HCAPLUS
(6) Davies, C; Polymer International 1996, V41(3), P215 HCAPLUS
(7) Farahat, M; Macromol Mater Eng 2000, V283, P1 HCAPLUS
(8) Fritz, D; Anal Chem 1979, V51(1), P7 HCAPLUS
(9) Fritz, J; Anal Chem 1959, V31(11), P1808
(10) Gao, F; Polymer preprints 2000, V4(1), P125
(11) Guclu, G; J Appl Polym Sci 1998, V69(12), P2311 HCAPLUS
(12) Halacheva, N; Polymer 1995, V36(4), P867 HCAPLUS
(13) Harper, C; Handbook of Plastics 1996
(14) Hoyle, C; Radiation Curing of Polymeric Materials 1990, P92
(15) Krzan, A; J Appl Polym Sci 1998, V69(6), P1115 HCAPLUS
(16) Krzan, A; Polym Adv Technol 1999, V10(10), P603 HCAPLUS
(17) Lee, S; J Appl Polym Sci 1994, V52, P869 HCAPLUS
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(18) Lusinchi, J; J Appl Polym Sci 1998, V69(4), P657
(19) Malucelli, G; J Appl Polym Sci 1996, V65, P491
(20) Malucelli, G; Polymer 1996, V37(12), P2565 HCAPLUS
(21) Mancini, S; J Appl Polym Sci 2000, V76(2), P266 HCAPLUS
(22) Odian, G; Principles of Polymerization 1981, P40
(23) Oku, A; J Appl Polym Sci 1997, V63(5), P595 HCAPLUS
(24) Priola, A; Polymer 1993, V34(17), P3653 HCAPLUS
(25) Rader, C; ACS symposium series 1995, V609, P152
(26) Rebeiz, K; J Appl Polym Sci 1992, V44, P1649 HCAPLUS
(27) Siggia, S; Quantitative organic analysis via functional group analysis
    1963, P8
(28) Sorenson, W; Preparative Method of Polymer Chemistry 1968, P155
(29) Stetzler, R; Anal Chem 1962, V34(2), P194
(30) Tong, S; Polymer 1983, V24, P469 HCAPLUS
IT
     400615-97-6P 400615-98-7P 400615-99-8P
     400616-00-4P 400616-01-5P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP
     (Preparation)
        (PET bottle waste recycling by glycolysis and prepn. of
        oligoesters for synthesis of acrylic and methacrylic PET
        polymers)
     400615-97-6 HCAPLUS
RN
     1,4-Benzenedicarboxylic acid, polymer with 1,2-ethanediol and
CN
     2,2'-oxybis[ethanol], 2-propenoate, polymer with oxydi-2,1-ethanediyl
     di-2-propenoate (9CI) (CA INDEX NAME)
     CM
          1
     CRN
         4074-88-8
     CMF
          C10 H14 O5
         0
H_2C = CH - C - O - CH_2 - CH_2 - O - CH_2 - CH_2 - O - C - CH = CH_2
     CM
          2
          400615-95-4
     CRN
          (C8 H6 O4 . C4 H10 O3 . C2 H6 O2)x . x C3 H4 O2
     CMF
          CM
               3
               79-10-7
          CRN
          CMF C3 H4 O2
```

CM 4

CRN 25052-77-1

CMF (C8 H6 O4 . C4 H10 O3 . C2 H6 O2)x

CCI PMS

CM 5

CRN 111-46-6 CMF C4 H10 O3

но- сн2- сн2- о- сн2- сн2- он

CM 6

CRN 107-21-1 CMF C2 H6 O2

 ${\rm HO-CH_2-CH_2-OH}$

CM 7

CRN 100-21-0 CMF C8 H6 O4

RN 400615-98-7 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, polymer with 1,2-ethanediol and 2,2'-oxybis[ethanol], 2-propenoate, polymer with oxydi-2,1-ethanediyl bis(2-methyl-2-proenoate) (9CI) (CA INDEX NAME)

CM 1

CRN 2358-84-1 CMF C12 H18 O5

CM 2

CRN 400615-95-4

CMF (C8 H6 O4 . C4 H10 O3 . C2 H6 O2)x . x C3 H4 O2

CM 3

CRN 79-10-7 CMF C3 H4 O2

CM 4

CRN 25052-77-1

CMF (C8 H6 O4 . C4 H10 O3 . C2 H6 O2) \times

CCI PMS

CM 5

CRN 111-46-6 CMF C4 H10 O3

 $HO-CH_2-CH_2-O-CH_2-CH_2-OH$

CM 6

CRN 107-21-1 CMF C2 H6 O2

 $HO-CH_2-CH_2-OH$

CM 7

CRN 100-21-0 CMF C8 H6 O4

RN 400615-99-8 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, polymer with 1,2-ethanediol and 2,2'-oxybis[ethanol], 2-propenoate, polymer with 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate) (9CI) (CA INDEX NAME)

CM 1

CRN 109-16-0 CMF C14 H22 O6

CRN 400615-95-4

(C8 H6 O4 . C4 H10 O3 . C2 H6 O2)x . x C3 H4 O2 CMF

CM

CRN 79-10-7 CMF C3 H4 O2

CM

CRN 25052-77-1

(C8 H6 O4 . C4 H10 O3 . C2 H6 O2)x CMF

CCI PMS

CM 5

CRN 111-46-6 CMF C4 H10 O3

HO-CH2-CH2-O-CH2-CH2-OH

CM 6

CRN 107-21-1

CMF C2 H6 O2

 $HO-CH_2-CH_2-OH$

CM 7

CRN 100-21-0 CMF C8 H6 O4

RN 400616-00-4 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, polymer with 1,2-ethanediol and 2,2'-oxybis[ethanol], 2-propenoate, polymer with 1,6-hexanediyl di-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 13048-33-4 CMF C12 H18 O4

CM 2

CRN 400615-95-4

CMF (C8 H6 O4 . C4 H10 O3 . C2 H6 O2)x . x C3 H4 O2

CM 3

CRN 79-10-7 CMF C3 H4 O2

CM 4

CRN 25052-77-1

CMF (C8 H6 O4 . C4 H10 O3 . C2 H6 O2)x

CCI PMS

CM 5

CRN 111-46-6 CMF C4 H10 O3

$${\tt HO-CH_2-CH_2-O-CH_2-CH_2-OH}$$

CM 6

CRN 107-21-1 CMF C2 H6 O2

но- сн2- сн2- он

CM 7

CRN 100-21-0 CMF C8 H6 O4

RN 400616-01-5 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, polymer with 1,2-ethanediol and 2,2'-oxybis[ethanol], 2-propenoate, polymer with 1,6-hexanediyl bis(2-methyl-2-propenoate) (9CI) (CA INDEX NAME)

CM 1

CRN 6606-59-3 CMF C14 H22 O4

CM 2

CRN 400615-95-4

CMF (C8 H6 O4 . C4 H10 O3 . C2 H6 O2)x . x C3 H4 O2

CM 3

CRN 79-10-7 CMF C3 H4 O2

CM 4

CRN 25052-77-1

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ZALUKAEVA 09/917674 Page 45
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CMF (C8 H6 O4 . C4 H10 O3 . C2 H6 O2)× CCI PMS

CM 5

CRN 111-46-6 CMF C4 H10 O3

 $HO-CH_2-CH_2-O-CH_2-CH_2-OH$

CM 6

CRN 107-21-1 CMF C2 H6 O2

 ${\rm HO}-{\rm CH_2}-{\rm CH_2}-{\rm OH}$

CM 7

CRN 100-21-0 CMF C8 H6 O4

IT 400615-96-5P

RL: PRP (Properties); SPN (Synthetic preparation); PREP

(Preparation)

(PET bottle waste recycling by glycolysis and prepn. of **oligoesters** for synthesis of acrylic and methacrylic PET-based polymers)

RN 400615-96-5 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, polymer with 1,2-ethanediol and 2,2'-oxybis[ethanol], 2-propenoate, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 400615-95-4

CMF (C8 H6 O4 . C4 H10 O3 . C2 H6 O2)x . x C3 H4 O2

CM 2

CRN 79-10-7 CMF C3 H4 O2

25052-77-1 CRN

(C8 H6 O4 . C4 H10 O3 . C2 H6 O2)x CMF

CCI PMS

CM

CRN 111-46-6 CMF C4 H10 O3

 ${\tt HO-CH_2-CH_2-O-CH_2-CH_2-OH}$

5 CM

CRN 107-21-1 CMF C2 H6 O2

 ${\tt HO-CH_2-CH_2-OH}$

CM 6

CRN 100-21-0 CMF C8 H6 O4

L48 ANSWER 9 OF 28 HCAPLUS COPYRIGHT 2003 ACS

2000:362692 HCAPLUS ΑN

DN 133:11961

Magnetic recording media using polyesters with hydroxyl group in the ΤI branches as binders for magnetic layers

IN Yatsuka, Takeshi; Doi, Kuniyuki

PΑ

Toyobo Co., Ltd., Japan Jpn. Kokai Tokkyo Koho, 9 pp. so

CODEN: JKXXAF

DTPatent

LΑ Japanese

IC ICM G11B005-702

ICS C09D167-00 77-8 (Magnetic Phenomena) CC Section cross-reference(s): 38 FAN.CNT 1 APPLICATION NO. DATE KIND DATE PATENT NO. _____ _____ A2 20000530 JP 1998-320770 19981111 JP 2000149241 PRAI JP 1998-320770 19981111 The magnetic recording media employs binders of modified polyesters prepd. AB by depolymn. of (a) polyesters with no.-av. mol. wt. (Mn) 4000-100,000 by using (b) compds. with Mn 300-5000 and bearing .gtoreq. 3 OH. The binders exhibit excellent dispersibility, filling property, abrasion resistance, and heat resistance. polyester hydroxyl branch magnetic recording tape; depolymn polyester ST polyhydric compd magnetic tape TIPolyesters, uses RL: DEV (Device component use); PNU (Preparation, unclassified); PREP (Preparation); USES (Uses) (acrylic; magnetic recording media using polyesters with hydroxyl group in the branches as binders for magnetic layers) ΙT Polyesters, uses RL: DEV (Device component use); PNU (Preparation, unclassified); PREP (Preparation); USES (Uses) (depolymn.; magnetic recording media using polyesters with hydroxyl group in the branches as binders for magnetic layers) IT Depolymerization (in polyester prepn.; magnetic recording media using polyesters with hydroxyl group in the branches as binders for magnetic layers) Magnetic tapes IT (magnetic recording media using polyesters with hydroxyl group in the branches as binders for magnetic layers) 25618-55-7, Polyglycerin ΙT RL: RCT (Reactant); RACT (Reactant or reagent) (PGL 10, polyester depolymn. with; magnetic recording media using polyesters with hydroxyl group in the branches as binders for magnetic lavers) IT 9005-65-6, Polyoxyethylene sorbitan monooleate 26916-03-0, Butvl acrylate-2-hydroxyethyl methacrylate-styrene copolymer RL: RCT (Reactant); RACT (Reactant or reagent) (polyester depolymn. with; magnetic recording media using polyesters with hydroxyl group in the branches as binders for magnetic layers) 270257-72-2P, Coronate HX-1,4-cyclohexanedimethanol-ethylene IT glycol-isophthalic acid-neopentyl glycol-PGL 10-5-sodiosulfoisophthalic acid-trimellitic anhydride copolymer 270257-73-3P, Coronate HX-ethylene glycol-neopentyl glycol monohydroxypivalate-PGL 10-5-sodiosulfoisophthalic acid-terephthalic acid copolymer 270257-74-4P, Coronate HX-ethylene glycol-isophthalic acid-neopentyl glycol-terephthalic acid-PGL 10 270257-75-5P, Coronate HX-ethylene glycol-isophthalic copolymer acid-neopentyl glycol-PGL 10-5-sodiosulfoisophthalic acid-terephthalic acid copolymer 270257-76-6P, Butyl acrylate-Coronate HX-ethylene glycol-2-hydroxyethyl methacrylate-isophthalic acid-neopentyl glycol-5-sodiosulfoisophthalic acid-styrene-terephthalic acid copolymer 270257-77-7P, Coronate HX-ethylene glycol-isophthalic acid-neopentyl glycol-polyoxyethylene sorbitan monooleate-5-sodiosulfoisophthalic acid-terephthalic acid copolymer 270257-78-8P, 1,4-Cyclohexanedimethanolethylene glycol-HDI trimer-isophthalic acid-neopentyl glycol-PGL 10-5-sodiosulfoisophthalic acid-trimellitic anhydride copolymer

270257-79-9P, Ethylene glycol-HDI trimer-neopentyl glycol

IT

RN

CN

```
monohydroxypivalate-PGL 10-5-sodiosulfoisophthalic acid-terephthalic acid
                270257-80-2P, Ethylene glycol-HDI trimer-isophthalic
    copolymer
    acid-neopentyl glycol-terephthalic acid-PGL 10 copolymer 270257-81-3P,
    Ethylene glycol-HDI trimer-isophthalic acid-neopentyl glycol-PGL
    10-5-sodiosulfoisophthalic acid-terephthalic acid copolymer
    270257-82-4P, Butyl acrylate-ethylene glycol-HDI
    trimer-2-hydroxyethyl methacrylate-isophthalic acid-neopentyl
    glycol-5-sodiosulfoisophthalic acid-styrene-terephthalic acid copolymer
    270257-83-5P, Ethylene glycol-HDI trimer-isophthalic acid-neopentyl
    glycol-polyoxyethylene sorbitan monooleate-5-sodiosulfoisophthalic
    acid-terephthalic acid copolymer
    RL: DEV (Device component use); PNU (Preparation, unclassified); PREP
     (Preparation); USES (Uses)
        (prepd. by depolymn.; magnetic recording media using polyesters with
       hydroxyl group in the branches as binders for magnetic
    270257-76-6P, Butyl acrylate-Coronate HX-ethylene
    glycol-2-hydroxyethyl methacrylate-isophthalic acid-neopentyl
    glycol-5-sodiosulfoisophthalic acid-styrene-terephthalic acid copolymer
    270257-82-4P, Butyl acrylate-ethylene glycol-HDI
    trimer-2-hydroxyethyl methacrylate-isophthalic acid-neopentyl
    glycol-5-sodiosulfoisophthalic acid-styrene-terephthalic acid copolymer
    RL: DEV (Device component use); PNU (Preparation, unclassified); PREP
     (Preparation); USES (Uses)
        (prepd. by depolymn.; magnetic recording media using polyesters with
       hydroxyl group in the branches as binders for magnetic
       layers)
    270257-76-6 HCAPLUS
    1,3-Benzenedicarboxylic acid, 5-sulfo-, monosodium salt, polymer with
    1,3-benzenedicarboxylic acid, 1,4-benzenedicarboxylic acid, butyl
    2-propenoate, Coronate HX, 2,2-dimethyl-1,3-propanediol, 1,2-ethanediol,
    ethenylbenzene and 2-hydroxyethyl 2-methyl-2-propenoate (9CI) (CA INDEX
    NAME)
    CM
         1
    CRN 144245-98-7
         Unspecified
    CMF
    CCI PMS, MAN
*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
    CM
         2
    CRN 6362-79-4
    CMF C8 H6 O7 S . Na
```

Na

CM 3

CRN 868-77-9 CMF C6 H10 O3

CM 4

CRN 141-32-2 CMF C7 H12 O2

$$\begin{array}{c} \text{O} \\ \parallel \\ \text{n-BuO-C-CH-----} \text{CH}_{2} \end{array}$$

CM 5

CRN 126-30-7 CMF C5 H12 O2

CM 6

CRN 121-91-5 CMF C8 H6 O4

CRN 107-21-1 CMF C2 H6 O2

 ${\tt HO-CH_2-CH_2-OH}$

CM 8

CRN 100-42-5 CMF C8 H8

 $H_2C = CH - Ph$

CM 9

CRN 100-21-0 CMF C8 H6 O4

RN 270257-82-4 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-sulfo-, monosodium salt, polymer with 1,3-benzenedicarboxylic acid, 1,4-benzenedicarboxylic acid, butyl 2-propenoate, 1,6-diisocyanatohexane trimer, 2,2-dimethyl-1,3-propanediol, 1,2-ethanediol, ethenylbenzene and 2-hydroxyethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 6362-79-4 CMF C8 H6 O7 S . Na

Na

CM 2

CRN 868-77-9 CMF C6 H10 O3

CM 3

CRN 141-32-2 CMF C7 H12 O2

$$\begin{array}{c} \text{O} \\ \parallel \\ \text{n-BuO-C-CH} \end{array}$$

CM 4

CRN 126-30-7 CMF C5 H12 O2

CM 5

CRN 121-91-5 CMF C8 H6 O4

CM 6

CRN 107-21-1 CMF C2 H6 O2 ·

 $_{\text{HO-CH}_2-\text{CH}_2-\text{OH}}$

CM T

CRN 100-42-5 CMF C8 H8

H2C=CH-Ph

CM 8

CRN 100-21-0 CMF C8 H6 O4

CM S

CRN 28574-90-5

CMF (C8 H12 N2 O2)3

CCI PMS

CM 10

CRN 822-06-0' CMF C8 H12 N2 O2

ocn-(cH₂)₆-nco

L48 ANSWER 10 OF 28 HCAPLUS COPYRIGHT 2003 ACS

KATHLEEN FULLER EIC 1700/PARKER LAW 308-4290

```
1999:814008 HCAPLUS
ΑN
DN
     132:57093
     Toner with good wax dispersibility
ΤI
     Matsunaga, Satoshi; Endo, Genichi; Michigami, Tadashi; Mizoo, Yuichi
IN
     Canon K. K., Japan
PA
     Jpn. Kokai Tokkyo Koho, 21 pp.
SO
     CODEN: JKXXAF
DT
     Patent
     Japanese
LΑ
     ICM G03G009-087
IC
     ICS G03G009-08
     74-3 (Radiation Chemistry, Photochemistry, and Photographic and Other
CC
     Reprographic Processes)
FAN.CNT 1
                                                            DATE
     PATENT NO.
                      KIND DATE
                                           APPLICATION NO.
                            _____
                      A2
                            19991224
                                           JP 1998-155094
                                                            19980604
     JP 11352720
PΙ
PRAI JP 1998-155094
                            19980604
     The toner, useful for electrophotog., electrostatic recording, toner-jet
     recording, etc., contains a binder resin, a colorant, and a wax, wherein
     the binder resin is manufd. by polymn. of arom. vinyl monomer,
     (meth)acrylate ester, and CO2H-contg. vinyl monomer to give vinyl
     copolymer (A) and polymn. of polyester-forming acid and alc. in the
     presence of wax and A. The wax is uniformly dispersed into the binder
     resin to give a toner showing good crushability, fixability, and offset
     resistance.
     toner electrophotog wax dispersibility binder resin; vinyl polymer
ST
     polyester binder electrophotog toner
     Polyesters, preparation
ΙT
     RL: PNU (Preparation, unclassified); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (binder; toner with good wax dispersibility)
     Electrophotographic toners
IT
        (toner with good wax dispersibility)
     Hydrocarbon waxes, uses
IT
     RL: TEM (Technical or engineered material use); USES (Uses)
        (toner with good wax dispersibility)
IT
     Binders
        (vinyl polymers and polyesters; toner with good wax dispersibility)
     25586-20-3P, Acrylic acid-butyl acrylate-styrene copolymer 96360-62-2P
IT
     252854-86-7P 252854-88-9P 252854-90-3P 252854-92-5P
     252854-93-6P
     RL: PNU (Preparation, unclassified); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (binder; toner with good wax dispersibility)
IT
     252854-90-3P 252854-92-5P
     RL: PNU (Preparation, unclassified); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (binder; toner with good wax dispersibility)
     252854-90-3 HCAPLUS
RN
     1,4-Benzenedicarboxylic acid, polymer with bis[2-[4-[1-[4-(2-
CN
     hydroxyethoxy)phenyl]-1-methylethyl]phenoxy]ethyl] (2E)-2-butenedioate,
     butyl 2-propenoate, 1,3-dihydro-1,3-dioxo-5-isobenzofurancarboxylic acid,
     ethenylbenzene, .alpha.,.alpha.'-[(1-methylethylidene)di-4,1-
     phenylene]bis[.omega.-hydroxypoly(oxy-1,2-ethanediyl)] and
     .alpha.,.alpha.'-{(1-methylethylidene)di-4,1-phenylene]bis[.omega.-
     hydroxypoly[oxy(methyl-1,2-ethanediyl)]] (9CI) (CA INDEX NAME)
```

CM 1

CRN 252854-89-0 CMF C42 H48 O10

Double bond geometry as shown.

PAGE 1-A

PAGE 1-B

CM 2

CRN 37353-75-6

CMF (C3 H6 O)n (C3 H6 O)n C15 H16 O2

CCI IDS, PMS

HO
$$(C3H_6)-O$$
 Me Me Me Me

CM 3

CRN 32492-61-8

CMF (C2 H4 O)n (C2 H4 O)n C15 H16 O2

CCI PMS

$$HO \longrightarrow CH_2 - CH_2 - O \longrightarrow Me$$
 Me
 Me
 Me
 Me

CRN 552-30-7 CMF C9 H4 O5

CM 5

CRN 141-32-2 CMF C7 H12 O2

CM 6

CRN 100-42-5 CMF C8 H8

 $_{\rm H2C} = _{\rm CH-Ph}$

CM '

CRN 100-21-0 CMF C8 H6 O4

RN 252854-92-5 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, polymer with bis[2-[4-[1-[4-(2-hydroxyethoxy)phenyl]-1-methylethyl]phenoxy]ethyl] (2Z)-2-butenedicate, butyl 2-propenoate, 1,3-dihydro-1,3-dioxo-5-isobenzofurancarboxylic acid, ethenylbenzene, .alpha.,.alpha.'-[(1-methylethylidene)di-4,1-phenylene]bis[.omega.-hydroxypoly(oxy-1,2-ethanediyl)] and .alpha.,.alpha.'-[(1-methylethylidene)di-4,1-phenylene]bis[.omega.-hydroxypoly[oxy(methyl-1,2-ethanediyl)]] (9CI) (CA INDEX NAME)

CM 1

CRN 252854-91-4 CMF C42 H48 O10

Double bond geometry as shown.

PAGE 1-B

CM 2

CRN 37353-75-6

CMF (C3 H6 O)n (C3 H6 O)n C15 H16 O2

CCI IDS, PMS

HO
$$(C_3H_6)-O$$
 Me Me Me Me

CRN 32492-61-8

CMF (C2 H4 O)n (C2 H4 O)n C15 H16 O2

CCI PMS

$$HO - CH_2 - CH_2 - O$$
 Me
 Me
 Me
 Me

CM 4

CRN 552-30-7

CMF C9 H4 O5

CM 5

CRN 141-32-2

CMF C7 H12 O2

CM 6

CRN 100-42-5

CMF C8 H8

 $H_2C = CH - Ph$

CM 7

CRN 100-21-0 CMF C8 H6 O4

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L48 ANSWER 11 OF 28 HCAPLUS COPYRIGHT 2003 ACS
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AN 1999:220207 HCAPLUS

DN 130:267894

TI Manufacture of crosslinked polymers as binders for electrophotographic toners

IN Bauer, Stephan; Czech, Erwin; Neus, Michael

PA BASF A.-G., Germany

SO Ger. Offen., 8 pp. CODEN: GWXXBX

DT Patent

LA German

IC ICM C08G063-12

ICS C08G063-127; C08G063-668; C08G069-44; C08G063-78; C08G081-02; C08F212-00; C08L025-04; G03G009-087

ICI C08F212-00, C08F220-18, C08F220-56, C08F222-06, C08F222-38, C08F220-26

CC 35-4 (Chemistry of Synthetic High Polymers)
Section cross-reference(s): 37, 74

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE

----PI DE 19741720 A1 19990325 DE 1997-19741720 19970922
WO 9915577 A1 19990401 WO 1998-EP5538 19980901

W: JP, US

RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE

PRAI DE 1997-19741720 19970922

AB A title polymer is manufd. by (A) prepg. a prepolymer by radical polymn. of (a) styrene or styrene derivs. (structures specified) with (b) (meth)acrylate esters, (meth)acrylonitrile, (N-substituted) (meth)acrylamides, maleic anhydride or maleamide, (c) OH-contg. (meth)acrylates, and (d) bifunctional comonomers as crosslinking agents, and (B) reacting the prepolymer A with a polyol, polyester or polyamide and an aliph. or arom. di- or polycarboxylic acid, ester or anhydride. The resulting polymers are then blended with pigments to give electrophotog. toners. For example, a prepolymer prepd. by radical polymn. of 1,4-butanediol monoacrylate, Bu methacrylate, di-Me terephthalate, divinylbenzene, neopentyl glycol and styrene was combined with neopentyl glycol and di-Me terephthalate, the mixt. was heated with

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Fascat 4100 catalyst and the resulting crosslinked polymer melt-blended
    with Neopen Yellow 075 to give a title toner.
    butanediol monoacrylate crosslinked copolymer manuf binder electrophotog
ST
     toner; methacrylate ester crosslinked copolymer manuf binder electrophotog
     toner; vinylbenzene crosslinker acrylic copolymer manuf binder
     electrophotog toner; neopentyl glycol polycondensation hydroxy functional
     polyacrylate binder electrophotog toner; terephthalate ester
    polycondensation hydroxy functional polyacrylate binder electrophotog
     toner
     Electrophotographic toners
IT
        (manuf. of crosslinked polymers as binders for)
     61901-87-9, c.i. Solvent Black 29
IT
     RL: TEM (Technical or engineered material use); USES (Uses)
        (Neopen Black X 55; manuf. of crosslinked polymers as binders for
        electrophotog. toners contg.)
     222045-98-9P, 1,4-Butanediol monoacrylate-Butyl
IT
     methacrylate-Dimethyl terephthalate-Divinylbenzene-Neopentyl
     glycol-Styrene copolymer 222045-99-0P, 1,4-Butanediol
     monoacrylate-Butyl methacrylate-Divinylbenzene-Dimethyl
     terephthalate-Methyl methacrylate-Neopentyl glycol-Styrene copolymer
     RL: IMF (Industrial manufacture); POF (Polymer in
     formulation); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (crosslinked; manuf. of crosslinked polymers as binders for
        electrophotog. toners)
IT
     104244-10-2, Neopen Yellow 075
                                      221681-45-4, Neopen Magenta SE 1378
     222295-50-3, Neopen Cyan FF 4238
     RL: TEM (Technical or engineered material use); USES (Uses)
        (pigment; manuf. of crosslinked polymers as binders for electrophotog.
        toners contg.)
     222045-98-9P, 1,4-Butanediol monoacrylate-Butyl
TT
     methacrylate-Dimethyl terephthalate-Divinylbenzene-Neopentyl
     glycol-Styrene copolymer 222045-99-0P, 1,4-Butanediol
     monoacrylate-Butyl methacrylate-Divinylbenzene-Dimethyl
     terephthalate-Methyl methacrylate-Neopentyl glycol-Styrene copolymer
     RL: IMF (Industrial manufacture); POF (Polymer in
     formulation); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (crosslinked; manuf. of crosslinked polymers as binders for
        electrophotog. toners)
     222045-98-9 HCAPLUS
RN
     1,4-Benzenedicarboxylic acid, dimethyl ester, polymer with butyl
CN
     2-methyl-2-propenoate, diethenylbenzene, 2,2-dimethyl-1,3-propanediol,
     ethenylbenzene and 4-hydroxybutyl 2-propenoate (9CI) (CA INDEX NAME)
     CM · 1
     CRN 2478-10-6
     CMF C7 H12 O3
              0
HO-(CH_2)_4-O-C-CH=CH_2
```

CRN 1321-74-0 CMF C10 H10 CCI IDS



$$2 \left[D1-CH=CH_2 \right]$$

CM 3

CRN 126-30-7 CMF C5 H12 O2

CM 4

CRN 120-61-6 CMF C10 H10 O4

CM 5

CRN 100-42-5 CMF C8 H8

 $H_2C = CH - Ph$

CM 6

CRN 97-88-1 CMF C8 H14 O2

$$\begin{array}{c|c} & \text{O} & \text{CH}_2 \\ & || & || \\ \text{n-BuO-C-C-Me} \end{array}$$

RN 222045-99-0 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, dimethyl ester, polymer with butyl 2-methyl-2-propenoate, diethenylbenzene, 2,2-dimethyl-1,3-propanediol, ethenylbenzene, 4-hydroxybutyl 2-propenoate and methyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 2478-10-6 CMF C7 H12 O3

CM 2

CRN 1321-74-0 CMF C10 H10 CCI IDS



CM 3

CRN 126-30-7 CMF C5 H12 O2

CM 4

CRN 120-61-6 CMF C10 H10 O4

CM 5

CRN 100-42-5 CMF C8 H8

$$H_2C = CH - Ph$$

CM 6

CRN 97-88-1 CMF C8 H14 O2

$$\begin{array}{c|c} & \text{O} & \text{CH}_2 \\ \parallel & \parallel \\ \text{n-BuO-C-C-Me} \end{array}$$

CM 7

CRN 80-62-6 CMF C5 H8 O2

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L48 ANSWER 12 OF 28 HCAPLUS COPYRIGHT 2003 ACS
    1998:816122 HCAPLUS
AN
DN
     130:53120
     Intercalates and exfoliates formed with n-alkenyl amides and/or
TI
     acrylate-functional pyrrolidone and allylic monomers, oligomers and
     copolymers and composite materials containing same
     Tsipursky, Semeon; Beall, Gary W.; Vinokour, Elena I.
IN
    AMCOL International Corporation, USA
PA
    U.S., 16 pp., Cont.-in-part of U.S. 5,760,121.
SO
     CODEN: USXXAM
DT
     Patent
     English
LΑ
     ICM C08J005-10
IC
     ICS C08K003-34; C08L033-00
     524450000
     37-6 (Plastics Manufacture and Processing)
FAN.CNT 11
                     KIND DATE
                                          APPLICATION NO.
     PATENT NO.
                           _____
                                          -----
     US 5849830
                      Α
                           19981215
                                          US 1997-951094
                                                           19971015
PΙ
                      Α
     US 5552469
                           19960903
                                          US 1995-488264
                                                           19950607
                     Α
                                          US 1995-480080
                                                           19950607
     US 5578672
                           19961126
     US 5698624
                     Α
                          19971216
                                          US 1995-488263
                                                           19950607
                                                           19950908
     US 5721306
                     Α
                          19980224
                                          US 1995-525416
                                                           19960502
     US 5760121
                     A
                          19980602
                                          US 1996-637092
                     AA 19990415
                                          CA 1998-2247607 19980917
     CA 2247607
                                          EP 1998-307748
                                                           19980924
     EP 909787
                     A1 19990421
            AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO
                                          MX 1998-8504
                                                           19981014
     MX 9808504
                     Α
                           20000831
     JP 11193350
                           19990721
                                          JP 1998-293588
                                                           19981015
                      A2
     US 6228903
                           20010508
                                          US 1999-283954
                                                           19990401
                      В1
PRAI US 1995-480080
                      Α2
                           19950607
     US 1995-488263
                      Α2
                           19950607
     US 1995-488264
                      Α2
                           19950607
     US 1995-525416
                      A2
                           19950908
     US 1996-637092
                      A2
                           19960502
                     В1
                           19960802
     US 1996-691689
     US 1997-951094
                     Α
                           19971015
     Nanocomposites are manufd. by combining a host material, such as an org.
AΒ
     solvent or a matrix polymer and exfoliated intercalates formed by
     contacting a phyllosilicate with an intercalant selected from the group
     consisting of (1) an N-alkenyl amide monomer and an allylic monomer; (2)
     an oligomer formed by copolymg. an N-alkenyl amide monomer and an allylic
     monomer; (3) a polymer formed by copolymg. an N-alkenyl amide monomer and
     an allylic monomer; and (4) mixts. thereof to adsorb or the intercalant
     between adjacent phyllosilicate platelets. Sufficient polymer is adsorbed
     between adjacent phyllosilicate platelets to expand the adjacent platelets
     to increase the spacing at least about 10 .ANG., preferably at least about
     20 .ANG. (as measured after water removal), up to about 100 .ANG. and
     preferably in the range of about 30-40 .ANG., so that the intercalate
     easily can be exfoliated, e.g., when mixed with an org. solvent or a
     polymer melt, to provide a carrier material for drugs and the like, or to
     provide a matrix polymer/platelet composite (nanocomposite) material--the
     platelets being exfoliated from the intercalate.
     nanocomposite phyllosilicate intercalate alkenyl amide polymer
```

```
Intercalation compounds
TΥ
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (intercalates and exfoliates formed with n-alkenyl amides and/or
        acrylate-functional pyrrolidone and allylic monomers, oligomers and
        copolymers and composite materials contg. same)
     Polyamides, uses
IT
     RL: POF (Polymer in formulation); USES (Uses)
        (intercalates and exfoliates formed with n-alkenyl amides and/or
        acrylate-functional pyrrolidone and allylic monomers, oligomers and
        copolymers and composite materials contg. same)
IT
     Polyesters, uses
     RL: POF (Polymer in formulation); USES (Uses)
        (intercalates and exfoliates formed with n-alkenyl amides and/or
        acrylate-functional pyrrolidone and allylic monomers, oligomers and
        copolymers and composite materials contg. same)
     Silicates, preparation
IT
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (phyllo-, intercalates with alkenylamide polymers; intercalates and
        exfoliates formed with n-alkenyl amides and/or acrylate-functional
        pyrrolidone and allylic monomers, oligomers and copolymers and
        composite materials contg. same)
     217489-50-4DP, intercalates with phyllosilicates
IT
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (intercalates and exfoliates formed with n-alkenyl amides and/or
        acrylate-functional pyrrolidone and allylic monomers, oligomers and
        copolymers and composite materials contg. same)
ΙT
     24968-12-5 25038-59-9, Polyethylene terephthalate, uses
     26062-94-2, Polybutylene terephthalate
                                               26336-38-9, Polyvinylamine
     RL: POF (Polymer in formulation); USES (Uses)
        (intercalates and exfoliates formed with n-alkenyl amides and/or
        acrylate-functional pyrrolidone and allylic monomers,
        oligomers and copolymers and composite materials contg. same)
              THERE ARE 44 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE.CNT
RE
(1) Alexander; US 4613542 1986 HCAPLUS
(2) Anon; GB 1146668 1969
(3) Anon; DE 1642122 1970 HCAPLUS
(4) Anon; GB 1565362 1980 HCAPLUS
(5) Anon; EP 0205281 A3 1986 HCAPLUS
(6) Anon; EP 0335653 A1 1989 HCAPLUS
(7) Anon; EP 0358415 A1 1990 HCAPLUS
(8) Anon; EP 0479031 A1 1992 HCAPLUS
(9) Anon; EP 0548940 A1 1993 HCAPLUS
(10) Anon; WO 93/04117 1993 HCAPLUS
(11) Anon; WO 93/04118 1993 HCAPLUS
(12) Anon; WO 93/11190 1993 HCAPLUS
(13) Anon; GB 0645181 A2 1995
(14) Anon; EP 0761739 A1 1997 HCAPLUS
(15) Burns; US 3795650 1974 HCAPLUS
(16) Duffield; US 3515626 1970
(17) Francis, C; Soil Science 1973, V115(1), P40 HCAPLUS
(18) Greigger; US 4609717 1986 HCAPLUS
(19) Hamilton; US 2035546 1936 HCAPLUS
(20) Hendrick; US 3419517 1968 HCAPLUS
(21) Herman; US 4210572 1980 HCAPLUS
(22) Hixenbaugh; US 3267058 1966
(23) Kamigaito; US 4472538 1984 HCAPLUS
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(24) Kishida; US 4546145 1985 HCAPLUS

- (25) Klare; US 3423341 1969 HCAPLUS
- (26) Koch; US 3483152 1969 HCAPLUS
- (27) Kojima, Y; J Mater Res 1993, V8(5), P1185 HCAPLUS
- (28) Laughlin; US 3929678 1975 HCAPLUS
- (29) Levy, R; Journal of Colloid and Interface Science 1975, V50(3), P442 HCAPLUS
- (30) Libor; US 4600744 1986 HCAPLUS
- (31) Lyons; US 4125411 1978
- (32) Mardis; US 4434075 1984 HCAPLUS
- (33) McKinley; US 4500670 1985 HCAPLUS
- (34) Mukamal; US 4400485 1983 HCAPLUS
- (35) Osborn; US 4251576 1981 HCAPLUS
- (36) Peppel; US 3268561 1966 HCAPLUS
- (37) Sekmakas; US 3457324 1969 HCAPLUS
- (38) Simone; US 3912532 1975 HCAPLUS
- (39) Suzuki, K; Clays and Clay Minerals 1988, V36(2), P147 HCAPLUS
- (40) Takahashi; US 3773708 1973 HCAPLUS
- (41) Thanawalla; US 4618703 1986 HCAPLUS
- (42) Ure; US 3419460 1968
- (43) Usuki, A; J Mater Res 1993, V8(5), P1179 HCAPLUS
- (44) Weber; US 4431755 1984 HCAPLUS
- 25038-59-9, Polyethylene terephthalate, uses ΙT

RL: POF (Polymer in formulation); USES (Uses)

(intercalates and exfoliates formed with n-alkenyl amides and/or

acrylate-functional pyrrolidone and allylic monomers,

oligomers and copolymers and composite materials contg. same)

- RN 25038-59-9 HCAPLUS
- Poly(oxy-1,2-ethanediyloxycarbonyl-1,4-phenylenecarbonyl) (9CI) CN NAME)

- ANSWER 13 OF 28 HCAPLUS COPYRIGHT 2003 ACS L48
- 1997:513671 HCAPLUS ΑN
- DN 127:191980
- Radiation-curable powder paint binder composition for heat sensitive ΤI substrate
- Jansen, Johan Franz Gradus Antonius; Stanssens, Dirk Armand Wim; De Jong, TN Evert Sjoerd; Udding-Louwrier, Saskia
- DSM N.V., Neth.; Jansen, Johan Franz Gradus Antonius; Stanssens, Dirk PA Armand Wim; De Jong, Evert Sjoerd; Udding-Louwrier, Saskia
- SO PCT Int. Appl., 28 pp. CODEN: PIXXD2
- DTPatent
- LΑ English
- IC ICM C09D005-03
- CC 42-10 (Coatings, Inks, and Related Products)

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FAN.CNT 1
                                          APPLICATION NO. DATE
                     KIND DATE
     PATENT NO.
                            19970731
                                           WO 1997-NL14
                                                            19970115
                     A1
    WO 9727253
PΙ
        W: AL, AU, BA, BB, BG, BR, CA, CN, CU, CZ, EE, GE, HU, IL, IS, JP,
             KP, KR, LC, LK, LR, LT, LV, MG, MK, MN, MX, NO, NZ, PL, RO, SG,
             SI, SK, TR, TT, UA, US, UZ, VN, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
         RW: KE, LS, MW, SD, SZ, UG, AT, BE, CH, DE, DK, ES, FI, FR, GB, GR,
             IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, ML,
             MR, NE, SN, TD, TG
                                           NL 1996-1002153 19960123
                      C2
                            19970804
     NL 1002153
                            19970820
                                           AU 1997-13211
                                                            19970115
    AU 9713211
                      Α1
PRAI NL 1996-1002153
                            19960123
                            19970115
     WO 1997-NL14
     The title paint binder compn. for matte powder coatings on wood or wood
AΒ
     fiber, plastics, etc., comprises a resin and optionally a crosslinker and
     photoinitiator, in which >0.5 mol % (based on the total amt. of
     polymerizable unsatn. of the binder) results from itaconic acid ester
     units. The polymer can be a polyester, a polyacrylate, a polyolefin or an
     addn. product of epoxy resins and itaconic acid. The paint compn. is
     storable at <40.degree. and exhibits good flow at 60-200.degree.. Thus,
     wood fiber panels were coated with a powder compn. contg. a polyester
     (acid value 7.6 mg KOH/g, OH value 61 mg KOH/g, glass transition temp.
     40.degree.) of itaconic acid, neopentyl glycol, terephthalic acid, and
     trimethylolpropane, a crosslinker Uralac ZW 3307P, Irgacure 184, and flow
     aid BYK 361 and cured with a UV lamp (1 J/cm2) to give coated panels with
     acetone resistance and 20/60.degree. gloss 7/1.
     radiation curable itaconic polymer powder paint; wood coating radiation
     curable powder; plastic coating radiation curable powder; matte powder
     paint radiation cured; flatted powder paint radiation cured; polyester
     itaconic powder paint radiation cured
IT
     Alkyd resins
     RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or
     engineered material use); PREP (Preparation); USES (Uses)
        (Uralac P 2115, reaction products with itaconic anhydride;
        radiation-curable powder paint binder compn. for heat sensitive
        substrate)
IT
     Polyesters, uses
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (itaconic group-contg.; radiation-curable powder paint binder compn.
        for heat sensitive substrate)
IT
     Coating materials
        (matte, powder, itaconic modified polyester binder for;
        radiation-curable powder paint binder compn. for heat sensitive
        substrate)
IT
        (radiation-curable powder paint binder compn. for heat sensitive
        substrate)
     2170-03-8DP, Itaconic anhydride, reaction product with hydroxy functional
IT
                 194032-56-9P 194227-68-4P
     RL: IMF (Industrial manufacture); PRP (Properties); TEM
     (Technical or engineered material use); PREP (Preparation); USES
     (Uses)
        (radiation-curable powder paint binder compn. for heat
        sensitive substrate)
IT
     194032-55-8P, Itaconic acid-neopentyl glycol-terephthalic
     acid-trimethylolpropane copolymer
```

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(radiation-curable powder paint **binder** compn. for heat sensitive substrate)

IT 7429-90-5, Aluminum, miscellaneous

RL: MSC (Miscellaneous)

(substrate; radiation-curable powder paint binder compn. for heat sensitive substrate)

IT 194227-68-4P

RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(radiation-curable powder paint **binder** compn. for heat sensitive substrate)

RN 194227-68-4 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, polymer with 2,2-dimethyl-1,3-propanediol, 2-ethyl-2-(hydroxymethyl)-1,3-propanediol, methylenebutanedioic acid and Uralac ZW 3307P (9CI) (CA INDEX NAME)

CM 1

CRN 194165-93-0

CMF Unspecified

CCI MAN

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

CM 2

CRN 126-30-7 CMF C5 H12 O2

CM 3

CRN 100-21-0 CMF C8 H6 O4

CM 4

CRN 97-65-4 CMF C5 H6 O4

$$^{\text{CH}_2}_{||}_{\text{HO}_2\text{C}-\text{C}-\text{CH}_2-\text{CO}_2\text{H}}$$

CRN 77-99-6 CMF C6 H14 O3

$$\begin{array}{c} \text{CH}_2-\text{OH} \\ | \\ \text{HO-CH}_2-\text{C-Et} \\ | \\ \text{CH}_2-\text{OH} \end{array}$$

IT 194032-55-8P, Itaconic acid-neopentyl glycol-terephthalic acid-trimethylolpropane copolymer

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(radiation-curable powder paint binder compn. for heat sensitive substrate)

RN 194032-55-8 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, polymer with 2,2-dimethyl-1,3-propanediol, 2-ethyl-2-(hydroxymethyl)-1,3-propanediol and methylenebutanedioic acid (9CI) (CA INDEX NAME)

CM 1

CRN 126-30-7 CMF C5 H12 O2

CM 2

CRN 100-21-0 CMF C8 H6 O4

CRN 97-65-4 CMF C5 H6 O4

$$_{\rm H0_2C-C-CH_2-CO_2H}^{\rm CH_2}$$

CM 4

CRN 77-99-6 CMF C6 H14 O3

$$\begin{array}{c} \text{CH}_2-\text{OH} \\ | \\ \text{HO-CH}_2-\text{C-Et} \\ | \\ \text{CH}_2-\text{OH} \end{array}$$

L48 ANSWER 14 OF 28 HCAPLUS COPYRIGHT 2003 ACS

AN 1997:388719 HCAPLUS

DN 127:18508

TI Nontoxic colored polyester-based aqueous dispersions with good adhesion to substrate surface

IN Maeda, Goshi; Shimomura, Tetsuo; Hotta, Yasunari; Yamada, Yozo

PA Toyobo Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 14 pp. CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C08L067-00

ICS C09D011-00; C09D011-10; C09D167-00

CC 37-6 (Plastics Manufacture and Processing)
Section cross-reference(s): 42

FAN.CNT 1

The dispersions useful as water-based marking and recording or printing inks, coatings, etc., are prepd. by dispersing fine pigmented or/and dyed particles contg. mainly polyesters in medium contg. mainly water to the substrate consistency S and dispersion viscosity .eta.i (in cPs at 20.degree.) in such a way that {log(.eta.i)-log(.eta.0)}/([S]2).gtoreq.5 (provided that 0.05.ltoreq.S.ltoreq.0.5 and .eta.0 is the viscosity of medium) for reducing feathering and smearing and clogging of delivery tool nozzle. Thus, dissolving 200 parts a polyester derived from di-Me isophthalate, di-Me terephthalate, di-Me 5-sodiosulfoisophthalate, neopentyl glycol and ethylene glycol in MEK 100 and THF 50 mixed solvent contg. T-77 dye 14 parts at 70.degree., adding the soln. to 500 parts

```
water at 70.degree., mixing and distg. until a fraction having b.p.
     103.degree. is reached gave an aq. dispersion having .eta.i 4.2, .eta.0
     1.7, S 21.0%, and good ink performance.
     colored polyester waterborne ink dispersion; marking ink aq polyester
ST
     dispersion; coating waterborne polyester dispersion; smear resistance aq
     polyester ink; recording ink waterborne polyester dispersion; printer ink
     waterborne polyester dispersion; feathering resistance aq polyester
     dispersion
     Polyesters, preparation
IT
     RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or
     engineered material use); PREP (Preparation); USES (Uses)
        (binders; nontoxic colored polyester-based aq. dispersions with good
        adhesion to substrate surface)
IT
     Coating materials
     Inks
        (nontoxic colored polyester-based aq. dispersions with good adhesion to
        substrate surface)
     81977-96-0P, Dimethyl isophthalate-dimethyl 5-sodiosulfoisophthalate-
IT
     dimethyl terephthalate-ethylene glycol-neopentyl glycol copolymer
     184900-08-1P, Dimethyl cyclohexanedicarboxylate-ethylene
     glycol-tricyclodecanedimethanol-trimellitic anhydride copolymer
     190124-76-6P, 4-Cyclohexene-1,2-dicarboxylic acid-dimethyl
     isophthalate-dimethyl 5-sodiosulfoisophthalate-dimethyl
     terephthalate-ethylene glycol-neopentyl glycol copolymer
     190124-86-8P, 4-Cyclohexene-1,2-dicarboxylic acid-dimethyl
     isophthalate-dimethyl 5-sodiosulfoisophthalate-dimethyl
     terephthalate-ethylene glycol-neopentyl glycol copolymer salt with
     dimethylaminoethyl methacrylate
     RL: IMF (Industrial manufacture); PRP (Properties); TEM
     (Technical or engineered material use); PREP (Preparation); USES
     (Uses)
        (binders; nontoxic colored polyester-based aq. dispersions
        with good adhesion to substrate surface)
     156108-08-6, T 77 (Toner)
IT
     RL: PRP (Properties); TEM (Technical or engineered material use); USES
     (Uses)
        (nontoxic colored polyester-based ag. dispersions with good adhesion to
        substrate surface)
     190124-86-8P, 4-Cyclohexene-1,2-dicarboxylic acid-dimethyl
IT
     isophthalate-dimethyl 5-sodiosulfoisophthalate-dimethyl
     terephthalate-ethylene glycol-neopentyl glycol copolymer salt with
     dimethylaminoethyl methacrylate
     RL: IMF (Industrial manufacture); PRP (Properties); TEM
     (Technical or engineered material use); PREP (Preparation); USES
        (binders; nontoxic colored polyester-based aq. dispersions
        with good adhesion to substrate surface)
     190124-86-8 HCAPLUS
RN
     1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt,
CN
     polymer with 4-cyclohexene-1,2-dicarboxylic acid, dimethyl
     1,3-benzenedicarboxylate, dimethyl 1,4-benzenedicarboxylate,
     2,2-dimethyl-1,3-propanediol and 1,2-ethanediol, compd. with
     2-(dimethylamino)ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)
     CM
          1
     CRN 2867-47-2
     CMF C8 H15 N O2
```

CRN 120-61-6 CMF C10 H10 O4

CM 7

CRN 107-21-1 CMF C2 H6 O2

но-сн2-сн2-он

CM 8

CRN 88-98-2 CMF C8 H10 O4

L48 ANSWER 15 OF 28 HCAPLUS COPYRIGHT 2003 ACS

AN 1997:377429 HCAPLUS

DN 127:18587

TI Polyester compositions for hot water-resistant water-thinned coatings

IN Nakatani, Ken; Riku, Shuichiro

PA Takamatsu Yushi K. K., Japan

SO Jpn. Kokai Tokkyo Koho, 6 pp.

CODEN: JKXXAF

DT Patent

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LΑ
     Japanese
IC
     ICM C09D163-00
     ICS B05D007-24; C08L051-08; C08L063-00; C09D011-10; C09D151-08;
         D06M015-55; D21H019-24
     37-6 (Plastics Manufacture and Processing)
CC
     Section cross-reference(s): 38, 40, 42
FAN.CNT 1
                                         APPLICATION NO. DATE
                    KIND DATE
     PATENT NO.
     ______
                                          _____
PI JP 09100440 A2 19970415
PRAI JP 1995-291598 19951002
                                          JP 1995-291598 19951002
     The compns., useful for nonwoven textile binders, coatings, antifogging
AΒ
     agents, etc., contain (A) modified polyesters obtained by polymg.
     radically polymerizable vinyl monomers contg. glycidyl groups in aq.
     solns. or dispersions of polyesters contg. carboxylic acid salts and with
     mol. wt. 2000-30,000 and (B) crosslinking agents contg. .gtoreq.2 epoxy
     groups at ratio Y/X = 0.2-5 (X = equiv. of CO2H in polyester; Y = equiv.
     of epoxy groups in crosslinking agent). Thus, 80 parts 15%-solid soln. of
     1163:217:365:94:362 isophthalic acid-ethylene glycol-neopentyl
     glycol-trimethylolpropane-trimellitic acid copolymer was polymd. with 5
     parts styrene and 3 parts glycidyl methacrylate at 70-80.degree. to give a
     23%-solid soln., 100 parts of which was blended with 3 parts Denacol EX
     313. A polyester nonwoven fabric was soaked in the blend, squeezed, and
     set at 150.degree. to give an abrasive cloth for ceramics.
     polyester glycidyl modification water resistant coating; glycidyl
ST
     methacrylate polyester nonwoven fabric binder; epoxy crosslinking agent
     glycidyl polyester coating; hardener epoxy glycidyl polyester coating
ΙT
     Antifogging agents
        (aq. polyester compns. contq. epoxies for hot water-resistant coatings,
        nonwoven fabric binders, and film antifogging agents)
IT
     Nonwoven fabrics
     RL: MSC (Miscellaneous)
        (aq. polyester compns. contg. epoxies for hot water-resistant coatings,
        nonwoven fabric binders, and film antifogging agents)
     Polyesters, preparation
IT
     Polyesters, preparation
     RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP
     (Properties); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (epoxy; aq. polyester compns. contg. epoxies for hot water-resistant
        coatings, nonwoven fabric binders, and film antifogging agents)
ΙT
     Coating materials
        (heat-resistant; aq. polyester compns. contg. epoxies for hot
        water-resistant coatings, nonwoven fabric binders, and film antifogging
        agents)
ΤΤ
     Epoxy resins, preparation
     Epoxy resins, preparation
     RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP
     (Properties); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (polyester-; aq. polyester compns. contg. epoxies for hot
        water-resistant coatings, nonwoven fabric binders, and film antifogging
        agents)
ΙT
     Coating materials
        (transparent; aq. polyester compns. contg. epoxies for hot
        water-resistant coatings, nonwoven fabric binders, and film antifogging
        agents)
IT
     Coating materials
```

Coating materials

(water-resistant, water-thinned; aq. polyester compns. contg. epoxies for hot water-resistant coatings, nonwoven fabric binders, and film antifogging agents)

IT 189940-10-1P 189940-11-2P 189940-13-4P 189940-14-5P

RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(aq. polyester compns. contg. epoxies for hot water-resistant coatings, nonwoven fabric binders, and film antifogging agents)

IT 189940-10-1P 189940-11-2P 189940-13-4P 189940-14-5P

RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(aq. polyester compns. contg. epoxies for hot water-resistant coatings, nonwoven fabric binders, and film antifogging agents)

RN 189940-10-1 HCAPLUS

CN 1,2,4-Benzenetricarboxylic acid, polymer with 1,3-benzenedicarboxylic acid, 1,3(or 2,3)-bis(oxiranylmethoxy)propanol, 2,2-dimethyl-1,3-propanediol, 1,2-ethanediol, ethenylbenzene, 2-ethyl-2-(hydroxymethyl)-1,3-propanediol and oxiranylmethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 528-44-9 CMF C9 H6 O6

CM 2

CRN 126-30-7 CMF C5 H12 O2

CM 3

CRN 121-91-5 CMF C8 H6 O4

CM 4

CRN 107-21-1 CMF C2 H6 O2

 ${\hbox{HO-CH}_2\hbox{--}\hbox{CH}_2\hbox{--}\hbox{OH}}$

CM 5

CRN 106-91-2 CMF C7 H10 O3

CM 6

CRN 100-42-5 CMF C8 H8

 $H_2C = CH - Ph$

CM 7

CRN 77-99-6 CMF C6 H14 O3

$$\begin{array}{c} \text{CH}_2-\text{OH} \\ | \\ \text{HO-CH}_2-\text{C-Et} \\ | \\ \text{CH}_2-\text{OH} \end{array}$$

CM 8

CRN 27043-36-3 CMF C9 H16 O5 CCI IDS

CM 9

CRN 556-52-5 CMF C3 H6 O2

CM 10

CRN 56-81-5 CMF C3 H8 O3

$$_{\rm HO-\,CH_2-\,CH-\,CH_2-\,OH}^{\rm OH}$$

RN 189940-11-2 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, dimethyl ester, polymer with 1,3(or 2,3)-bis(oxiranylmethoxy)propanol, 1,3-dihydro-1,3-dioxo-5-isobenzofurancarboxylic acid, dimethyl 1,4-benzenedicarboxylate, 2,2-dimethyl-1,3-propanediol, 1,2-ethanediol, methyl 2-methyl-2-propenoate and oxiranylmethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 1459-93-4 CMF C10 H10 O4

CM 2

CRN 552-30-7 CMF C9 H4 O5

CRN 126-30-7 CMF C5 H12 O2

$$\begin{array}{c} \text{Me} \\ \mid \\ \text{HO-CH}_2\text{-C-CH}_2\text{-OH} \\ \mid \\ \text{Me} \end{array}$$

CM 4

CRN 120-61-6 CMF C10 H10 O4

CM 5

CRN 107-21-1 CMF C2 H6 O2

 $HO-CH_2-CH_2-OH$

CM 6

CRN 106-91-2 CMF C7 H10 O3

CRN 80-62-6 CMF C5 H8 O2

CM 8

CRN 27043-36-3 CMF C9 H16 O5 CCI IDS

CM 9

CRN 556-52-5 CMF C3 H6 O2

CM 10

CRN 56-81-5 CMF C3 H8 O3

RN 189940-13-4 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, dimethyl ester, polymer with Denacol EX 1101, 1,3-dihydro-1,3-dioxo-5-isobenzofurancarboxylic acid, dimethyl 1,4-benzenedicarboxylic acid, 2,2-dimethyl-1,3-propanediol, 1,2-ethanediol, 2-hydroxyethyl 2-methyl-2-propenoate and oxiranylmethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 151354-28-8

CMF Unspecified CCI PMS, MAN

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

CM 2

CRN 1459-93-4 CMF C10 H10 O4

CM 3

CRN 868-77-9 CMF C6 H10 O3

CM 4

CRN 552-30-7 CMF C9 H4 O5

CM 5

CRN 126-30-7 CMF C5 H12 O2

CRN 120-61-6 CMF C10 H10 O4

CM '

CRN 107-21-1 CMF C2 H6 O2

 $_{\text{HO--}}$ $_{\text{CH}_2}-_{\text{CH}_2}-_{\text{OH}}$

CM 8

CRN 106-91-2 CMF C7 H10 O3

RN 189940-14-5 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, polymer with 1,4-benzenedicarboxylic acid, 1H,3H-benzo[1,2-c:4,5-c']difuran-1,3,5,7-tetrone, 1,3(or 2,3)-bis(oxiranylmethoxy)propanol, butyl 2-methyl-2-propenoate, 1,2-ethanediol, oxiranylmethyl 2-methyl-2-propenoate and 2,2'-oxybis[ethanol] (9CI) (CA INDEX NAME)

CM 1

CRN 121-91-5 CMF C8 H6 O4

CRN 111-46-6 CMF C4 H10 O3

 ${\tt HO-CH_2-CH_2-O-CH_2-CH_2-OH}$

CM 3

CRN 107-21-1 CMF C2 H6 O2

 $HO-CH_2-CH_2-OH$

CM 4

CRN 106-91-2 CMF C7 H10 O3

$$\overset{\text{O}}{\longleftarrow} \overset{\text{O}}{\underset{\text{CH}_2-\text{O-C-C-Me}}{\parallel}} \overset{\text{CH}_2}{\parallel}$$

CM 5

CRN 100-21-0 CMF C8 H6 O4

CM 6

CRN 97-88-1 CMF C8 H14 O2

$$\begin{array}{c|c} & \text{O} & \text{CH}_2 \\ & || & || \\ \text{n-BuO-C-C-Me} \end{array}$$

CM 7

CRN 89-32-7 CMF C10 H2 O6

CM 8

CRN 27043-36-3 CMF C9 H16 O5 CCI IDS

CM 9

CRN 556-52-5 CMF C3 H6 O2

CM 10

CRN 56-81-5 CMF C3 H8 O3

$$\begin{array}{c} \text{OH} \\ | \\ \text{HO-CH}_2\text{--CH-CH}_2\text{--OH} \end{array}$$

L48 ANSWER 16 OF 28 HCAPLUS COPYRIGHT 2003 ACS

AN 1997:361181 HCAPLUS

DN 127:12436

TI Magnetic recording material containing quaternary ammonium-substituted polyester-polyurethane binder

KATHLEEN FULLER EIC 1700/PARKER LAW 308-4290

IN Tsunoda, Shohei; Sasahara, Toshiaki; Konishi, Shin

PA Nippon Polyurethane Industry Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 13 pp. CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM G11B005-702

ICS C09D005-23; C09D175-06

CC 77-8 (Magnetic Phenomena)

Section cross-reference(s): 38

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
		:		
PI JP 09073625	A2	19970318	JP 1995-251827	19950905
PRAI JP 1995-2518	327	19950905		
GI			•	

$$-N = A^{1}$$

$$-N + N = A^2$$

$$-N \stackrel{R^1}{>} N - = A^3$$

The recording material has a magnetic layer contg. a polyurethane having a structure OCOCR'HCH2ACH2CR'HCO2 [I; A = N+RR3.X-, A1, A2; R = (substituted) alkyl, aryl, aralkyl; R' = H, alkyl; R1, R2 = alkylene; R3 = alkyl; X- = anion], which is preferably obtained by treating an org. diisocyanate with a long-chain polyols contg. I (A = NR, A3) and an optional chain-extending agent, followed by reaction with an alkylating agent to quaternize the tertiary amino group to quaternary one. Magnetic powders in the magnetic layer showed good dispersibility and dispersion stability in the polyurethane binder.

ST magnetic recording quaternary ammonium polyurethane binder; polyester polyurethane magnetic recording binder dispersibility

IT Magnetic recording materials

(magnetic recording material contg. quaternary ammonium-substituted polyester-polyurethane binder)

IT Polyurethanes, uses

Polyurethanes, uses

RL: DEV (Device component use); IMF (Industrial manufacture); PREP (Preparation); USES (Uses)

```
(polycarbonate-; magnetic recording material contg. quaternary
        ammonium-substituted polyester-polyurethane binder)
     Polyurethanes, uses
IT
     RL: DEV (Device component use); IMF (Industrial manufacture); PREP
     (Preparation); USES (Uses)
        (polyester-; magnetic recording material contg. quaternary
        ammonium-substituted polyester-polyurethane binder)
IT
     Polycarbonates, uses
     Polycarbonates, uses
     RL: DEV (Device component use); IMF (Industrial manufacture); PREP
     (Preparation); USES (Uses)
        (polyurethane-; magnetic recording material contg. quaternary
        ammonium-substituted polyester-polyurethane binder)
     189081-17-2DP, 4,4'-Diphenylmethane diisocyanate-ethylene
IT
     glycol-2-hydroxyethyl acrylate-isophthalic acid-neopentyl
     glycol-piperazine-terephthalic acid copolymer, quaternized
     189081-18-3DP, Adipic acid-1,4-butanediol-4,4'-diphenylmethane
     diisocyanate-2-hydroxyethyl acrylate-piperazine copolymer, quaternized
     189081-19-4DP, .epsilon.-Caprolactone-4,4'-diphenylmethane
     diisocyanate-ethylene glycol-2-hydroxyethyl acrylate-isophthalic
     acid-neopentyl glycol-piperazine-terephthalic acid copolymer, quaternized
     189081-20-7DP, 2-Butyl-2-ethyl-1,3-propanediol-.epsilon.-
     caprolactone-4,4'-diphenylmethane diisocyanate-2-hydroxyethyl
     acrylate-isophthalic acid-neopentyl glycol-piperazine-terephthalic acid
     copolymer, quaternized 189081-21-8DP, .epsilon.-Caprolactone-
     4,4'-diphenylmethane diisocyanate-ethylene glycol-2-hydroxyethyl
     acrylate-isophthalic acid-neopentyl glycol-piperazine-terephthalic
     acid-trimethylolpropane copolymer, quaternized 189081-23-0DP,
     N, N-Dimethyl-1, 3-propanediamine-4, 4'-diphenylmethane diisocyanate-ethylene
     glycol-2-hydroxyethyl acrylate-isophthalic acid-neopentyl
     glycol-terephthalic acid copolymer, quaternized 189081-27-4DP,
     .epsilon.-Caprolactone-N, N-dimethyl-1, 3-propanediamine-4, 4'-
     diphenylmethane diisocyanate-ethylene glycol-2-hydroxyethyl
     acrylate-isophthalic acid-neopentyl glycol-terephthalic acid copolymer,
     quaternized
                  189081-31-0DP, .epsilon.-Caprolactone-diethyl
     carbonate-N, N-dimethyl-1, 3-propanediamine-4, 4'-diphenylmethane
     diisocyanate-1,6-hexanediol-2-hydroxyethyl acrylate copolymer, quaternized
     189081-33-2DP, .epsilon.-Caprolactone-1,4-cyclohexanedimethanol-diethyl
     carbonate-N, N-dimethyl-1, 3-propanediamine-4, 4'-diphenylmethane
     diisocyanate-2-hydroxyethyl acrylate copolymer, quaternized
     189081-35-4DP, Adipic acid-1,4-butanediol-N,N-dimethyl-1,3-
     propanediamine-4,4'-diphenylmethane diisocyanate-ethylene
     glycol-2-hydroxyethyl acrylate-isophthalic acid-neopentyl
     glycol-terephthalic acid copolymer, quaternized 189638-33-3DP,
     1,4-Butanediol-2-butyl-2-ethyl-1,3-propanediol-.epsilon.-caprolactone-4,4'-
     diphenylmethane diisocyanate-2-hydroxyethyl acrylate-isophthalic
     acid-neopentyl glycol-piperazine-terephthalic acid copolymer, quaternized
     189638-34-4DP, 4,4'-Diphenylmethane diisocyanate-ethylene
     glycol-2-hydroxyethyl acrylate-isophthalic acid-neopentyl
     qlycol-piperazine-terephthalic acid-trimethylolpropane copolymer,
     quaternized 189638-35-5DP, 1-Butyl-2-ethyl-1,3-propanediol-N,N-
     dimethyl-1,3-propanediamine-4,4'-diphenylmethane diisocyanate-2-
     hydroxyethyl acrylate-isophthalic acid-neopentyl glycol-terephthalic acid
     copolymer, quaternized 189638-36-6DP, 1-Butyl-2-ethyl-1,3-
     propanediol-.epsilon.-caprolactone-N, N-dimethyl-1, 3-propanediamine-4, 4'-
     diphenylmethane diisocyanate-2-hydroxyethyl acrylate-isophthalic
     acid-neopentyl glycol-terephthalic acid copolymer, quaternized
     189638-37-7DP, Adipic acid-1,4-butanediol-1-butyl-2-ethyl-1,3-
```

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propanediol-N, N-dimethyl-1, 3-propanediamine-4, 4'-diphenylmethane
     diisocyanate-2-hydroxyethyl acrylate-isophthalic acid-neopentyl
    glycol-terephthalic acid copolymer, quaternized
    RL: DEV (Device component use); IMF (Industrial manufacture);
     PREP (Preparation); USES (Uses)
        (magnetic recording material contg. quaternary ammonium-substituted
        polyester-polyurethane binder)
     189081-17-2Dp, 4,4'-Diphenylmethane diisocyanate-ethylene
IT
    glycol-2-hydroxyethyl acrylate-isophthalic acid-neopentyl
    glycol-piperazine-terephthalic acid copolymer, quaternized
     189081-19-4DP, .epsilon.-Caprolactone-4,4'-diphenylmethane
    diisocyanate-ethylene glycol-2-hydroxyethyl acrylate-isophthalic
     acid-neopentyl glycol-piperazine-terephthalic acid copolymer, quaternized
     189081-20-7DP, 2-Butyl-2-ethyl-1,3-propanediol-.epsilon.-
     caprolactone-4,4'-diphenylmethane diisocyanate-2-hydroxyethyl
     acrylate-isophthalic acid-neopentyl glycol-piperazine-terephthalic acid
     copolymer, quaternized 189081-21-8DP, .epsilon.-Caprolactone-
     4,4'-diphenylmethane diisocyanate-ethylene glycol-2-hydroxyethyl
    acrylate-isophthalic acid-neopentyl glycol-piperazine-terephthalic
     acid-trimethylolpropane copolymer, quaternized 189081-23-0DP,
    N, N-Dimethyl-1, 3-propanediamine-4, 4'-diphenylmethane diisocyanate-ethylene
     glycol-2-hydroxyethyl acrylate-isophthalic acid-neopentyl
     glycol-terephthalic acid copolymer, quaternized 189081-27-4DP,
     .epsilon.-Caprolactone-N, N-dimethyl-1, 3-propanediamine-4, 4'-
     diphenylmethane diisocyanate-ethylene glycol-2-hydroxyethyl
     acrylate-isophthalic acid-neopentyl glycol-terephthalic acid copolymer,
     quaternized 189081-35-4DP, Adipic acid-1,4-butanediol-N,N-
     dimethyl-1,3-propanediamine-4,4'-diphenylmethane diisocyanate-ethylene
     qlycol-2-hydroxyethyl acrylate-isophthalic acid-neopentyl
     glycol-terephthalic acid copolymer, quaternized 189638-33-3DP,
     1,4-Butanediol-2-butyl-2-ethyl-1,3-propanediol-.epsilon.-caprolactone-4,4'-
     diphenylmethane diisocyanate-2-hydroxyethyl acrylate-isophthalic
     acid-neopentyl glycol-piperazine-terephthalic acid copolymer, quaternized
     189638-34-4DP, 4,4'-Diphenylmethane diisocyanate-ethylene
     glycol-2-hydroxyethyl acrylate-isophthalic acid-neopentyl
     glycol-piperazine-terephthalic acid-trimethylolpropane copolymer,
     quaternized 189638-35-5DP, 1-Butyl-2-ethyl-1,3-propanediol-N,N-
     dimethyl-1,3-propanediamine-4,4'-diphenylmethane diisocyanate-2-
     hydroxyethyl acrylate-isophthalic acid-neopentyl glycol-terephthalic acid
     copolymer, quaternized 189638-36-6DP, 1-Butyl-2-ethyl-1,3-
     propanediol-.epsilon.-caprolactone-N, N-dimethyl-1, 3-propanediamine-4, 4'-
     diphenylmethane diisocyanate-2-hydroxyethyl acrylate-isophthalic
     acid-neopentyl glycol-terephthalic acid copolymer, quaternized
     189638-37-7DP, Adipic acid-1,4-butanediol-1-butyl-2-ethyl-1,3-
     propanediol-N, N-dimethyl-1, 3-propanediamine-4, 4'-diphenylmethane
     diisocyanate-2-hydroxyethyl acrylate-isophthalic acid-neopentyl
     glycol-terephthalic acid copolymer, quaternized
     RL: DEV (Device component use); IMF (Industrial manufacture);
     PREP (Preparation); USES (Uses)
        (magnetic recording material contg. quaternary ammonium-substituted
        polyester-polyurethane binder)
RN
     189081-17-2 HCAPLUS
CN
     1,3-Benzenedicarboxylic acid, polymer with 1,4-benzenedicarboxylic acid,
     2,2-dimethyl-1,3-propanediol, 1,2-ethanediol, 2-hydroxyethyl 2-propenoate,
     1,1'-methylenebis[4-isocyanatobenzene] and piperazine (9CI) (CA INDEX
     NAME)
```

CRN 818-61-1 CMF C5 H8 O3

$$\begin{array}{c} \text{O} \\ || \\ \text{HO-CH}_2\text{-CH}_2\text{-O-C-CH------} \text{CH}_2 \end{array}$$

CM 2

CRN 126-30-7 CMF C5 H12 O2

$$\begin{array}{c} \text{Me} \\ \mid \\ \text{HO-CH}_2\text{--C-CH}_2\text{--OH} \\ \mid \\ \text{Me} \end{array}$$

CM 3

CRN 121-91-5 CMF C8 H6 O4

CM 4

CRN 110-85-0 CMF C4 H10 N2

CM 5

CRN 107-21-1 CMF C2 H6 O2 $_{\text{HO}-\,\text{CH}_2-\,\text{CH}_2-\,\text{OH}}$

CM 6

CRN 101-68-8 CMF C15 H10 N2 O2

CM 7

CRN 100-21-0 CMF C8 H6 O4

RN 189081-19-4 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, polymer with 1,4-benzenedicarboxylic acid, 2,2-dimethyl-1,3-propanediol, 1,2-ethanediol, 2-hydroxyethyl 2-propenoate, 1,1'-methylenebis[4-isocyanatobenzene], 2-oxepanone and piperazine (9CI) (CA INDEX NAME)

CM 1

CRN 818-61-1 CMF C5 H8 O3

$$\begin{array}{c} {\rm O} \\ || \\ {\rm HO-CH_2-CH_2-O-C-CH} \end{array}$$

CM 2

CRN 502-44-3 CMF C6 H10 O2

CRN 126-30-7 CMF C5 H12 O2

$$\begin{array}{c} \text{Me} \\ \mid \\ \text{HO-CH}_2\text{--C-CH}_2\text{--OH} \\ \mid \\ \text{Me} \end{array}$$

CM 4

CRN 121-91-5 CMF C8 H6 O4

CM 5

CRN 110-85-0 CMF C4 H10 N2

CM 6

CRN 107-21-1 CMF C2 H6 O2

но-сн2-сн2-он

CRN 101-68-8

CMF C15 H10 N2 O2

CM 8

CRN 100-21-0 CMF C8 H6 O4

RN 189081-20-7 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, polymer with 1,4-benzenedicarboxylic acid, 2-butyl-2-ethyl-1,3-propanediol, 2,2-dimethyl-1,3-propanediol, 2-hydroxyethyl 2-propenoate, 1,1'-methylenebis[4-isocyanatobenzene], 2-oxepanone and piperazine (9CI) (CA INDEX NAME)

CM 1

CRN 818-61-1 CMF C5 H8 O3

$$\begin{array}{c} {\rm O} \\ || \\ {\rm HO-CH_2-CH_2-O-C-CH} \end{array} \\ {\rm CH_2} \\$$

CM 2

CRN 502-44-3 CMF C6 H10 O2

CM 3

CRN 126-30-7 CMF C5 H12 O2

$$\begin{array}{c} \text{Me} \\ \mid \\ \text{HO-CH}_2\text{--C-CH}_2\text{--OH} \\ \mid \\ \text{Me} \end{array}$$

CM 4

CRN 121-91-5 CMF C8 H6 O4

CM 5

CRN 115-84-4 CMF C9 H20 O2

$$\begin{array}{c} \text{CH}_2-\text{OH} \\ \mid \\ \text{Et-C-Bu-n} \\ \mid \\ \text{CH}_2-\text{OH} \end{array}$$

CM 6

CRN 110-85-0 CMF C4 H10 N2

CM 7

CRN 101-68-8 CMF C15 H10 N2 O2

CM 8

CRN 100-21-0 CMF C8 H6 O4

RN 189081-21-8 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, polymer with 1,4-benzenedicarboxylic acid, 2,2-dimethyl-1,3-propanediol, 1,2-ethanediol, 2-ethyl-2-(hydroxymethyl)-1,3-propanediol, 2-hydroxyethyl 2-propenoate, 1,1'-methylenebis[4-isocyanatobenzene], 2-oxepanone and piperazine (9CI) (CA INDEX NAME)

CM 1

CRN 818-61-1 CMF C5 H8 O3

$$0 \\ | | \\ HO-CH_2-CH_2-O-C-CH = CH_2$$

CM 2

CRN 502-44-3 CMF C6 H10 O2

CM 3

CRN 126-30-7

CMF C5 H12 O2

$$\begin{array}{c} \text{Me} \\ \mid \\ \text{HO-CH}_2\text{--C-CH}_2\text{--OH} \\ \mid \\ \text{Me} \end{array}$$

CM 4

CRN 121-91-5 CMF C8 H6 O4

CM 5

CRN 110-85-0 CMF C4 H10 N2



CM 6

CRN 107-21-1 CMF C2 H6 O2

 ${\hbox{HO-CH}_2\hbox{--}\hbox{CH}_2\hbox{--}\hbox{OH}}$

CM 7

CRN 101-68-8 CMF C15 H10 N2 O2

CRN 100-21-0 CMF C8 H6 O4

CM S

CRN 77-99-6 CMF C6 H14 O3

$$\begin{array}{c} \text{CH}_2-\text{OH} \\ | \\ \text{HO-CH}_2-\text{C-Et} \\ | \\ \text{CH}_2-\text{OH} \end{array}$$

RN 189081-23-0 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, polymer with 1,4-benzenedicarboxylic acid, N,N-dimethyl-1,3-propanediamine, 2,2-dimethyl-1,3-propanediol, 1,2-ethanediol, 2-hydroxyethyl 2-propenoate and 1,1'-methylenebis[4-isocyanatobenzene] (9CI) (CA INDEX NAME)

CM 1

CRN 818-61-1 CMF C5 H8 O3

$$\begin{array}{c} {\rm O} \\ || \\ {\rm HO-CH_2-CH_2-O-C-CH} \end{array}$$

CM 2

CRN 126-30-7 CMF C5 H12 O2

CM 3

CRN 121-91-5 CMF C8 H6 O4

CM 4

CRN 109-55-7 CMF C5 H14 N2

 $H_2N-(CH_2)_3-NMe_2$

CM 5

CRN 107-21-1 CMF C2 H6 O2

 ${\tt HO-CH_2-CH_2-OH}$

CM 6

CRN 101-68-8

CMF C15 H10 N2 O2

CM T

CRN 100-21-0 CMF C8 H6 O4

KATHLEEN FULLER EIC 1700/PARKER LAW 308-4290

RN 189081-27-4 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, polymer with 1,4-benzenedicarboxylic acid, N,N-dimethyl-1,3-propanediamine, 2,2-dimethyl-1,3-propanediol, 1,2-ethanediol, 2-hydroxyethyl 2-propenoate, 1,1'-methylenebis[4-isocyanatobenzene] and 2-oxepanone (9CI) (CA INDEX NAME)

CM 1

CRN 818-61-1 CMF C5 H8 O3

$$\begin{array}{c} \text{O} \\ || \\ \text{HO-CH}_2\text{-CH}_2\text{-O-C-CH-----} \text{CH}_2 \end{array}$$

CM 2

CRN 502-44-3 CMF C6 H10 O2

CM 3

CRN 126-30-7 CMF C5 H12 O2

CM 4

CRN 121-91-5 CMF C8 H6 O4

CM 5

CRN 109-55-7 CMF C5 H14 N2

 $H_2N-(CH_2)_3-NMe_2$

CM 6

CRN 107-21-1 CMF C2 H6 O2

 ${\rm HO-CH_2-CH_2-OH}$

CM 7

CRN 101-68-8 CMF C15 H10 N2 O2

CM 8

CRN 100-21-0 CMF C8 H6 O4

RN 189081-35-4 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, polymer with 1,4-benzenedicarboxylic acid, 1,4-butanediol, N,N-dimethyl-1,3-propanediamine, 2,2-dimethyl-1,3-propanediol, 1,2-ethanediol, hexanedioic acid, 2-hydroxyethyl 2-propenoate

and 1,1'-methylenebis[4-isocyanatobenzene] (9CI) (CA INDEX NAME)

CM 1

CRN 818-61-1 CMF C5 H8 O3

$$\begin{array}{c} & \text{O} \\ || \\ \text{HO-CH}_2\text{--CH}_2\text{--O-C-CH} \end{array}$$

CM 2

CRN 126-30-7 CMF C5 H12 O2

CM 3

CRN 124-04-9 CMF C6 H10 O4

$$HO_2C-(CH_2)_4-CO_2H$$

CM 4

CRN 121-91-5 CMF C8 H6 O4

CM 5

CRN 110-63-4 CMF C4 H10 O2 $^{\rm HO-}$ (CH₂)₄ $^{\rm -}$ OH

CM 6

CRN 109-55-7 CMF C5 H14 N2

 $H_2N-(CH_2)_3-NMe_2$

CM 7

CRN 107-21-1 CMF C2 H6 O2

 ${\tt HO-CH_2-CH_2-OH}$

CM 8

CRN 101-68-8 CMF C15 H10 N2 O2

CM 9

CRN 100-21-0 CMF C8 H6 O4

RN 189638-33-3 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, polymer with 1,4-benzenedicarboxylic acid, 1,4-butanediol, 2-butyl-2-ethyl-1,3-propanediol, 2,2-dimethyl-1,3-propanediol, 2-hydroxyethyl 2-propenoate, 1,1'-methylenebis[4-isocyanatobenzene], 2-oxepanone and piperazine (9CI) (CA INDEX NAME)

CM 1

CRN 818-61-1 CMF C5 H8 O3

$$\begin{array}{c} \text{O} \\ || \\ \text{HO-CH}_2\text{--CH}_2\text{--O-C-CH} \end{array}$$

CM 2

CRN 502-44-3 CMF C6 H10 O2

CM 3

CRN 126-30-7 CMF C5 H12 O2

$$\begin{array}{c} \text{Me} \\ \mid \\ \text{HO-CH}_2\text{--C-CH}_2\text{--OH} \\ \mid \\ \text{Me} \end{array}$$

CM 4

CRN 121-91-5 CMF C8 H6 O4

CM 5

CRN 115-84-4 CMF C9 H20 O2

CRN 110-85-0 CMF C4 H10 N2



CM 7

ČRN 110-63-4 CMF C4 H10 O2

 $HO-(CH_2)_4-OH$

CM 8

CRN 101-68-8 CMF C15 H10 N2 O2

CM 9

CRN 100-21-0 CMF C8 H6 O4

189638-34-4 HCAPLUS RN

1,3-Benzenedicarboxylic acid, polymer with 1,4-benzenedicarboxylic acid, 2,2-dimethyl-1,3-propanediol, 1,2-ethanediol, 2-ethyl-2-(hydroxymethyl)-1,3-propanediol, 2-hydroxyethyl 2-propenoate, 1,1'-methylenebis[4-isocyanatobenzene] and piperazine (9CI) (CA INDEX NAME) CN

CM

CRN 818-61-1 CMF C5 H8 O3

$$\begin{array}{c} \text{O} \\ || \\ \text{HO-CH}_2\text{-CH}_2\text{-O-C-CH} \end{array}$$

CM

CRN 126-30-7 CMF C5 H12 O2

CM 3

CRN 121-91-5 CMF C8 H6 O4

CM

CRN 110-85-0 CMF C4 H10 N2

CRN 107-21-1 CMF C2 H6 O2

 $_{\text{HO--}}$ С $_{\text{H2--}}$ ОН

CM 6

CRN 101-68-8

CMF C15 H10 N2 O2

CM 7

CRN 100-21-0 CMF C8 H6 O4

CM 8

CRN 77-99-6 CMF C6 H14 O3

RN 189638-35-5 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, polymer with 1,4-benzenedicarboxylic acid, N,N-dimethyl-1,3-propanediamine, 2,2-dimethyl-1,3-propanediol, 2-ethyl-1,3-heptanediol, 2-hydroxyethyl 2-propenoate and 1,1'-methylenebis[4-isocyanatobenzene] (9CI) (CA INDEX NAME)

CM 1

CRN 39775-59-2 CMF C9 H20 O2

CM 2

CRN 818-61-1 CMF C5 H8 O3

$$\begin{array}{c|c} & \circ & \\ | & | \\ \text{HO-CH}_2\text{--CH}_2\text{--O-C-CH} & \text{CH}_2 \end{array}$$

CM 3

CRN 126-30-7 CMF C5 H12 O2

$$\begin{array}{c} \text{Me} \\ \mid \\ \text{HO-CH}_2\text{--C-CH}_2\text{--OH} \\ \mid \\ \text{Me} \end{array}$$

CM 4

CRN 121-91-5 CMF C8 H6 O4

CM 5

CRN 109-55-7 CMF C5 H14 N2 $H_2N-(CH_2)_3-NMe_2$

CM 6

CRN 101-68-8

CMF C15 H10 N2 O2

CM 7

CRN 100-21-0

CMF C8 H6 O4

RN 189638-36-6 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, polymer with 1,4-benzenedicarboxylic acid, N,N-dimethyl-1,3-propanediamine, 2,2-dimethyl-1,3-propanediol, 2-ethyl-1,3-heptanediol, 2-hydroxyethyl 2-propenoate, 1,1'-methylenebis[4-isocyanatobenzene] and 2-oxepanone (9CI) (CA INDEX NAME)

CM 1

CRN 39775-59-2 CMF C9 H20 O2

CM 2

CRN 818-61-1 CMF C5 H8 O3

CRN 502-44-3 CMF C6 H10 O2

CM 4

CRN 126-30-7 CMF C5 H12 O2

CM 5

CRN 121-91-5 CMF C8 H6 O4

CM 6

CRN 109-55-7 CMF C5 H14 N2

 $H_2N-(CH_2)_3-NMe_2$

CM 7

CRN 101-68-8 CMF C15 H10 N2 O2

CM 8

CRN 100-21-0 CMF C8 H6 O4

RN 189638-37-7 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, polymer with 1,4-benzenedicarboxylic acid, 1,4-butanediol, N,N-dimethyl-1,3-propanediamine, 2,2-dimethyl-1,3-propanediol, 2-ethyl-1,3-heptanediol, hexanedioic acid, 2-hydroxyethyl 2-propenoate and 1,1'-methylenebis[4-isocyanatobenzene] (9CI) (CA INDEX NAME)

CM 1

CRN 39775-59-2 CMF C9 H20 O2

CM 2

CRN 818-61-1 CMF C5 H8 O3

CM 3

CRN 126-30-7 CMF C5 H12 O2

CM 4

CRN 124-04-9 CMF C6 H10 O4

$$HO_2C-(CH_2)_4-CO_2H$$

CM 5

CRN 121-91-5 CMF C8 H6 O4

CM 6

CRN 110-63-4 CMF C4 H10 O2

$$^{\rm HO-}$$
 (CH₂)₄-OH

CM 7

CRN 109-55-7 CMF C5 H14 N2

 $H_2N-(CH_2)_3-NMe_2$

CM 8

CRN 101-68-8 CMF C15 H10 N2 O2

CM 9

CRN 100-21-0 CMF C8 H6 O4

L48 ANSWER 17 OF 28 HCAPLUS COPYRIGHT 2003 ACS

AN 1997:243709 HCAPLUS

DN 126:239454

TI Antistatic films with good blocking and back-transfer resistance and bonding properties

IN Miura, Sadami; Kitazawa, Satoshi

PA Teijin Ltd, Japan

SO Jpn. Kokai Tokkyo Koho, 11 pp. CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C08J007-04

ICS B05D005-12; B05D007-04; B05D007-24; B29C055-02; B32B027-00; B32B027-18; B32B027-36; G11B005-704; B29K067-00; B29L007-00; B29L009-00; C08L067-00

CC 38-3 (Plastics Fabrication and Uses) Section cross-reference(s): 42

FAN.CNT 1

	PATENT NO.		KIND	DATE	APPLICATION NO.	DATE
			-			
ΡI	JP	09031222	A2	19970204	JP 1995-185773	19950721
	JP	3210211	B2	20010917		
PRAI	JP	1995-185773		19950721		•
GT				·		

Title films are obtained by applying aq. coatings contg. antistatic AΒ polymers (prepd. by polymg. thiophene and/or thiophene derivs.) on .qtoreq.1 side of polyester films, drying, and stretching [surface resistivity of the coating film (SR) 105-1012 .OMEGA./square]. Thus, a poly(ethylene terephthalate) undrawn film was stretched 3.5 times lengthwise, coated with a compn. contg. an antistatic polymer [prepd. by doping 24 parts p-styrenesulfonic acid homopolymer (27 mol% ionized) in 14 parts copolymer I] 38, 5:3:5:92:37:4:54 (mol) adipic acid-1,4-butanedioldiethylene glycol-ethylene glycol-isophthalic acid-Na 5-sulfoisophthalateterephthalic acid copolymer 35, and 14:45:7:3:3:32 Et acrylate-Et methacrylate-glycidyl methacrylate-2-hydroxyethyl acrylate-Nmethoxymethylacrylamide-Me methacrylate copolymer 15, poly(vinyl alc.) 2, and ethylene oxide-propylene oxide block copolymer 10% on one side of the film, dried, and stretched 3.9 times crosswise to give a laminated film having SR = 1.5 .times. 1011 .OMEGA./square, good back-transfer resistance, and good ink-bonding properties.

ST antistatic polyester film coating; thiophene polymer coating antistatic film; sulfonated polymer doped antistatic agent

IT Polyesters, uses

RL: PEP (Physical, engineering or chemical process); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(acrylic, binders, coatings contg.; thiophene polymer coatings for antistatic polyester films with good blocking and back-transfer resistances and bonding property)

IT Polyurethanes, uses

RL: PEP (Physical, engineering or chemical process); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(binders, coatings contg.; thiophene polymer coatings for antistatic polyester films with good blocking and back-transfer resistances and bonding property)

IT Acrylic polymers, uses

RL: PEP (Physical, engineering or chemical process); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(polyester-, binders, coatings contg.; thiophene polymer coatings for antistatic polyester films with good blocking and back-transfer resistances and bonding property)

IT Coating materials

(thiophene polymer coatings for antistatic polyester films with good blocking and back-transfer resistances and bonding property)

IT Polyesters, uses

RL: PEP (Physical, engineering or chemical process); PRP (Properties); TEM (Technical or engineered material use); PROC (Process); USES (Uses) (thiophene polymer coatings for antistatic polyester films with good blocking and back-transfer resistances and bonding property)
Antistatic agents

(thiophene polymers; thiophene polymer coatings for antistatic polyester films with good blocking and back-transfer resistances and bonding property)

IT 28210-41-5D, p-Styrenesulfonic acid homopolymer, hydrolyzed RL: PEP (Physical, engineering or chemical process); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(antistatic agent, dopant; thiophene polymer coatings for antistatic polyester films with good blocking and back-transfer resistances and bonding property)

IT 188297-89-4

IT

RL: PEP (Physical, engineering or chemical process); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(binder, coatings contg.; thiophene polymer coatings for antistatic polyester films with good blocking and back-transfer resistances and bonding property)

IT 187103-65-7 188297-86-1

RL: PEP (Physical, engineering or chemical process); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(binders, coatings contg.; thiophene polymer coatings for antistatic polyester films with good blocking and back-transfer resistances and bonding property)

IT 188297-88-3D, polymers with thiophene and 3-(sulfonylalkyl)thiophene sodium salt

RL: PEP (Physical, engineering or chemical process); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(doped with polystyrenesulfonic acid K salt, antistatic agent, coatings contg.; thiophene polymer coatings for antistatic polyester films with good blocking and back-transfer resistances and bonding property)

IT 110-02-1D, Thiophene, polymers with 3-sulfonylalkyl- and ethoxylated thiophene, sodium salts 616-44-4D, 3-MethylThiophene, polymers with 3-sulfonylalkyl- and 3-hydroxyalkylthiophenes, sodium salts RL: PEP (Physical, engineering or chemical process); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(doped with polystyrenesulfonic acid Na salt, antistatic agent, coatings contg.; thiophene polymer coatings for antistatic polyester films with good blocking and back-transfer resistances and bonding property)

IT 126213-51-2

RL: PEP (Physical, engineering or chemical process); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(doped with polystyrenesulfonic acid anion, antistatic agent, coatings contg.; thiophene polymer coatings for antistatic polyester films with good blocking and back-transfer resistances and bonding property)

IT 25038-59-9, uses

RL: PEP (Physical, engineering or chemical process); PRP (Properties); TEM (Technical or engineered material use); PROC (Process); USES (Uses) (thiophene polymer coatings for antistatic polyester films with good

blocking and back-transfer resistances and bonding property)

1T 9011-99-8, Polystyrenesulfonic acid potassium salt 9080-79-9,
Polystyrenesulfonic acid sodium salt
RL: PEP (Physical, engineering or chemical process); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material)

use); PROC (Process); USES (Uses)
(thiophene polymer doped with, antistatic agent, coatings contg.;
thiophene polymer coatings for antistatic polyester films with good
blocking and back-transfer resistances and bonding property)

IT 188297-89-4

RL: PEP (Physical, engineering or chemical process); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(binder, coatings contg.; thiophene polymer coatings for antistatic polyester films with good blocking and back-transfer resistances and bonding property)

RN 188297-89-4 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-sulfo-, monosodium salt, polymer with 1,3-benzenedicarboxylic acid, 1,4-benzenedicarboxylic acid, 1,4-butanediol, 1,2-ethanediol, ethyl 2-methyl-2-propenoate, ethyl 2-propenoate, hexanedioic acid, 2,2'-oxybis[ethanol] and 2-propenoic acid, graft (9CI) (CA INDEX NAME)

CM 1

CRN 6362-79-4 CMF C8 H6 O7 S . Na

Na

CM 2

CRN 140-88-5 CMF C5 H8 O2

CM 3

CRN 124-04-9 CMF C6 H10 O4

$$HO_2C-(CH_2)_4-CO_2H$$

CRN 121-91-5 CMF C8 H6 O4

CM 5

CRN 111-46-6 CMF C4 H10 O3

$${\tt HO-CH_2-CH_2-O-CH_2-CH_2-OH}$$

CM 6

CRN 110-63-4 CMF C4 H10 O2

$$HO-(CH_2)_4-OH$$

CM 7

CRN 107-21-1 CMF C2 H6 O2

$${\rm HO-CH_2-CH_2-OH}$$

CM 8

CRN 100-21-0 CMF C8 H6 O4

CRN 97-63-2 CMF C6 H10 O2

$$\begin{array}{c|c} ^{H2C} & \text{O} \\ & \parallel & \parallel \\ \text{Me-} \text{C-} \text{C-} \text{OEt} \end{array}$$

CM 10

CRN 79-10-7 CMF C3 H4 O2

L48 ANSWER 18 OF 28 HCAPLUS COPYRIGHT 2003 ACS

AN 1996:567044 HCAPLUS

DN 125:198702

TI Antistatic plastic films

IN Miura, Sadami; Kitazawa, Satoshi

PA Teijin Ltd, Japan

SO Jpn. Kokai Tokkyo Koho, 10 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C08J007-04

ICS B29C055-02; B32B027-18; B32B027-20; B32B027-30; B32B027-36; C09D005-00; C09D133-00; C09D133-14; C09D167-00

ICI B29K067-00, B29L009-00

CC 42-10 (Coatings, Inks, and Related Products)

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 08157625	A2	19960618	JP 1994-302161	19941206
PRAI	JP 1994-302161		19941206		

AB Title films are prepd. by coating polyester films with aq. compns. contg.
(A) quaternary ammonium compds. and/or quaternary ammonium group-contg.
polymers and (B) org. phosphates and/or phosphate ester group-contg.
polymers, drying, and drawing. A monoaxially drawn (in the machine direction) PET film was coated with an aq. compn. contg.
butanediol-diethylene glycol-ethylene glycol-isophthalic acid-neopentyl

glycol-5-sodiosulfonatoisophthalic acid-terephthalic acid copolymer 32, Et acrylate-Me methacrylate-3-(trimethylammonium)propyl acrylate methylsulfate copolymer 24, and (3-acryloxypropyl) decyl potassium phosphate-Et methacrylate-Me acrylate copolymer 29% to a 0.11-.mu.m thickness, drying, drawing in the transverse direction, and hot fixing gave a film with transparency 85% and resistivity 7.4 .times. 1011 .OMEGA..

- ST antistatic agent quaternary ammonium contg polymer; phosphate ester contg polymer antistatic agent; polyester film thin antistatic coating
- IT Quaternary ammonium compounds, uses
 RL: MOA (Modifier or additive use); TEM (Technical or engineered material
 use); USES (Uses)

(blends with phosphate ester-contg. compds. or polymers as antistatic agents for coatings for polyester films)

- IT Polyesters, miscellaneous
 - RL: MSC (Miscellaneous)

(films; quaternary ammonium group-contg. compds. or polymers and phosphate ester-contg. compds. or polymers as antistatic agents for coatings for polyester films)

IT Antistatic agents

(quaternary ammonium group-contg. compds. or polymers and phosphate ester-contg. compds. or polymers for coatings for polyester films)

- IT Polyesters, uses
 - RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (acrylic, graft, coating binder; quaternary ammonium group-contg. compds. or polymers and phosphate ester-contg. compds. or polymers as antistatic agents for coatings for polyester films)
- IT Acrylic polymers, uses
 - RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (polyester-, graft, coating binder; quaternary ammonium group-contg. compds. or polymers and phosphate ester-contg. compds. or polymers as antistatic agents for coatings for polyester films)
- Acrylic polymers, uses
 RL: IMF (Industrial manufacture); MOA (Modifier or additive use); TEM
 (Technical or engineered material use); PREP (Preparation); USES (Uses)
 (quaternary ammonium group-contg., blends with phosphate ester-contg.
 compds. or polymers as antistatic agents for coatings for polyester
 films)
- IT 181119-44-8P
 - RL: IMF (Industrial manufacture); MOA (Modifier or additive use); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (blends with phosphate ester-contg. compds. or polymers as antistatic agents for coatings for polyester films)
- IT 181119-53-9
 - RL: MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses)
 - (blends with phosphate ester-contg. compds. or polymers as antistatic agents for coatings for polyester films)
- IT 181119-49-3P
 - RL: IMF (Industrial manufacture); MOA (Modifier or additive use); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (blends with quaternary ammonium group-contg. compds. or polymers as

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antistatic agents for coatings for polyester films)
     50910-72-0, DisodiumTridecyl phosphate
IT
     RL: MOA (Modifier or additive use); TEM (Technical or engineered material
     use); USES (Uses)
        (blends with quaternary ammonium group-contg. compds. or polymers as
        antistatic agents for coatings for polyester films)
     181119-57-3P 181119-62-0P 181119-66-4P
IT
     RL: IMF (Industrial manufacture); POF (Polymer in
     formulation); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (coating binder; quaternary ammonium group-contg. compds. or
        polymers and phosphate ester-contg. compds. or polymers as antistatic
        agents for coatings for polyester films)
     181119-62-0P 181119-66-4P
IT
     RL: IMF (Industrial manufacture); POF (Polymer in
     formulation); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (coating binder; quaternary ammonium group-contg. compds. or
        polymers and phosphate ester-contg. compds. or polymers as antistatic
        agents for coatings for polyester films)
RN
     181119-62-0 HCAPLUS
     1,3-Benzenedicarboxylic acid, 5-sulfo-, monosodium salt, polymer with
CN
     1,3-benzenedicarboxylic acid, 1,4-benzenedicarboxylic acid, butanediol,
     2,2-dimethyl-1,3-propanediol, 1,2-ethanediol, ethyl 2-propenoate,
     2-hydroxyethyl 2-propenoate, N-(methoxymethyl)-2-propenamide, methyl
     2-methyl-2-propenoate, oxiranylmethyl 2-methyl-2-propenoate,
     2,2'-oxybis[ethanol] and potassium 2-propenoate, graft (9CI)
                                                                   (CA INDEX
     NAME)
     CM
          1
     CRN 25265-75-2
     CMF C4 H10 O2
     CCI IDS
H3C-CH2-CH2-CH3
   2 (D1-OH)
     CM
          2
     CRN 10192-85-5
     CMF
         C3 H4 O2 . K
HO-C-CH-CH2
```

K

CRN 6362-79-4 CMF C8 H6 O7 S . Na

Na

CM 4

CRN 3644-11-9 CMF C5 H9 N O2

$$\begin{tabular}{l} \begin{tabular}{l} \begin{tabu$$

CM 5

CRN 818-61-1 CMF C5 H8 O3

$$\begin{array}{c} \text{O} \\ || \\ \text{HO-CH}_2\text{-CH}_2\text{-O-C-CH-----} \text{CH}_2 \end{array}$$

CM 6

CRN 140-88-5 CMF C5 H8 O2

CM 7

CRN 126-30-7 CMF C5 H12 O2

CM 8

CRN 121-91-5 CMF C8 H6 O4

CM S

CRN 111-46-6 CMF C4 H10 O3

$$HO-CH_2-CH_2-O-CH_2-CH_2-OH$$

CM 10

CRN 107-21-1 CMF C2 H6 O2

$$HO-CH_2-CH_2-OH$$

CM 11

CRN 106-91-2 CMF C7 H10 O3

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ZALUKAEVA 09/917674 Page 118
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CRN 100-21-0 CMF C8 H6 O4

CM 13

CRN 80-62-6 CMF C5 H8 O2

RN 181119-66-4 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-sulfo-, monosodium salt, polymer with ammonium 2-propenoate, 1,3-benzenedicarboxylic acid, 1,4-benzenedicarboxylic acid, butanediol, 2,2-dimethyl-1,3-propanediol, 1,2-ethanediol, ethyl 2-methyl-2-propenoate, 2-hydroxyethyl 2-methyl-2-propenoate, methyl 2-propenoate, oxiranylmethyl 2-methyl-2-propenoate and 2,2'-oxybis[ethanol], graft (9CI) (CA INDEX NAME)

CM 1

CRN 25265-75-2 CMF C4 H10 O2 CCI IDS

H3C-CH2-CH2-CH3

2 (D1-OH)

CM 2

CRN 10604-69-0 CMF C3 H4 O2 . H3 N

$$0 \\ || \\ HO-C-CH = CH_2$$

● инз

CM 3

CRN 6362-79-4

CMF C8 H6 O7 S . Na

Na

CM 4

CRN 868-77-9 CMF C6 H10 O3

CM 5

CRN 126-30-7 CMF C5 H12 O2

$$\begin{array}{c} & \text{Me} \\ | \\ \text{HO-CH}_2 - \text{C-CH}_2 - \text{OH} \\ | \\ \text{Me} \end{array}$$

CM 6

CRN 121-91-5 CMF C8 H6 O4

CM 7

CRN 111-46-6 CMF C4 H10 O3

 ${\tt HO-CH_2-CH_2-O-CH_2-CH_2-OH}$

CM 8

CRN 107-21-1 CMF C2 H6 O2

но-сн2-сн2-он

CM 9

CRN 106-91-2 CMF C7 H10 O3

CM 10

CRN 100-21-0 CMF C8 H6 O4

CM 11

```
C6 H10 O2
     CMF
 H<sub>2</sub>C
   \parallel \parallel
Me-C-C-OEt
          12
     CM
          96-33-3
     CRN
     CMF C4 H6 O2
    0
MeO-C-CH-CH2
L48 ANSWER 19 OF 28 HCAPLUS COPYRIGHT 2003 ACS
     1996:523764 HCAPLUS
AN
DN
     125:171057
ΤI
     Grafting oligomers onto polymers
     Padget, John Christopher; Bedells, Alison Dawn; Overbeek, Gerardus
TN
PA
     Zeneca Limited, UK; Zeneca Resins B.V.
     PCT Int. Appl., 37 pp.
SO
     CODEN: PIXXD2
DT
     Patent
     English
LA
TC
     ICM C08F291-00
     ICS C09D151-00; C09J151-00; C08L051-00
     42-10 (Coatings, Inks, and Related Products)
CC
     Section cross-reference(s): 35
FAN.CNT 2
                                           APPLICATION NO. DATE
     PATENT NO.
                      KIND DATE
                     ____
                            _____
                                           _____
     WO 9619512
                      A1
                            19960627
                                           WO 1995-EP5021
                                                            19951219
PΙ
         W: AL, AM, AT, AU, BB, BG, BR, BY, CA, CH, CN, CZ, DE, DK, EE, ES,
             FI, GB, GE, HU, IS, JP, KE, KG, KP, KR, KZ, LK, LR, LS, LT, LU,
             LV, MD, MG, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI,
             SK, TJ
         RW: KE, LS, MW, SD, SZ, UG, AT, BE, CH, DE, DK, ES, FR, GB, GR, IE,
             IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR,
             NE, SN, TD, TG
     AU 9643875
                            19960710
                                           AU 1996-43875
                                                             19951219
                       Α1
     EP 799259
                       Α1
                            19971008
                                           EP 1995-942696
                                                             19951219
     EP 799259
                       В1
                           19990707
         R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE
     AT 181933
                           19990715
                                           AT 1995-942696
                      E
                                                            19951219
     ES 2133839
                       Т3
                            19990916
                                           ES 1995-942696
                                                             19951219
     US 5981642
                            19991109
                                           US 1997-849907
                       Α
                                                            19970805
PRAI GB 1994-26048
                       Α
                            19941221
     GB 1995-15719
                            19950801
                       Α
     WO 1995-EP5021
                            19951219
                       W
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97-63-2

CRN

A water-sol. org. oligomer(s) is grafted onto an org. polymer(s) in aq. AΒ latex form by free-radical polymn. of .gtoreq.1 monomer in a water-based mixt. of a water-sol. oligomer(s) and an aq. latex of a polymer(s). The extent of grafting is easily controlled. The products exhibit better response to associative thickeners (e.g., in paint formulations), coating compn. open time, and coating film properties than blends of the corresponding oligomer(s) and the polymer(s). A typical grafted product was manufd. by free-radical polymn. of Me methacrylate (I) in a mixt. of a 28.1% aq. 20:80 methacrylic acid-I copolymer soln. and a 45% solids 5:20:50:25 acrylic acid-Bu acrylate-Bu methacrylate-I copolymer latex. oligomer grafting polymer; methacrylate oligomer grafting acrylic polymer; ST coating oligomer grafted polymer Adhesives IT (grafting oligomers onto polymers in aq. media in presence of free-radical-polymerizable monomers for adhesives) IT (grafting oligomers onto polymers in aq. media in presence of free-radical-polymerizable monomers for inks) Polishing materials IT (grafting oligomers onto polymers in aq. media in presence of free-radical-polymerizable monomers for polishes) IT Polyamides, reactions Polyesters, reactions Urethane polymers RL: RCT (Reactant); RACT (Reactant or reagent) (oligomer; grafting oligomers onto polymers in aq. media in presence of free-radical-polymerizable monomers) IT Polyesters, uses RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (acrylic, graft, grafting oligomers onto polymers in aq. media in presence of free-radical-polymerizable monomers) IT Polymerization (graft, grafting oligomers onto polymers in aq. media in presence of free-radical-polymerizable monomers for coatings) IT Coating materials (lacquers, grafting oligomers onto polymers in aq. media in presence of free-radical-polymerizable monomers for lacquers) ITCoating materials (paints, grafting oligomers onto polymers in aq. media in presence of free-radical-polymerizable monomers for paints) IT Acrylic polymers, uses RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (polyester-, graft, grafting oligomers onto polymers in aq. media in presence of free-radical-polymerizable monomers) IT RL: RCT (Reactant); RACT (Reactant or reagent) (polymers, oligomer; grafting oligomers onto polymers in ag. media in presence of free-radical-polymerizable monomers) IT Coating materials (varnishes, grafting oligomers onto polymers in aq. media in presence of free-radical-polymerizable monomers for varnishes) IT 180604-56-2P

RL: IMF (Industrial manufacture); TEM (Technical or engineered material

(crosslinked coating; grafting oligomers onto polymers in aq. media in

use); PREP (Preparation); USES (Uses)

presence of free-radical-polymerizable monomers)

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TΤ
     180348-59-8P
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (crosslinked coating; grafting oligomers onto polymers in aq. media in
        presence of free-radical-polymerizable monomers for coatings)
ΙT
     180468-78-4P
     RL: IMF (Industrial manufacture); POF (Polymer in
     formulation); PREP (Preparation); USES (Uses)
        (grafting oligomers onto polymers in aq. media in presence of
        free-radical-polymerizable monomers for coatings)
     180348-58-7P
IT
     RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT
     (Reactant or reagent)
        (grafting oligomers onto polymers in aq. media in presence of
        free-radical-polymerizable monomers for coatings)
     180254-05-1P, Acrylic acid-butyl acrylate-butyl methacrylate-methacrylic
ΙT
     acid-methyl methacrylate graft copolymer
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (grafting oligomers onto polymers in aq. media in presence of
        free-radical-polymerizable monomers for coatings)
IT
     180468-78-4P
     RL: IMF (Industrial manufacture); POF (Polymer in
     formulation); PREP (Preparation); USES (Uses)
        (grafting oligomers onto polymers in aq. media in presence of
        free-radical-polymerizable monomers for coatings)
RN
     180468-78-4 HCAPLUS
CN
     1,3-Benzenedicarboxylic acid, sulfo-, monosodium salt, polymer with
     1,4-benzenedicarboxylic acid, butyl 2-methyl-2-propenoate, butyl
     2-propenoate, 4-cyclohexene-1,2-dicarboxylic acid, 2,2-dimethyl-1,3-
     propanediol, 2,2'-oxybis[ethanol] and 2-propenoic acid, ester with
     .alpha.-methyl-.omega.-hydroxypoly(oxy-1,2-ethanediyl), graft (9CI)
     INDEX NAME)
     CM
          1
     CRN
          9004-74-4
     CMF
          (C2 H4 O)n C H4 O
     CCI
но сн<sub>2</sub> - сн<sub>2</sub> - о сн<sub>3</sub>
     CM
          180254-06-2
     CRN
          (C8 H14 O2 . C8 H10 O4 . C8 H6 O7 S . C8 H6 O4 . C7 H12 O2 . C5 H12
     CMF
          O2 . C4 H10 O3 . C3 H4 O2 . Na)x
     CCI
          PMS
          CM
               3
               6362-79-4
          CRN
          CMF C8 H6 O7 S . Na
```

Na

CM 4

CRN 141-32-2 CMF C7 H12 O2

CM 5

CRN 126-30-7 CMF C5 H12 O2

CM 6

CRN 111-46-6 CMF C4 H10 O3

$${\tt HO-CH_2-CH_2-O-CH_2-CH_2-OH}$$

CM 7

CRN 100-21-0 CMF C8 H6 O4

CM 8

CRN 97-88-1 CMF C8 H14 O2

$$\begin{array}{c|c} & \text{O} & \text{CH}_2 \\ \parallel & \parallel \\ \text{n-BuO-C-C-Me} \end{array}$$

CM 9

CRN 88-98-2 CMF C8 H10 O4

CM 10

CRN .79-10-7 CMF C3 H4 O2

L48 ANSWER 20 OF 28 HCAPLUS COPYRIGHT 2003 ACS

AN 1996:241585 HCAPLUS

DN 124:268454

TI Water permeable paving materials and their application

IN Okamoto, Taneo; Shimooka, Shizuo

PA Kanebo Nsc Kk, Japan

SO Jpn. Kokai Tokkyo Koho, 16 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM E01C007-30

ICA C08F290-06

CC 58-4 (Cement, Concrete, and Related Building Materials)

Section cross-reference(s): 38

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE

PI JP 08013407 A2 19960116 JP 1994-146744 19940628

PRAI JP 1994-146744 19940628

AB The paving materials are aggregates mixed with a binder, which contains a polymerizable unsatd. monomer, a vinyl ester resin, an air oxidizable component, and a thixotropic agent. The paving mixts. are applied on substrates.

ST water permeable pavement polymer aggregate mixt; thixotropic additive polymer aggregate paving mixt; unsatd polyester water permeable pavement

IT Pavements and Roads

(binder compns. for water permeable paving materials)

IT 36425-15-7P 167747-71-9P 175596-67-5P **175596-68-6P** 175596-69-7P

RL: IMF (Industrial manufacture); PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (Uses)

(binder compns. for water permeable paving materials)

IT 80-62-6, Methyl methacrylate 97-86-9, Iso-butyl methacrylate 97-88-1, n-Butyl methacrylate 141-32-2

RL: TEM (Technical or engineered material use); USES (Uses) (binder compns. for water permeable paving materials)

IT 175596-68-6P

RL: IMF (Industrial manufacture); PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (Uses)

(binder compns. for water permeable paving materials)

RN 175596-68-6 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, polymer with 2-(hydroxymethyl)-2-[[(1-oxo-2-propenyl)oxy]methyl]-1,3-propanediyl di-2-propenoate and 2,2'-oxybis[ethanol] (9CI) (CA INDEX NAME)

CM 1

CRN 3524-68-3 CMF C14 H18 O7

CM 2

CRN 111-46-6 CMF C4 H10 O3

HO-CH2-CH2-O-CH2-CH2-OH

CRN 100-21-0 CMF C8 H6 O4

L48 ANSWER 21 OF 28 HCAPLUS COPYRIGHT 2003 ACS

AN 1995:838997 HCAPLUS

DN 123:325722

TI Binder resin for electrophotography, its manufacture, and toner composition for electrophotography

IN Aoki, Katsutoshi

PA Kao Corp, Japan

SO Jpn. Kokai Tokkyo Koho, 10 pp. CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM G03G009-08 ICS G03G009-087

CC 74-3 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

Section cross-reference(s): 38

FAN.CNT 1

PRAI JP 1993-327041

AB The binder resin, which may have a softening point of 95.0-170.0.degree. and a glass transition point of 50.0-80.0.degree., comprises a fluoromica of .alpha.MF..beta.(aMgF2.bMgO)..gamma.SiO2 (M = Na, Li, K; .alpha. = 0.1-2; .beta. = 2-3.5; .gamma. = 3-4; a, b = 0-1, a + b = 1). The fluoromica may be obtained by heat-treatment of mixed micropowders of 10-35 wt.% of M2SiF6 (M = Na, Li, K) and talc. The toner compn. contains the electrophotog. binder resin, colorant and charge controlling agent. In the manuf., the fluoromica is added in the synthetic process of the binder resin. The binder resin shows good chargeability, transparency and low-temp. fixability.

ST fluoromica electrophotog binder resin toner

IT Polyesters, preparation

RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(binder resin for electrophotog. with fluoromica)

IT Mica-group minerals, preparation

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(prepn. of fluoromica from talc and alkali fluorosilicate)

IT Electrophotographic developers

(toners, fluoromica-contg. binder resin for electrophotog. toner)

IT Polyesters, preparation

RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(vinyl group-contg., binder resin for electrophotog. with fluoromica)

IT 25213-39-2P, Styrene-butyl methacrylate copolymer 89993-85-1P 164254-86-8P 164366-98-7P

RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(binder resin for electrophotog. with fluoromica)

IT 14807-96-6, Talc, reactions 16893-85-9, Disodium hexafluorosilicate
RL: RCT (Reactant); RACT (Reactant or reagent)

(in prepn. of fluoromica from talc and alkali fluorosilicate)

IT 164254-86-8P

RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(binder resin for electrophotog. with fluoromica)

RN 164254-86-8 HCAPLUS

CN 1,2,4-Benzenetricarboxylic acid, polymer with 1,3-benzenedicarboxylic acid, butyl 2-methyl-2-propenoate, 2,2-dimethyl-1,3-propanediol, ethenylbenzene and 2-methyl-2-propenoic acid (9CI) (CA INDEX NAME)

CM 1

CRN 528-44-9 CMF C9 H6 O6

CM 2

CRN 126-30-7 CMF C5 H12 O2

CM 3

CRN 121-91-5 CMF C8 H6 O4

CRN 100-42-5 CMF C8 H8

 $H_2C = CH - Ph$

CM 5

CRN 97-88-1 CMF C8 H14 O2

$$\begin{array}{c|c} & \text{O} & \text{CH}_2 \\ & \parallel & \parallel \\ \text{n-BuO-C-C-Me} \end{array}$$

CM 6

CRN 79-41-4 CMF C4 H6 O2

$$\begin{array}{c} \text{CH}_2 \\ || \\ \text{Me-C-CO}_2\text{H} \end{array}$$

L48 ANSWER 22 OF 28 HCAPLUS COPYRIGHT 2003 ACS

AN 1995:571019 HCAPLUS

DN 123:22117

TI Polyesters containing unsaturated end groups for low-temperature fixing electrophotographic toner binders

IN Bayley, Robert D.; Fox, Carol A.; Hoffend, Thomas; Packson, James R.; Sacripante, Guerino G.

PA Xerox Corp., Japan

SO Jpn. Kokai Tokkyo Koho, 11 pp. CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM G03G009-087

CC 74-3 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

FAN.CNT 1

```
APPLICATION NO. DATE
                      KIND
                          DATE
     PATENT NO.
                                           _____
                            _____
                                           JP 1994-174496
                                                            19940726
                            19950310
     JP 07064329
                      A2
PΙ
                            19930803
PRAI US 1993-100937
     The title polyesters contains a monofunctional unsatd. end group(s).
AΒ
     toner using the above polyesters showed excellent offset resistance and
     superior vinyl offset properties.
     electrophotog development toner polyester binder
ΙT
     Polyesters, preparation
     RL: DEV (Device component use); IMF (Industrial manufacture); PREP
     (Preparation); USES (Uses)
        (electrophotog. toner binders)
ΙT
     Electrophotographic developers
        (toners, polyesters contg. unsatd. end groups for low-temp. fixing
        electrophotog. toner binders)
                                              163310-40-5P
ΙT
     162752-35-4P 163310-38-1P 163310-39-2P
     163310-41-6P
     RL: DEV (Device component use); IMF (Industrial manufacture);
     PREP (Preparation); USES (Uses)
        (electrophotog. toner binders)
     163310-38-1P 163310-39-2P 163310-41-6P
TΤ
     RL: DEV (Device component use); IMF (Industrial manufacture);
     PREP (Preparation); USES (Uses)
        (electrophotog. toner binders)
     163310-38-1 HCAPLUS
RN
     1,4-Benzenedicarboxylic acid, dimethyl ester, polymer with butyl
CN
                                                                            (CA
     2-methyl-2-propenoate, 2,2'-oxybis[ethanol] and 1,2-propanediol (9CI)
     INDEX NAME)
     CM
          1
     CRN 120-61-6
     CMF C10 H10 O4
```

CRN 111-46-6 CMF C4 H10 O3

 $HO-CH_2-CH_2-O-CH_2-CH_2-OH$

CM 3

CRN 97-88-1 CMF C8 H14 O2

$$\begin{array}{c|c} & \text{O} & \text{CH}_2 \\ & || & || \\ \text{n-BuO-C-C-Me} \end{array}$$

CM 4

CRN 57-55-6 CMF C3 H8 O2

RN 163310-39-2 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, dimethyl ester, polymer with methyl 2-methyl-2-propenoate, 2,2'-oxybis[ethanol] and 1,2-propanediol, graft (9CI) (CA INDEX NAME)

CM 1

CRN 120-61-6 CMF C10 H10 O4

CM 2

CRN 111-46-6 CMF C4 H10 O3

$$HO-CH_2-CH_2-O-CH_2-CH_2-OH$$

CM 3

CRN 80-62-6 CMF C5 H8 O2

$$H_2C$$
 O \parallel \parallel \parallel $Me-C-C-OMe$

CRN 57-55-6 CMF C3 H8 O2

RN 163310-41-6 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, dimethyl ester, polymer with butyl 2-methyl-2-propenoate, 2,2'-oxybis[ethanol] and 1,2-propanediol, graft (9CI) (CA INDEX NAME)

CM 1

CRN 120-61-6 CMF C10 H10 O4

CM 2

CRN 111-46-6 CMF C4 H10 O3

 $HO-CH_2-CH_2-O-CH_2-CH_2-OH$

CM 3

CRN 97-88-1 CMF C8 H14 O2

CRN 57-55-6 CMF C3 H8 O2

ОН | | | Н3С-СН-СН2-ОН

L48 ANSWER 23 OF 28 HCAPLUS COPYRIGHT 2003 ACS

AN 1995:331339 HCAPLUS

DN 123:34795

TI Electron beam-curable adhesives containing polyesters for laminating plastics and metals

IN Nishimoto, Tadashi; Suzuki, Yoshio; Shioda, Atsushi; Masuda, Hideki; Tako, Noboru

PA Kansai Paint Co Ltd, Japan

SO Jpn. Kokai Tokkyo Koho, 9 pp. CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C09J004-02

ICA C08F299-04

CC 38-3 (Plastics Fabrication and Uses) Section cross-reference(s): 55, 56

FAN CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE

----PI JP 06299121 A2 19941025 JP 1993-118906 19930412
PRAI JP 1993-118906 19930412

Title adhesives, useful for laminating polyester films and metal plates for manuf. of of cans, etc., contain (A) 100 parts mixts. of (a) 100 parts polyesters of no. av. mol. wt. (M) 7000-40,000 and (b) 2-200 parts polyester oligomers of M 300-5000 including 0.3-5.0 mol/kg polymerizable unsatd. double bonds (X), (B) 0.2-50 parts maleated-modified polypropylene and/or [(CH2:CR)CO2(CH2)m]nOPO(OH)3-n (R = H, Me; m = 2, 3; n = 1, 2) as agents for enhancing adhesion, and (C) 0.5-250 parts stress-relaxation agents. Thus, 227 parts 30% soln. of 18.3:11.7:11.9:20.5:37.6 di-Me terephthalate-di-Me isophthalate-ethylene glycol-neopentyl glycol-azelaic acid copolymer (M 30,000) in 1:1 mixt. of MEK and cyclohexanone, 17 parts 1.0:2.0:1.0:3.0 (mol) trimethylolpropane-triethylene glycol-1,6-hexanedioladipic acid copolymer acrylate (M 850, X 3.0 mol/kg), 0.5 part methacryloyloxyethyl phosphate, 0.5 part bis(methacryloyloxyethyl) phosphate, and 50 parts Aerosil 300 were mixed and dild. with MEK to give 30%-solid adhesive soln. The soln. was gravure-coated onto E 5100 (polyester film), dried at 100.degree. for 10 s, laminated to a tinplate substrate, rolled at 180.degree., and electron beam-irradiated to give a test piece showing good resistance to whitening after 30-min exposure to

```
steam in autoclave at 125.degree..
    polyester vinyl polymer blend adhesive; electron beam curable adhesive;
ST
    metal plate polyester film adhesive; can polyester film metal adhesive;
    unsatd acid polyester ester adhesive; maleic anhydride modified
    polypropylene tackifier; phosphate ester unsatd tackifier adhesive;
    acrylic polyester oligomer curable adhesive; stress relaxation silica
    blend adhesive; retort resistance metal laminate adhesive; steam pressure
    resistance adhesive can
    Crosslinking
IT
        (electron beam crosslinking of polyester-based adhesive contg. vinyl
        components)
ΙT
    Polyesters, uses
    RL: POF (Polymer in formulation); TEM (Technical or engineered material
    use); USES (Uses)
        (electron beam-curable adhesives contg. polyesters and unsatd.
        polyester oligomers and vinyl compds. for laminating metal plates)
IT
    Tinplate
     RL: MSC (Miscellaneous)
        (plates, substrates; electron beam-curable adhesives contg. polyesters
        and unsatd. polyester oligomers and vinyl compds. for laminating metal
        plates)
TТ
     Polyesters, uses
    RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (acrylic, electron beam-curable adhesives contg. polyesters and unsatd.
        polyester oligomers and vinyl compds. for laminating metal plates)
IT
    Adhesives
        (electron-beam-curable, electron beam-curable adhesives contg.
        polyesters and unsatd. polyester oligomers and vinyl compds. for
        laminating metal plates)
                                 164203-76-3P
                                               164204-73-3P
                                                                164204-74-4P
IT
     58086-24-1P
                   63929-60-2P
     RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM
     (Technical or engineered material use); PREP (Preparation); USES (Uses)
        (electron beam-curable adhesives contg. polyesters and unsatd.
        polyester oligomers and vinyl compds. for laminating metal plates)
IT
     164204-76-6P
     RL: IMF (Industrial manufacture); TEM (Technical or engineered
    material use); PREP (Preparation); USES (Uses)
        (electron beam-curable adhesives contg. polyesters and unsatd.
        polyester oligomers and vinyl compds. for laminating metal
        plates)
IT
     25038-59-9, PET, miscellaneous
     RL: MSC (Miscellaneous)
        (films, substrates; electron beam-curable adhesives contg. polyesters
        and unsatd. polyester oligomers and vinyl compds. for laminating metal
        plates)
     7727-43-7, Barium sulfate
IT
     RL: MOA (Modifier or additive use); USES (Uses)
        (stress-relaxation agents, Barifine; electron beam crosslinking of
        polyester-based adhesive contq. vinyl components)
IT
     13463-67-7, Titania, uses
     RL: MOA (Modifier or additive use); USES (Uses)
        (stress-relaxation agents, JR 300; electron beam crosslinking of
        polyester-based adhesive contg. vinyl components)
IT
     7631-86-9, Aerosil 300, uses
     RL: MOA (Modifier or additive use); USES (Uses)
        (stress-relaxation agents; electron beam crosslinking of
        polyester-based adhesive contg. vinyl components)
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IT 164204-76-6P
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RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (electron beam-curable adhesives contg. polyesters and unsatd. polyester oligomers and vinyl compds. for laminating metal plates)

RN 164204-76-6 HCAPLUS

1,3-Benzenedicarboxylic acid, dimethyl ester, polymer with dimethyl 1,4-benzenedicarboxylate, 2,2-dimethyl-1,3-propanediol, 1,2-ethanediol and nonanedioic acid, 2-propenoate, polymer with 2,2'-[1,2-ethanediylbis(oxy)]bis[ethanol] polymer with 2-ethyl-2-(hydroxymethyl)-1,3-propanediol, hexanedioic acid and 1,6-hexanediol 2-propenoate, and 2-hydroxyethyl 2-methyl-2-propenoate phosphate and phosphinicobis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate) (9CI) (CA INDEX NAME)

CM I

CRN 32435-46-4 CMF C12 H19 O8 P

CM 2

CRN 164204-75-5 CMF (C10 H10 O4 . C10 H10 O4 . C9 H16 O4 . C5 H12 O2 . C2 H6 O2)x . x C3 H4 O2

CM 3

CRN 79-10-7 CMF C3 H4 O2

CM 4

CRN 58086-24-1

CMF (C10 H10 O4 . C10 H10 O4 . C9 H16 O4 . C5 H12 O2 . C2 H6 O2)x CCI PMS

CM 5

CRN 1459-93-4 CMF C10 H10 O4

CM 6

CRN 126-30-7 CMF C5 H12 O2

$$\begin{array}{c} \text{Me} \\ \mid \\ \text{HO-CH}_2\text{--}\text{C-CH}_2\text{--OH} \\ \mid \\ \text{Me} \end{array}$$

CM 7

CRN 123-99-9 CMF C9 H16 O4

 $HO_2C-(CH_2)_7-CO_2H$

CM 8

CRN 120-61-6 CMF C10 H10 O4

CM 9

CRN 107-21-1 CMF C2 H6 O2 $HO-CH_2-CH_2-OH$

CM 10

CRN 69772-12-9

CMF (C6 H14 O4 . C6 H14 O3 . C6 H14 O2 . C6 H10 O4)x . x C3 H4 O2

CM 11

CRN 79-10-7 CMF C3 H4 O2

о || но- с- сн== сн₂

CM 12

CRN 68310-36-1

CMF (C6 H14 O4 . C6 H14 O3 . C6 H14 O2 . C6 H10 O4) x

CCI PMS

CM 13

CRN 629-11-8

CMF C6 H14 O2

 $^{\rm HO-}$ (CH₂)₆ $^{\rm -}$ OH

CM 14

CRN 124-04-9

CMF C6 H10 O4

 $HO_2C-(CH_2)_4-CO_2H$

CM 15

CRN 112-27-6

CMF C6 H14 O4

 $HO-CH_2-CH_2-O-CH_2-CH_2-O-CH_2-CH_2-OH$

CM 16

CRN 77-99-6 CMF C6 H14 O3

$$\begin{array}{c|c} & \text{CH}_2-\text{OH} \\ & | \\ \text{HO-CH}_2-\text{C-Et} \\ & | \\ \text{CH}_2-\text{OH} \end{array}$$

CM 17

CRN 52628-03-2 CMF C6 H10 O3 . x H3 O4 P

CM 18

CRN 7664-38-2 CMF H3 O4 P

CM 19

CRN 868-77-9 CMF C6 H10 O3

L48 ANSWER 24 OF 28 HCAPLUS COPYRIGHT 2003 ACS

AN 1994:137310 HCAPLUS

DN 120:137310

TI Magnetic coating materials containing polyurethane binders

IN Kinoshita, Koji; Nakama, Yasutaka; Komazaki, Shigeru; Oooka, Masataka

PA Dainippon Ink & Chemicals, Japan

SO Jpn. Kokai Tokkyo Koho, 16 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C09D005-23

ICS C09D175-04; G11B005-702

CC 42-10 (Coatings, Inks, and Related Products)

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Section cross-reference(s): 77
FAN.CNT 1
                                           APPLICATION NO. DATE
     PATENT NO.
                      KIND DATE
                                           _____
                                                             _____
                      A2
                            19930907
                                           JP 1992-31964
                                                            19920219
     JP 05230398
PΙ
PRAI JP 1992-31964
                            19920219
     The title coating materials providing recording media with good smoothness
     and abrasion resistance, comprise vinyl polymer-polyurethane block
     copolymers, magnetic powders, polyurethanes, and optionally
     polyisocyanates. Thus, styrene 180, Me methacrylate 550, 2-hydroxyethyl
     methacrylate 50, and 2-acrylamido-2-methylpropanesulfonic acid 20 parts
     were polymd. in the presence of 500 parts polymeric azo initiator [prepd.
     from azobis(cyanopropanol), dipropylene glycol, and HDI] to give a block
     copolymer with no.-av. mol. wt. 220 .times. 102 (I). Then, a soln. contg.
     I 70, adipic acid-1,4-butanediol-TDI copolymer 30, Fe-Ni alloy powder 250,
     carbon black 8.6, a lubricant 3.5, Burnock D 750 14.3, and MEK 738 parts
     was applied on a PET film, dried, and calendered to give a smooth magnetic
     tape with 45.degree. gloss 115%, Taber abrasion test 655 cycles, and good storage stability at 70.degree. and 95% relative humidity.
     polyurethane binder magnetic recording coating; polyacrylate polyurethane
ST
     magnetic coating smoothness; abrasion resistance magnetic coating
     polyurethane
IT
     Coating materials
        (abrasion-resistant, magnetic, binders for, block vinyl
        polymer-polyurethanes as, for recording tapes)
IT
     Urethane polymers, preparation
     RL: PREP (Preparation)
        (acrylic, block, prepn. of, binders, for abrasion-resistant smooth
        magnetic coatings)
     Recording materials
IT
        (magnetic, coatings for, block vinyl polymer-polyurethane binders in)
IT
     Urethane polymers, preparation
     RL: PREP (Preparation)
        (vinyl polymer-, block, prepn. of, binders, for abrasion-resistant
        smooth magnetic coatings)
                    152242-15-4P
                                   152242-16-5P
                                                   152242-17-6P
ΙT
     152242-14-3P
                    152242-19-8P
                                   153301-21-4P
     152242-18-7P
     RL: PREP (Preparation)
        (prepn. of, binders, for smooth magnetic coatings for
        recording materials)
IT
     152242-18-7P
     RL: PREP (Preparation)
        (prepn. of, binders, for smooth magnetic coatings for
        recording materials)
     152242-18-7 HCAPLUS
RN
     1,3-Benzenedicarboxylic acid, polymer with 2,2'-azobis[3-hydroxy-2-
CN
     methylpropanenitrile], 1,4-benzenedicarboxylic acid, Burnock D 750,
     1,4-butanediol, 1,6-diisocyanatohexane, 2,2-dimethyl-1,3-propanediol,
     1,2-ethanediol, ethenylbenzene, hexanedioic acid, methyl
     2-methyl-2-propenoate and 2-methyl-2-[(1-oxo-2-propenyl)amino]-1-
     propanesulfonic acid, block (9CI) (CA INDEX NAME)
     CM
     CRN 50813-68-8
     CMF Unspecified
     CCI MAN
```

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

CM 2

CRN 19706-80-0 CMF C8 H12 N4 O2

CM 3

CRN 15214-89-8 CMF C7 H13 N O4 S

CM 4

CRN 822-06-0 CMF C8 H12 N2 O2

$$ocn-(ch_2)_6-nco$$

CM 5

CRN 126-30-7 CMF C5 H12 O2

CM 6

CRN 124-04-9 CMF C6 H10 O4

 $HO_2C-(CH_2)_4-CO_2H$

CM 7

CRN 121-91-5 CMF C8 H6 O4

CM 8

CRN 110-63-4 CMF C4 H10 O2

 $^{\rm HO-}$ (CH₂)₄-OH

CM 9

CRN 107-21-1 CMF C2 H6 O2

 ${\tt HO-CH_2-CH_2-OH}$

CM 10

CRN 100-42-5 CMF C8 H8

 $H_2C = CH - Ph$

CM 11

CRN 100-21-0 CMF C8 H6 O4

CRN 80-62-6 CMF C5 H8 O2

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L48 ANSWER 25 OF 28 HCAPLUS COPYRIGHT 2003 ACS
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AN 1992:540595 HCAPLUS

DN 117:140595

TI Electrostatographic toner for heat-roll fixing

IN Kuriyama, Kazuya; Matsukuri, Kinji; Sugawara, Ryozo; Furuta, Hideyuki; Tomita, Yasushi

PA Dainippon Inki Kagaku Kogyo K. K., Japan

SO Jpn. Kokai Tokkyo Koho, 7 pp. CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM G03G009-087

CC 74-3 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE

PI JP 04086828 A2 19920319 JP 1990-203720 19900731

PRAI JP 1990-203720 19900731

AB The title toner contains a binder resin obtained by copolymn. of an unsatd. polyester and vinyl monomers including a multifunctional vinyl monomer(s) 0.01 - 5.0 wt.% relative to the total amt. of vinyl monomers used. The unsatd. polyester may incorporate an aliph. unsatd. dibasic acid(s) 0.2 - 5.0 wt.% and have no. av. mol. wt. 500 - 10,000. This toner shows good fixability and improved resistance to offset.

ST electrostatog toner binder heat roll fixing; graft copolymer binder electrophotog toner

IT Electrography

(developers, toners, binders, graft copolymers as, for good fixability)

IT Electrophotographic developers

(toners, binders, graft copolymers as, for good fixability)

IT 143410-78-0P

RL: PREP (Preparation)

(prepn. of, as binder, electrostatog toner contg.)

IT 143410-77-9P

RL: PREP (Preparation)

(prepn. of, as binder, for electrostatog toner)

IT 143410-78-0P

RL: PREP (Preparation)

(prepn. of, as binder, electrostatog toner contg.)

RN 143410-78-0 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, polymer with butyl 2-propenoate, diethenylbenzene, 2,2-dimethyl-1,3-propanediol, ethenylbenzene, 2,5-furandione and methyl 2-methyl-2-propenoate, graft (9CI) (CA INDEX NAME)

CM 1

CRN 1321-74-0 CMF C10 H10 CCI IDS



CM 2

CRN 141-32-2 CMF C7 H12 O2

$$\begin{array}{c} \text{O} \\ \parallel \\ \text{n-BuO-C-CH} \end{array} \text{CH}_2$$

CM 3

CRN 126-30-7 CMF C5 H12 O2

CM 4

CRN 108-31-6 CMF C4 H2 O3

CRN 100-42-5 CMF C8 H8

 $H_2C = CH - Ph$

CM 6

CRN 100-21-0 CMF C8 H6 O4

CM 7

CRN 80-62-6 CMF C5 H8 O2

L48 ANSWER 26 OF 28 HCAPLUS COPYRIGHT 2003 ACS

AN 1990:79614 HCAPLUS

DN 112:79614

TI Polyurethanes as binders for coatings

IN Dejima, Hironari; Takano, Masahiro; Hirayama, Yoichi; Akasaka, Nobuyuki

PA Fujikura Kasei Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 7 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C08G018-83 ICS C08G018-65

ICA C09D003-72; C09J003-16

CC 42-10 (Coatings, Inks, and Related Products)

Section cross-reference(s): 35

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE

PI JP 01188515 A2 19890727 JP 1988-10762 19880122 PRAI JP 1988-10762 19880122

AB Title polymers, useful for coatings on polystyrene substrates, are prepd. by polymg. (meth)acryloyl-contg. diols, high-mol.-wt. diols, chain extenders, and diisocyanates and grafting (meth)acrylate esters on the resulting polymers. Thus, glycerol methacrylate 2.56, adipic acid-dimethyl terephthalate-ethylene glycol-neopentyl glycol copolymer 403.3, 1,4-butanediol 18.7, MDI 100, dibutyltin dilaurate 0.3, MEK 600, and toluene 624 parts were heated at 70-80.degree. under N to give a polymer with no.-av. mol. wt. 20,000 and methacryloyl content 2 mol%, 300 parts of which was heated with 90 parts iso-Bu methacrylate and AIBN in MEK at 70.degree. and dild. with MEK to 30% solids. The soln. with viscosity 230 cP was storage-stable and formed a transparent film on a polystyrene sheet with good adhesion.

ST acryloyl contg polyurethane binder coating; graft acrylate polyurethane binder coating; transparency polyurethane binder coating; storage stability polyurethane binder coating; polystyrene coating polyurethane binder

IT Coating materials

(binders for, (meth)acrylate-grafted (meth)acryloyl-contg. polyurethanes as, with good adhesion to polystyrene)

IT 9003-53-6, Polystyrene

RL: USES (Uses)

(coatings for, binders for)

IT 125198-72-3P 125198-73-4P

RL: PREP (Preparation)

(manuf. of, binders for coatings, storage-stable, transparent, with good adhesion to polystyrene)

IT 125198-72-3P 125198-73-4P

RL: PREP (Preparation)

(manuf. of, binders for coatings, storage-stable, transparent, with good adhesion to polystyrene)

RN 125198-72-3 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, dimethyl ester, polymer with 1,4-butanediol, 2,2-dimethyl-1,3-propanediol, 1,2-ethanediol, hexanedioic acid, 1,1'-methylenebis[4-isocyanatobenzene], 2-methylpropyl 2-methyl-2-propenoate and 1,2,3-propanetriol mono(2-methyl-2-propenoate), block, graft (9CI) (CA INDEX NAME)

CM 1

CRN 126-30-7 CMF C5 H12 O2

CM 2

CRN 124-04-9 CMF C6 H10 O4 $HO_2C-(CH_2)_4-CO_2H$

CM 3

CRN 120-61-6 CMF C10 H10 O4

CM 4

CRN 110-63-4 CMF C4 H10 O2

 ${\tt HO-(CH_2)_4-OH}$

CM 5

CRN 107-21-1 CMF C2 H6 O2

 ${\rm HO}-{\rm CH_2}-{\rm CH_2}-{\rm OH}$

CM 6

CRN 101-68-8 CMF C15 H10 N2 O2

CM 7

CRN 97-86-9 CMF C8 H14 O2

$$\begin{array}{c|c} \text{O} & \text{CH}_2 \\ \parallel & \parallel \\ \text{i-BuO-C-C-Me} \end{array}$$

CM 8

CRN 50853-28-6 CMF C7 H12 O4

CCI IDS

CM 9

CRN 79-41-4 CMF C4 H6 O2

$$\begin{array}{c} \text{CH}_2 \\ || \\ \text{Me-C-CO}_2 \text{H} \end{array}$$

CM 10

CRN 56-81-5 CMF C3 H8 O3

RN 125198-73-4 HCAPLUS

CN 1,4-Benzenedicarboxylic acid, dimethyl ester, polymer with 1,4-butanediol, butyl 2-methyl-2-propenoate, 2,2-dimethyl-1,3-propanediol, 1,2-ethanediol, hexanedioic acid, 1,1'-methylenebis[4-isocyanatobenzene] and 1,2,3-propanetriol mono(2-methyl-2-propenoate), block, graft (9CI) (CA INDEX NAME)

CM 1

CRN 126-30-7 CMF C5 H12 O2

$$\begin{array}{c} \text{Me} \\ \mid \\ \text{HO-CH}_2\text{-C-CH}_2\text{-OH} \\ \mid \\ \text{Me} \end{array}$$

CRN 124-04-9 CMF C6 H10 O4

 $HO_2C-(CH_2)_4-CO_2H$

CM 3

CRN 120-61-6 CMF C10 H10 O4

CM 4

CRN 110-63-4 CMF C4 H10 O2

 $_{\text{HO}^-}$ (CH₂)₄ $^-$ OH

CM 5

CRN 107-21-1 CMF C2 H6 O2

но-сн2-сн2-он

CM 6

CRN 101-68-8 CMF C15 H10 N2 O2

CM 7

CRN 97-88-1 CMF C8 H14 O2

$$\begin{array}{c|c} & \text{O} & \text{CH}_2 \\ \parallel & \parallel \\ \text{n-BuO-C-C-Me} \end{array}$$

CM 8

CRN 50853-28-6 CMF C7 H12 O4 CCI IDS

CM 9

CRN 79-41-4 CMF C4 H6 O2

$$\begin{array}{c} \text{CH}_2 \\ || \\ \text{Me-C-CO}_2 \text{H} \end{array}$$

CM 10

CRN 56-81-5 CMF C3 H8 O3

L48 ANSWER 27 OF 28 HCAPLUS COPYRIGHT 2003 ACS

AN 1987:34752 HCAPLUS

DN 106:34752

TI Radiation-curable binders for magnetic coating materials

IN Ansel E, Robert; Ukaji, Takashi; Bettsho, Keiichi; Kumano, Koji; Matsumura, Yoshio

PA DeSoto, Inc., Japan

SO Jpn. Kokai Tokkyo Koho, 5 pp.

CODEN: JKXXAF DT Patent LΑ Japanese ICM C09D005-00 IC ICS C09D003-727; C09D005-23 ICA C08F299-02 42-7 (Coatings, Inks, and Related Products) Section cross-reference(s): 37, 77 FAN.CNT 1 PATENT NO. KIND DATE APPLICATION NO. DATE ____ A2 19860814 JP 61181872 JP 1985-16357 19850130 PΙ PRAI JP 1985-16357 19850130 The title binders having good compatibility with magnetic powders and low AΒ viscosity for good workability and leveling and forming abrasion-resistant coatings with excellent magnetic characteristics were described including various acrylic-terminated polymers (contg. urethane, urea, amide, and/or ester linkages) of mol. wt. 2000-100,000. Thus, a mixt. of methylenebis(4-cyclohexyl isocyanate) 71.9, dibutyltin dilaurate 0.2, and MEK 300 g at 60.degree. was treated with a mixt. of 125.6 g Teracol 650 and 15.5 g Epikote 828 diacrylate, stirred at 60.degree. for 4 h, treated with 9.6 g pentaerythritol triacrylate at 60.degree. for 2 h, and treated with 77 g HOZQZQZOH (Z = polyoxytetramethylene; Q = pyromellitic acid residue) at 60.degree. for 7 h to give an electron beam-curable binder resin. acrylic electron beam curable coating; magnetic coating electron beam ST curable; polyurethane acrylate coating radiation curable; epoxy acrylate polyurethane coating; polyoxytetramethylene polypyromellitate acrylic coating IT Coating materials (electron-beam-curable, magnetic, binders for, urethane- and urea- and amide-group-contg. acrylic polymers as) Recording materials ΙT (magnetic, binders for, urethane- and urea- and amide-group-contg. acrylic polymers as, electron beam-curable) 106056-72-8P 106072-77-9P 106100-63-4P 106100-64-5P 106209-18-1P IT 106209-62-5P 106222-72-4P 106222-74-6P 106222-75-7P 106247-18-1P 106284-62-2P RL: PREP (Preparation) (manuf. of, as electron beam-curable binders, for magnetic coatings) TT 106284-62-2P RL: PREP (Preparation) (manuf. of, as electron beam-curable binders, for magnetic RN 106284-62-2 HCAPLUS Hexanedioic acid, polymer with 1,4-butanediol, 1,2-ethanediamine, CN 1,2-ethanediol, .alpha.-hydro-.omega.-hydroxypoly[oxy(methyl-1,2ethanediyl)] ether with 4,6-bis[[(2-hydroxymethylethyl)amino]carbonyl]-1,3benzenedicarboxylic acid (5:4) bis(2-aminomethylethyl) ether, 2-(hydroxymethyl)-2-[[(1-oxo-2-propenyl)oxy]methyl]-1,3-propanediyl di-2-propenoate, .alpha.-[2-hydroxy-3-[(1-oxo-2-propenyl)oxy]propyl]-.omega.-[2-hydroxy-3-[(1-oxo-2-propenyl)oxy]propoxy]poly(oxy-1,2ethanediyl) and 1,1'-methylenebis[4-isocyanatobenzene] (9CI) (CA INDEX NAME)

CM 1

CRN 106284-61-1

CMF (C3 H6 O)n C70 H88 N10 O26

CCI IDS, PMS

PAGE 1-A

10 (D1-Me)

PAGE 1-B

$$-CH_2-CH_2-CH_2-CH_2-CH_2-NH-C-NH-CH_2-CH_2-NH-C-NH-CH_2-CO_2H$$

PAGE 1-C

$$-CH_{2} - CH_{2} - CH_{2} - CH_{2} - CH_{2} - NH - C - NH - CH_{2} - CH_{2} - NH - CH_{2} - CO_{2}H$$

PAGE 1-D

$$-CH_2$$
 $-CH_2$ $-CH_$

PAGE 1-E

$$-CH_2$$
 $O-(C_3H_6)$ $O-CH_2$ CH_2 NH_2

CM 2

CRN 87719-53-7

CMF (C2 H4 O)n C12 H18 O7

CCI PMS

PAGE 1-B

CM 3

CRN 3524-68-3 CMF C14 H18 O7

CM 4

CRN 124-04-9 CMF C6 H10 O4

 $HO_2C-(CH_2)_4-CO_2H$

CM 5

CRN 110-63-4 CMF C4 H10 O2

 $HO-(CH_2)_4-OH$

CM 6

CRN 107-21-1 CMF C2 H6 O2

 ${\tt HO-CH_2-CH_2-OH}$

CM 7

CRN 107-15-3 CMF C2 H8 N2

 $H_2N-CH_2-CH_2-NH_2$

CM 8

CRN 101-68-8 CMF C15 H10 N2 O2

L48 ANSWER 28 OF 28 HCAPLUS COPYRIGHT 2003 ACS

AN 1969:29403 HCAPLUS

DN 70:29403

TI Preparation and three-dimensional polymerization of polyester methacrylates

AU Mikhailov, Marin; Budevska, Kh.; Nenkov, G.; Gerdzhikova, S.

CS Inst. Org. Chem., Sofia, Bulg.

SO Journal of Polymer Science, Polymer Symposia (1968), Volume Date 1965, No. 16(Pt. 7), 3811-20 CODEN: JPYCAQ; ISSN: 0360-8905

DT Journal

LA English

CC 35 (Synthetic High Polymers)

AB Two methods for the prepn. of polyester methacrylates are described: (1) Methacrylation of hydroxy-contg. oligoesters obtained by the

reesterification of the esters of alcs. and polyhydric acids with polyhydric alcs. Terephthalic, furan-2,5-dicarboxylic, and benzylphosphonic polyester methacrylates, which cannot be obtained by means of condensation telomerization, are synthesized by this method. (2) Addn. telomerization of mixts. of epoxy compds. with anhydrides of methacrylic and polyhydric acids. A closer study of this reaction revealed that it permits the prepn. of a wider range of polyester methacrylates than does condensation telomerization. The three-dimensional polymn. of some of the synthesized polyester methacrylates was investigated. polymethacrylates; polyester methacrylates polymn; polymn polyester methacrylates; methacrylates polyester polymn Polyesters, preparation

ΙT

RL: PREP (Preparation)

(oligomeric, hydroxy-terminated, dimethacrylates)

Methacrylic acid ΙT

ST

RL: SPN (Synthetic preparation); PREP (Preparation) (diester with polyester oligomers, prepn. of)

27598-47-6P 27598-48-7P 28727-78-8 IT **9003-68-3P**, preparation 28728-19-0

RL: PREP (Preparation)

(oligomeric, hydroxy-terminated, dimethacrylate)

27598-51-2P, 27598-49-8P, preparation 27598-50-1P, preparation IT 28724-20-1P 28724-21-2P 27707-54-6P, preparation preparation 28724-22-3P

RL: SPN (Synthetic preparation); PREP (Preparation) (prepn. of)

ΙT 9003-68-3P, preparation

RL: PREP (Preparation)

(oligomeric, hydroxy-terminated, dimethacrylate)

9003-68-3 HCAPLUS RN

1,4-Benzenedicarboxylic acid, polymer with 1,2-ethanediol (9CI) (CA INDEX CN NAME)

CM 1

CRN 107-21-1 CMF C2 H6 O2

но- сн2- сн2- он

CM 2

CRN 100-21-0 CMF C8 H6 O4

CRN 190124-76-6

CMF (C10 H10 O7 S . C10 H10 O4 . C10 H10 O4 . C8 H10 O4 . C5 H12 O2 . C2

H6 O2 . Na)x

CCI PMS

CM 3

CRN 3965-55-7

Na

CM 4

CRN 1459-93-4 CMF C10 H10 O4

CM 5

CRN 126-30-7 CMF C5 H12 O2